Book of Nanomedicine

(Editor, Dr. Thomas. M.S. Chang)

Biocompatibility of a Highly Concentrated Fluid of Hemoglobin-vesicles as a

Transfusion Alternative.

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1

1. Introduction

Blood transfusion systems have greatly benefited human health and welfare. Nevertheless,

some problems remain: possibility of infection, blood type mismatching, immunological

response, and a short shelf life that is insufficient for stockpiling for emergency situations.

Realization of artificial O₂ carriers is anticipated to solve such problems. The most abundant

protein in blood is Hb ([Hb] = 12-16 g/dL in healthy human blood), indicating that oxygen

transport to tissues is the most important function of blood. To design an oxygen carrying

fluid to substitute the function of blood, the Hb concentration of the fluid should be high and

comparable with that of blood Hb concentration. Chemically modified and cell-free Hb based

oxygen carriers (HBOCs), such as intramolecularly crosslinked, polymerized, and

polymer-conjugated Hbs were synthesized to prevent toxic effects of cell-free Hbs [1]. The

hydrodynamic radius of such cell-free HBOCs is less than 20 nm. On the other hand,

hemoglobin-vesicles (HbV) or so-called liposome-encapsulated Hb, encapsulate a

concentrated Hb solution in phospholipid vesicles, are developed [2,3]. One particle of about

 $250\mbox{-}280~\mbox{nm}$ encapsulates nearly 30,000 Hb molecules. In the case of the chemically modified

cell-free HBOCs, they are "dissolved" in aqueous solutions like plasma proteins do. On the

other hand, HbV as a cellular HBOC is "dispersed" in aqueous solutions like blood cells do.

The difference between dissolution and dispersion results in considerable difference in

physicochemical characteristics of the fluids containing a high concentration of Hb. The fluid

properties should be adjusted within the biocompatible and physiological conditions for a

massive blood exchange. In this chapter I firstly summarized some views of physicochemical

differences between cell-free and cellular HBOCs. Secondly, biocompatibility of the dispersed particle is of course important. The blood compatibility, biodegradability, excretion, and immunological responses to the massive injection of such dispersed small particles are summarized.

2. Biocompatible Solution properties of HbV fluids

2-1. Colloid osmotic pressure

Albumin, dissolved in a blood plasma at ca. 5 g/dl, provides sufficient colloid osmotic pressure (COP, ca. 20 Torr) to play an important role in equilibrating COP between blood and interstitial fluid, thereby maintaining the overall blood volume. This COP is one requisite for a transfusion alternative to sustain blood circulation for transporting oxygen and metabolites. The extremely high concentration of the HbV suspension ([Hb]) 10 g/dL; [lipids]) 6 g/dL, volume fraction, ca. 40 vol %) imparts an O₂ carrying capacity that is comparable to that of blood. The HbV suspension does not possess a colloid osmotic pressure (COP), because one HbV particle (ca. 250 nm diameter) contains about 30 000 Hb molecules. In fact, HbV acts as a particle, not as a solute. Therefore, HbV must be suspended in or coinjected with an aqueous solution of a plasma substitutes. This requirement is identical to that for emulsified perfluorocarbon, which does not possess COP [4, 5]; it contrasts to characteristics of other Hb-based O₂ carriers, intramolecular cross-linked Hbs, polymerized Hbs, and polymer-conjugated Hbs, which all possess very high COP as protein solutions [6, 7] (Fig. 1). These chemically modified Hb solution can be categorized as "O₂-carrying plasma

expanders" [8], because it has the oxygen carrying capacity and colloid osmotic pressure. However, the problem of PEG-modified Hb solution is that the COP is too high that the Hb concentration of the resulting fluid is as low as 4-6 g/dL. On the other hand, HbV suspended in any plasma expander shows COP of the suspending medium, at any Hb concentration. When HbV is suspended in 5%-human serum albumin solution (HSA), COP is nearly 20 mmHg which is in a physiological range of COP.

According to the guideline for safer blood transfusion, a transfusion trigger (the critical Hb level) is 6 g/dL to minimize unnecessary transfusion strictly or to avoid allogeneic transfusion as long as possible. But the problem of HBOCs with low Hb concentration is that injection of HBOCs cannot increase blood Hb level. In fact, according to the retrospective description of Nose, pyridoxalated Hb polyoxyethylene conjugated (PHP) had the same problem and it was not easily approved for clinical study during the negotiation of FDA [9]. The Hb concentration of HbV is adjusted to 10 g/dL, which is higher than the concentration of transfusion trigger.

2-2. Flocculate formation and viscosity increase in the presence of plasma expanders

Animal tests of HbV suspended in plasma-derived HSA or recombinant HSA (rHSA) showed

an O₂ transporting capacity that is comparable to that of blood [10, 11]. We reported that HbV

suspended in plasma-derived HSA or rHSA was almost Newtonian: no aggregation was

detected microscopically [12,13]. In Japan, rHSA was very recently approved for clinical use.

3

in May 2008 [14], but various plasma substitutes are used worldwide, such as hydroxylethyl starch (HES), dextran (DEX), and modified fluid gelatin (MFG). The selection among these plasma substitutes is best determined not only according to their safety and efficacy, but also according to their associated price, experience of clinicians, and customs of respective countries. Water-soluble polymers generally interact with particles such as polystyrene beads, liposomes, and RBCs to induce aggregation or flocculation [15, 16]. As for the cell-free HBOCs dissolved in saline, they are Newtonian fluid, and does not interact with plasma expanders. In the case of cellular HBOC dispersed in saline, it is important to determine the compatibility of HbV with these plasma substitutes. With that background, we studied rheological properties of HbV suspended in these plasma substitute solutions using a complex rheometer and a microchannel array [17]. The rheological property of an HBOC is important because the infusion amount is expected to be considerably large, which might affect the blood viscosity and hemodynamics. The HbV suspended in rHSA was nearly Newtonian (Figure 2). Its viscosity was similar to that of blood, and the mixtures with RBCs at various mixing ratios showed viscosities of 3-4 cP. Other polymers, HES, DEX, and MFG, induced flocculation of HbV, possibly by depletion interaction, and rendered the suspensions as non-Newtonian with the shear-thinning profile. These HbV suspensions showed high viscosity and a high storage modulus (G') because of the presence of flocculated HbV. On the other hand, HbV suspended in rHSA exhibited a very low G'. The viscosities of HbV suspended in DEX, MFG, and high-molecular-weight HES solutions responded quickly to rapid step changes of shear rates of 0.1-100 s⁻¹ and a return to 0.1 s⁻¹, indicating that

flocculation formation is both rapid and reversible. Microscopically, the flow pattern of the flocculated HbV perfused through microchannels (4.5 μ m deep, 7 μ m wide, 20 cmH₂O applied pressure) showed no plugging.

The mechanism of flocculate formation of liposome is controversial [18]. However, we believe PEG-modified liposomes is flocculated by depletion interaction. The flocculation level increased with hydrodynamic radius (R_h) or radius of gyration (R_g) of series of HES or DEX with different molecular weights at a constant polymer concentration (4 wt %). However, the flocculation level differed markedly among the polymers (Figure 3). A crowding index (C_i) representing the crowding level of a polymer solution is defined as (excluded volume of one polymer) × (molar concentration) × Avogadro's number, using R_h or R_g . All polymers' flocculation level increases when C_i approaches 1: when the theoretical total excluded volumes approach the entire solution volume, the excluded HbV particles are forced to flocculate.

2-3. In vivo study of co-injection of HbV and a series of plasma expanders

It remained unknown whether such flocculate formation of HbV in blood might affect animal's hemodynamics. Using a rat model, we tested infusion of a series of plasma expander (MFG, HES₆₇₀, HES₁₃₀, HES₇₀, rHSA) to maintain the blood volume (level of blood exchange led to 60%) at repeated hemorrhages and the subsequent infusion of HbV (20 ml/kg, 36% of blood volume) [19]. (In this experiment we did not use dextran because rats shows

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anaphylactic reaction to dextran [20]). All rats survived for 4 hr after the infusion of HbV; hemodynamic and respiratory functions were preserved, indicating that the flocculation does not induce capillary embolism. Blood exchange with rHSA and subsequent infusion of HbV showed more stable systemic parameters because of the longer retention of rHSA in blood than other plasma substitutes, indicating that rHSA is suitable for combination with HbV in this experimental model.

2-4. Solution properties affects on reactions of Hb and NO

It has been regarded that lower blood viscosity after hemodilution is effective for tissue perfusion. However, microcirculatory observation shows that, in some cases, lower "plasma viscosity" decreases shear stress on the vascular wall, causing vasoconstriction and reducing the functional capillary density [21]. Therefore, an appropriate viscosity might exist, which maintains the normal tissue perfusion level. The large molecular dimension of HbV can result in a transfusion fluid with high viscosity. A large molecular dimension is also effective to reduce vascular permeability and to minimize the reaction with NO and CO as vasorelaxation factors [22-25].

Increased fluid viscosity of a solution of hemoglobin-based oxygen carriers (HBOCs) reduces vasoconstrictive effects because increased shear stress on the vascular wall enhances the production of vasorelaxation factors such as NO. Nevertheless, on a microcirculatory level, it remains unclear how viscosity affects the reaction of HBOCs and NO. To clarify the effect of

viscosity on the NO-binding, different HBOCs were perfused through narrow gas-permeable tubes (25 µm inner diameter at 1 mm/s centerline velocity; hemoglobin concentration [Hb]=5 g/dL) [26]. The reaction was examined microscopically based on the Hb visible-light absorption spectrum. When immersed in a NO atmosphere, the NO-binding of deoxygenated Hb solution (viscosity, 1.1 cP at 1000 s⁻¹) in the tube occurred about twice as rapidly as that of red blood cells (RBCs): 1.6 cP (Figure 4). Binding was reduced by PEGylation (PEG-Hb, 7.7 cP), by addition of a high molecular weight hydroxyethyl starch (HES) (2.8 cP), and by encapsulation to form Hb-vesicles (HbVs, 1.5 cP; particle size 279 nm). However, the reduction was not as great as that shown for RBCs. A mixture of HbVs and HES (6.2 cP) showed almost identical NO-binding to that of RBCs, Higher viscosity and particle size might reduce lateral diffusion when particles are flowing. The HbVs with HES showed the slowest NO-binding. Furthermore, Hb encapsulation and PEGylation, but not HES-addition, tended to retard CO-binding. Increased viscosity reportedly enhances production of endothelium NO. In addition, our results show that the increased viscosity also inhibits the reaction with NO. Each effect might mitigate vasoconstriction.

3. Biocompatibility of HbV in terms of immunological responses

3-1. Complement activation

A so-called injection reaction, or pseudo-allergy, resulting from complement activation after injection of a small amount of liposome is well known, giving rise to anaphylatoxins, which trigger various hypersensitivityreactions [27-29]. Transient thrombocytopenia and pulmonary

7

hypertension in relation to complement activation is an extremely important hematologic effect observed in rodent and porcine models after infusion of LEH (containing DPPG) developed by the US Naval Research Laboratory [30]. Neo red cells (Terumo Corp.) containing stearic acid showed pulmonary hypertension in beagle and porcine models [31], but not in monkeys. In our group, exchange transfusion of prototype HbV (containing DPPG, no PEG modification) in anesthetized rats engendered transient thrombocytopenia and slight hypertension [32]. The transient reduction in platelet counts and increase of thromboxane B2 caused by complement-bound liposomes was also associated with sequestration of platelets in the lung and liver [30]. In the present formation of HbV, we use a negatively charged lipid (DHSG) instead of DPPG. It does not induce thrombocytopenia or complement activation in animal experiments [33,34], probably because it contains PEGylated lipids and a different type of negatively charged lipid (DHSG), instead of DPPG or a fatty acid. The in vitro human blood compatibility of HbV has been extensively studied [33,35-37]. The present PEG-modified HbV containing DHSG does not affect the extrinsic or intrinsic coagulation activities of human plasma, although HbV-containing DPPG and no PEGmodification tends to shorten the intrinsic coagulation time. The kallikrein-kinin cascade of plasma was activated slightly by the prototype DPPG-HbV, but not by the present PEG-DHSG-HbV. The exposure of human platelets to high concentrations of the this HbV (up to 40%) in vitro do not cause platelet activation and do not affect adversely the formation and secretion of prothrombotic substances or proinflammatory substances that are triggered by platelet agonists [38]. These results imply that HbV, at concentrations of up to 40%, do not have

9

aberrant interactions with either unstimulated or agonist-induced platelets. It can be concluded that the PEG-DHSG-HbV described here have higher blood compatibility.

3-2. RES trap, degradation, and excretion

Biodistribution of HbV was examined using ^{99m}Tc-conjugated homocysteine or glutathione containing HbV [39] and HbV containing ¹²⁵I-labeled Hb [40]. These experiments show that HbV are finally captured by macrophages, mainly in the spleen and liver. Electron microscopic observation can detect the presence of Hb-encapsulating particles in the phagosomes of macrophages because of the high densities of protein and electrons (derived from Fe) in the particles such as RBCs. The HbV particles disappear in 1 week [41].

Immunohistochemical staining with antihuman Hb antibody and antimethoxy-PEG indicates that Hb and PEG of HbV disappear in 2 weeks [41-43]. It was shown recently that ¹²⁵I-labeled Hb and ³H-labeled cholesterol in HbV have identical blood clearance, indicating that HbV retains its integrity in the bloodstream, and distributes to the reticuloendothelial system together. However, ¹²⁵I mainly appears in urine, and ³H in feces, showing different metabolic routes in the macrophages [44].

3-3. Transient immunosuppressive effect

Accumulation of considerable amounts of liposome in a MPS can affect immunologic response. Actually, phagocytic index measured by carbon particle clearance in rats showed significant reduction of phagocytic index one day after injection of HbV. While, it increased

considerably 3 days after injection [41], indicating the increased defense function. On the other hand, HbV showed transient suppressive effect on the proliferation of rat splenic T cells. Takahashi et al. of Hokkaido Red Cross Blood Center [45] elucidated the mechanism underlying that phenomenon and its effect on both local and systemic immune response. HbV was intravenously at a volume of 20% of whole blood into rats. Then their spleens were removed, and T cell responses to concanavalin A (Con A) or keyhole limpet hemocyanin (KLH) were evaluated by measuring the amount of [3H]thymidine incorporated into DNA. Results showed that T cell proliferation in response to Con A or KLH was inhibited from 6 h to 3 days after the liposome injection. The phagocytosis of the large load of liposomes by rat CD11b/c+, class II immature monocytes temporarily renders them highly immunosuppressive, but most importantly, the systemic immune response was unaffected.

3. Conclusion

Liposomes are clinically utilized for cancer and antifungal therapies, and other usages as a vehicle of functional molecules are developed aggressively. HbV is one liposomal product but the differences to such conventional liposomal products are that it is a highly concentrated fluid and it inevitably requires a massive dose (like 20 mL/kg body weight) as it will be utilized as a substitute of a RBC concentrate. Therefore, injection of HbV would affect spontaneously on hemorheology, hemodynamics, immune system, phagocytosis, gas exchange reactions between tissue and blood, etc. It is also important to have stability as a capsule during storage and during blood circulation to shield a toxic effect of molecular Hb. It

11

also requires instability to be decomposed by macrophages and complete excretion from a body. In this chapter we discuss about such important biocompatibilities of HbV. We believe the above mentioned biocompatibility of HbV guarantees the safety of HbV and a potential for versatile clinical application.

Acknowledgments

The author greatly appreciates Emeritus Professor Eishun Tsuchida, Waseda University, for his support for the Project of Oxygen Infusion. Research of Hb-vesicles has been conducted by an academic consortium comprising many domestic and overseas research institutes. The author acknowledges the contribution of the collaborators. This research has been supported by Health and Sciences Grants from Ministry of Health Labour and Welfare, Japan; and a Grant in Aid for Scientific Research from Japan Society for the promotion of Sciences (JSPS).

Disclosure

Hiromi Sakai is an inventor holding some patents related to the production and utilization of Hb-vesicles.

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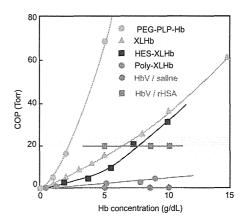
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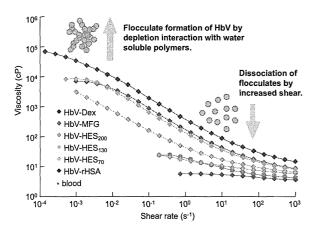
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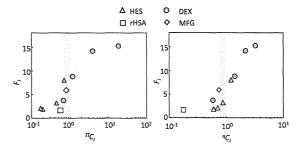
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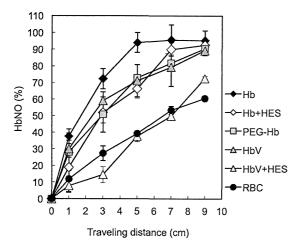
Figure Legend

- Figure 1. Colloid osmotic pressures of cell-free and cellular HBOCs [6]. COP of HbV is determined by the suspending medium. For example, 5% albumin has 20 torr COP. It does not exceed the physiological condition at any Hb concentration.
- Figure 2. Shear-thinning profiles of HbV suspended in a series of plasma expanders. [Hb] = 10 g/dL [17].
- Figure 3. Flocculation index (F_i) increases by the addition of dextran HES, MFG, or HSA, showing that the crowding index, C_i , of the polymer aqueous solution is the determining factor of flocculate formation [18]. We defined the flocculation index (F_i), as $F_i = (\eta_{10} \eta_0) / (\eta_{1000} \eta_0)$. In that equation, η_{10} and η_{1000} respectively represent the viscosity at the shear rates of 10 and 1000 s⁻¹. Crowding index (${}^{\circ}C_i$, and ${}^{\circ}C_i$) representing the crowding level of a polymer solution is defined using R_h and R_g , respectively, as [(excluded volume of one polymer) × (molar concentration) × Avogadro's number]. Adapted with permission from (Sakai H, Sato A, Takeoka S, Tsuchida E. (2009) Mechanism of flocculate formation of highly concentrated phospholipid vesicles suspended in a series of water-soluble biopolymers. Biomacromolecules 10: 2344-2350.). Copyright (2009) American Chemical Society.
- Figure 4. Change of the level of NO-binding reactions of the Hb containing fluids, Hb solution, PEG-Hb, HbV, Hb+HES, HbV+HES, and RBC (black circles) with traveling distance [26] (Permission obtained from Elsevier).









Review paper for "Hemoglobin-based oxygen carriers
- principles, approaches and current status"
(Springer-Verlag)
Editors, Dr. Hae Won Kim, Dr. A. Gerson Greenburg

Chapter 11: Cellular-type hemoglobin-based oxygen carriers to mimic the red blood cell structure

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ABSTRACT

Hemoglobin (Hb), the most abundant protein in blood (12–15 g/dL), can be separated from blood components quite easily. Nevertheless, various side effects of stroma-free Hb have emerged during the long development of Hb-based oxygen carriers (HBOCs). The physiological significance of the RBC structure is undergoing reconsideration. Fundamentally, excessive native Hb molecules are toxic, but encapsulation can shield this toxic effect. Cellular type HBOCs of various kinds are developed using liposome and polymer membranes that mimic the RBC cellular structure. Even though many cell-free HBOCs are undergoing clinical trials, cellular type HBOCs remain in the preclinical stage. Encapsulation of Hb can shield the toxic effects of Hb, but encapsulation presents its own obstacles to realization, such

Key words:

Encapsulated hemoglobin, capsule, hemolysis, nanobiotechnology, blood substitutes

as efficient encapsulation, biocompatibility, and biodegradability.

1. Chemically modified cell-free Hb and encapsulated Hb

The concentration of hemoglobin (Hb) in healthy human blood is around 12–15 g/dL, making Hb the most abundant protein in blood. Hb is an oxygen binding protein that is compartmentalized in red blood cells (RBCs) with an intracellular Hb concentration of about 35 g/dL. Packed RBCs derived from blood donation can be stored only for 6 weeks in the US and for 3 weeks in Japan. Historically, a crude Hb solution was tested as a substitute for RBCs in 1898 (Von Stark), but it was not successful because of various side effects. Since the late 1960s, chemically modified Hb solutions have been developed (Vandegriff & Winslow 1991). Many materials have progressed to use in clinical studies, but many such studies have been suspended because of side effects (Natanson et al., 2008). Recombinant human Hb was also tested, but it failed in clinical trials (Murray et al., 1995). Actually, an earthworm, as a lower organism, has no RBCs, but it does have gigantic Hb molecules. Mammalians, as higher animals, have RBCs for several physiological reasons. It seems difficult to create an RBC substitute with cell-free Hb solutions. Even though Hb is the most abundant protein in blood, it becomes toxic once released from RBCs.

We believe in the physiological importance of the cellular structure of RBCs, and continue to develop Hb-vesicles (HbV) as a cellular-type HBOC (Sakai et al., 2008; Tsuchida et al., 2009). By considering the physiological importance of RBCs, it is easy to understand the side effects of cell-free HBOCs. An RBC has a biconcave disk structure with 8 μm long-axis diameter, encapsulating about 2 million Hb molecules (Mw. 64500) at a concentration of about 35 g/dL. The physiological reasons for Hb compartmentalization in RBCs are the following: i) shielding direct contact of toxic Hb and vasculature (Burhop et al., 2004); ii) prevention of extravasation of dissociated Hb dimers through renal glomeruli, and prolonged circulation time; iii) circumvention of high colloid osmotic pressure and viscosity of concentrated Hb solution (Sakai et al., 2000); iv) co-encapsulation of electrolytes, ATP, glycolytic, and metHb reducing enzymatic systems, etc.; v) retarded reaction of Hb with NO and CO as vasorelaxation factors and retarded O₂-release in the vasculature (Sakai et al., 2008,

2010); vi) RBCs tend to flow near the centerline in vasculature (centralization), avoiding contact with vascular walls where shear stress is the greatest. This flow style is appropriate for preventing hemolysis (Sakai et al., 2009); vii) Moreover, the high viscosity of blood is mainly attributable to the presence of RBCs, producing a non-Newtonian fluid, which is important for blood circulation, especially in microcirculation, from a physiological perspective.

2. Attempts to produce cellular type HBOCs using polymeric materials

Chang (McGill University) was the first to test encapsulation of Hb solution with a polymer membrane in 1957 (Chang, 2007) as one example of "artificial cells". In Japan. Kimoto and his colleagues tested Hb encapsulation from around 1961 using polystyrene, gelatin, and rubber membranes (Toyoda 1966; Kimoto et al., 1968; Kitajima et al., 1970). Although their attempts were original, they were unsuccessful: the particle size could not be reduced to less than capillary diameter (< 4 µm). Later, polymeric materials of various kinds with biodegradable properties became available through the use of polypeptides (Arakawa et al., 1975; Palath et al., 2007), polycaprolactone, and polylactide (Zhao et al., 2007; Zhang et al., 2008) with much smaller diameters. These capsules have permeability of small ionic molecules, which would be advantageous for the reduction of intracellular methemoglobin by reducing agent dissolved in plasma. However, it is speculated that hydrolysis of the polymeric materials during preservation (before injection) and during blood circulation might induce hemolysis: leakage of the encapsulated Hb. Polymersomes are new materials for encapsulation of Hb solution (Rameez et al. 2008). Kishimura et al. (2007) reported encapsulated myoglobin using PEGylated polyion complex vesicles (Table 1). These new materials have been mostly described in reports published in chemistry journals. They await detailed in vivo and in vitro examination to assess their safety and efficacy.

3. Cellular type HBOCs using liposome

In 1964, Bangham and Horne discovered the formation of vesicles (liposomes) when phospholipid was dispersed in aqueous phase. After this discovery, many researchers tested

encapsulation of functional molecules in liposomes, especially for anticancer therapy. In 1977, Djorjevici and Miller (University of Illinois, Chicago) reported encapsulation of Hb in liposomes, called "synthetic erythrocytes" (Table 2). Subsequently, many groups throughout the world attempted so-called liposome encapsulated Hb (LEH). However, most of those efforts were not successful because of their low encapsulation efficiency, polydispersibility of particle size, and instability. The US Naval Research Laboratory aggressively developed freeze-dried powder LEH from the 1980s (Gaber et al., 1983), but the laboratory terminated its development in the late 1990s (Flower & Rudolph 1999), presumably because of low Hb encapsulation efficiency and induction of anaphylactoid reactions (Szebeni et al., 1999), despite the important LEH advantage of long-term storage as a freeze-dried powder using cryoprotectant saccharides. Terumo Corp. (Japan) started development of Neo Red Cells from around 1985 (Suzuki et al., 1988; Takahashi 1995; Pape et al., 2008) using particles that had been surface-modified with PEG chains. However, it suspended its preclinical studies in 2012. As Table 2 shows, most research groups use lipid composition of phosphatidylcholine, cholesterol, negatively charged lipid, and PEG-lipid. Cholesterol not only improves membrane stability; it also reduces the curvature for large unilamellar vesicles. Addition of a small amount of negatively charged lipid increases the repulsive force between the lipid membranes and reduces the lamellarity in addition to controlling the zeta potential for blood compatibility. Saturated phospholipids, such as HSPC, DSPC, and DPPC in Table 2, are preferred to unsaturated lipids such as EYL and soy phosphatidylcholines because of the synergistic, facilitated oxidation of both unsaturated lipids and Hb and physical instability (Szebeni et al., 1985), but cholesterol lowers such Hb denaturation to some degree. Utilization of carbonylhemoglobin (HbCO) is effective to prevent denaturation of Hb during preparation procedures.

Our team at Waseda University has worked to improve the encapsulation efficiency and particle size distribution from the viewpoint of molecular assembly by regulating the electrostatic and hydrophobic interactions between the components (Hb and lipids) (Sakai et

5

al., 2009). The resulting Hb-vesicles (HbV) encapsulate nearly 30,000 Hb molecules (35 g/dL Hb solution) within a 5 nm thin lipid membrane. The selection of lipids was also important for stability and biocompatibility. The starting material, Hb solution, is purified from outdated NAT-inspected red blood cells provided by the Japanese Red Cross. Bovine Hb and swine Hb are also available for the preparation of HbV (Sakai et al., 2002). Carbonylation of Hb (HbCO) prevents metHb formation and denaturation of Hb, and enables pasteurization at 60°C for 10 hr, thereby ensuring the utmost safety from infection. HbCO encapsulated in HbV can be converted easily to HbO₂ by photodissociation using illumination of visible light under O₂ atmosphere. We formerly used polymerizable phospholipids (containing dienoyl group in acyl chain) to stabilize the resulting encapsulated Hb because it was believed that liposome had a fragile structure. However, the problem was that the polymerized liposome was so stable that it was not degraded and it remained in the liver and spleen after intravenous administration into rats. Now we use other combination of conventional phospholipid (DPPC), cholesterol, negatively charged synthetic lipid (Sou & Tsuchida, 2008), and PEG-conjugated phospholipid. The resulting liposome sufficiently prevents aggregation. Complete deoxygenation of the HbV suspension enables long-term storage for years at room temperature (Sakai et al., 2000). Without decarbonylation, HbCO is stable. It can be stored for a long time. Moreover, injection of a cellular HBOC as an HbCO form is beneficial for some pathological conditions (Sakai et al., 2009) and should be studied intensively.

Details of *in vivo* results of safety and efficacy of HbV are summarized in some review papers (Sakai et al., 2008; Tsuchida et al., 2009; Sakai et al., 2011). The *in vivo* oxygen transport capacity of HbV as a resuscitative fluid is described by Dr. Horinouchi in this book.

4. Advantages of gas reactions of encapsulated Hbs

One important physiological aspect of cellular type HBOCs is that their particles are much larger than those of cell-free HBOCs. They do not seem to induce vasoconstriction or hypertension (Nakai et al., 1998; Sakai et al., 2000). Physiochemical analysis of NO reactions

of a series of cell-free HBOCs solutions showed that NO binding rate constants are fast and mostly identical to that of stroma-free Hb (Rohlfs et al., 1988). However, one cellular type of HBOCs, Hb-vesicles (HbV), showed retarded NO binding because of the formation of intracellular diffusion barrier of NO simply by encapsulation of a concentrated Hb solution (Sakai et al., 2008). In fact, HbV encapsulating a diluted Hb solution provides a larger NO binding rate constant: a value similar to that of stroma-free Hb solution.

Moreover, a larger particle shows a slower lateral diffusion in an arteriole that retards the gas reaction at a vascular wall (Sakai et al., 2010). HbV showed a lower rate of NO binding, CO binding, and O₂ release in the model vessels, each of which relates to the vascular tone. In addition, the larger particles prevent penetration across the perforated endothelium to approach to a space between the endothelium and the smooth muscle where NO is produced to bind to soluble guanylate cyclase. In fact, RBCs showed the slowest rate of NO binding, CO binding, and O₂ release. These data imply that RBCs are evolutionally designed to retard gas reactions in blood circulation.

5. Intrinsic difficulties to be considered for realization of encapsulated Hb

Even though Hb encapsulation might shield all the toxic effects of cell-free Hb, cellular HBOCs have their own hurdles that impede their realization. Several are explained here.

5-1. Particle size and encapsulation efficiency

The RBC structure is deformable, facilitating its flow through a capillary with a narrower diameter. However, that attribute of deformability is difficult to mimic artificially. Accordingly, the particle should be smaller than the capillary diameter. It is important to encapsulate a concentrated Hb solution in the particle. To improve the particle function, the weight ratio of the encapsulated Hb to the capsular material is one parameter that must be considered. The Hb concentration in blood is around 12–15 g/dL. A fluid of a cellular HBOC dispersion should have a comparable Hb concentration if it is intended for use as a blood

substitute. For this purpose, the intracellular Hb concentration must be as high as intracellular Hb concentration of RBCs, which is around 35 g/dL.

5-2. Stability of the capsule

The capsule should be stable to retain Hb inside the capsules during storage for a long time, and after injection in the blood circulation until it disappears, because elimination of cell-free Hb is the purpose of Hb encapsulation. The encapsulated Hbs are usually captured by the reticuloendothelial system (RES). The capsule material should be degradable in the macrophage. Their components and their degraded or metabolic materials should never be deposited for a long time in the organs. Accordingly, the capsule material should have both stable and unstable characteristics. The pharmacokinetics of both Hb and capsule should be examined (Taguchi et al., 2009).

Trace amounts of ascorbic acid and thiol compounds are present in plasma, and oxidized cell-free HBOCs can be reduced by these compounds. Because of the stability of a capsule, ionic transport through the capsular membrane is shielded to some degree in the absence of a substitute for ion channels. Encapsulated Hb autoxidizes to form metHb and loses its oxygen binding ability. A remedy for such metHb formation must be considered, such as establishing a reduction system in the capsules (Chang T et al., 2000; Tsuchida et al., 2009).

5-3. Blood compatibility of the capsule

Some of the liposomal products for anticancer therapy induce complement activation. The so-called injection reaction is being clarified continually as clinical experience accumulates, such as dyspensa, tachypenia, tachycardia, hypotension and hypertension, chest pain, and back pain (Szebeni 2005). We confirmed that our prototype HbV, containing phosphatidyl glycerol, induced marked anaphylactoid reactions and cardiopulmonary disorders, manifested as systemic and pulmonary hypertension, increased vascular resistance,

decreased cardiac output, thrombocytopenia, tachycardia, etc. (Sakai et al., 2012). Therefore, it is extremely important to confirm the absence of complement activation of the capsule material (Chang & Lister 1994).

Because the cellular type HBOCs are not dissolved but dispersed in the fluid, the particles sometimes aggregate in the presence of plasma protein by ionic interaction, or depletion interaction. Accordingly, the particle surface would need some surface modification to prevent aggregation.

5-4. Influence on clinical instruments

Light scattering of the particle dispersion, and a stable capsule that cannot be easily destroyed by a detergent, are the chief causes of interference in clinical laboratory tests based on colorimetric and turbidimetric analysis (including quantitative measurement of Hb in blood) and in clinical diagnostic tools such as laser pulsed oxymetry. The level of interference effect should be examined carefully, and a remedy should be considered in advance (Sakai et al., 2002; Suzaki et al., 2008).

Another important point to be considered includes impacts of the RES trap after a massive dose of cellular HBOCs, which might include immunosuppression (Takahashi et al. 2011). This point was discussed at length by our collaborators in other chapter (Azuma et al.) in this book. Even though cellular HBOCs are more complicated than cell-free HBOCs, resolving the issues presented above can realize the successful development of cellular HBOC.

Acknowledgments

Research of Hb-vesicles has been conducted by an academic consortium comprising many domestic and overseas research institutes. The author acknowledges the contribution of those

collaborators. This research has been supported by Health and Labour Sciences Grants (Health Science Research Including Drug Innovation) from the Ministry of Health, Labour and Welfare, Japan, and a Grant-in-Aid for Scientific Research from the Japan Society for the Promotion of Sciences (JSPS).

Disclosure

Hiromi Sakai is an inventor holding some patents related to the production and utilization of Hb-vesicles.

Abbreviations

Hb, hemoglobin; HBOCs, Hb-based oxygen carriers; RBC, red blood cell; HbV, Hb-vesicles; LEH, liposome-encapsulated Hb; HbCO, carbonylhemoglobin

9

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