

et al., 2007). Thus, in the light of the type of mutations induced by safrole, it is unlikely that 8-OHdG formation contributes to safrole-induced genotoxicity, although there is a possibility of any other oxidized DNA damages being involved. The present data showing a significant increase in PCNA-positive hepatocytes might suggest the possible participation of oxidative stress in cell proliferation.

It is well known that the results of bioassay using GST-P-positive foci show good correlation with those of the 2-year cancer bioassay (Ito et al., 2000; Ogiso et al., 1985). Therefore, it has been widely accepted that the analysis of GST-P positive foci may be a useful indicator to predict carcinogenicity of agents. In the present study, the number and the area of GST-P positive foci were significantly increased in both sexes in the 0.5% group. The data on quantitative analysis for GST-P foci using *gpt* delta rats are also in agreement with the carcinogenicity data previously reported by Long et al. (1963).

In conclusion, the present medium-term animal model using F344 *gpt* delta rats confirmed previous reports of the hepatotoxicity and hepatocarcinogenicity of safrole. The genotoxicity of safrole, which remained unknown, so far, was clearly demonstrated in the target organ by this *in vivo* model. Thus, this animal model might be a promising tool for investigating comprehensive toxicities of agents. The acquisition of additional data on key events in chemical carcinogenesis, such as base modification and cell proliferation, could assist in understanding the modes of action. Applications of this model should be further expanded in future studies.

Conflict of interest statement

None.

Acknowledgements

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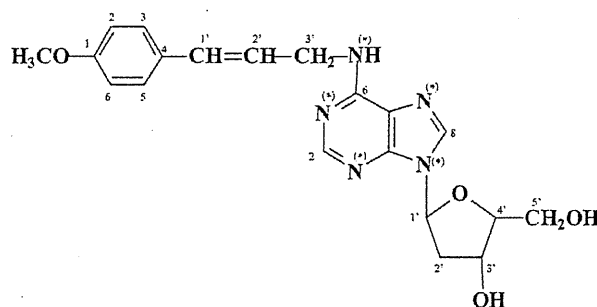
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Detection and Quantification of Specific DNA Adducts by Liquid Chromatography–Tandem Mass Spectrometry in the Livers of Rats Given Estragole at the Carcinogenic Dose

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ABSTRACT: Estragole (ES) is a natural constituent of several herbs and spices that acts as a carcinogen in the livers of rodents. Given that the proximal electrophilic form of ES with a reactive carbocation is generated by cytochrome P450 and a sulfotransferase metabolizing pathway, there is a possibility that the resultant covalent adducts with DNA bases may play a key role in carcinogenesis. The existence of ES-specific deoxyguanosine (dG) and deoxyadenosine (dA) adducts has already been reported with the precise chemical structures of the dG adducts being confirmed. In the present study, we examined ES-specific dA adduct formation using LC-ESI/MS after the reaction of dA with 1'-acetoxy-ES produced by a sulfotransferase metabolic pathway mimic. Although two peaks were observed in the LC-ESI/MS chromatogram, the identification of ES-3'-N⁶-dA as the measurable peak was determined by NMR analysis. To confirm ES-specific dG and dA adduct formation in vivo, an isotope dilution LC-ESI/MS/MS method applicable to in vivo samples for ES-3'-N⁶-dA together with the two major dG adducts, that is, ES-3'-C8-dG and ES-3'-N²-dG, was developed using selected ion recording. The limit of quantification was 0.2 fmol on column for ES-3'-C8-dG and ES-3'-N²-dG and 0.06 fmol on column for ES-3'-N⁶-dA, respectively. Using the developing analytical method, we attempted to measure adduct levels in the livers of rats treated with ES at a possible carcinogenic dose (600 mg/kg bw) for 4 weeks. ES-3'-C8-dG, ES-N²-dG, and ES-3'-N⁶-dA were detected at levels of 3.5 ± 0.4, 4.8 ± 0.8, and 20.5 ± 1.6/10⁶ dG or dA in the livers of ES-treated rats. This quantitative data and newly developed technique for adduct observation in vivo might be helpful for ES hepatocarcinogenesis investigations.



INTRODUCTION

Estragole (4-allyl-1-methoxybenzene; ES) is a natural constituent of essential oils of various herbs and spices (including tarragon, basil, fennel, and anise) present in food.¹ Previous studies have revealed that ES has genotoxicity and carcinogenicity in the livers of mice, concluding that ES is a naturally occurring genotoxic carcinogen.^{2–4} ES is metabolized to the proximate carcinogen, 1'-hydroxy-ES, by cytochrome P450 enzymes (P450), and sulfotransferase (SULT) converts 1'-hydroxy-ES to 1'-sulfoxy-ES.^{2–5} The ultimate electrophilic carbocation structure, which can form covalent adducts with DNA bases, is formed from 1-sulfoxy-ES through dissociation of the sulfate group.^{6,7} Because postlabeling analysis demonstrated the formation of four different adducts such as ES-3'-N²-dG (deoxyguanosine), ES-3'-C8-dG, ES-1'-N²-dG, and ES-3'-N⁶-dA (deoxyadenosine) in the livers of mice administered the proximate carcinogenic metabolite 1'-hydroxy-ES, it has been accepted that the carcinogenicity of ES is caused by the formation of these adducts.^{8–11} However, the precise amount of these adducts formed in vivo under continuous administration with a

carcinogenic dose of ES still remains unknown. Transient base modifications of DNA do not always result in gene mutations because of chemical DNA instability¹² and the existence of specific DNA repair systems.¹³ Because the specific activity of repair enzymes is primarily dependent on stereochemical structures and concentrations of their substrates, confirmation of DNA base modifications and precise quantification are necessary for understanding their biological significance.^{14,15}

The dG modifications have generally been reported as the major adduct,^{16,17} because dG is the most potent nucleophilic nucleoside that can efficiently react with the electrophilic carbocation form.¹⁸ Furthermore, the findings of Y-family polymerase κ , which selectively acts on translesion synthesis on N²-dG adducts, showed the presence of a linkage between dG modification and mutation in the chemical carcinogenicity.^{19–21} However, several carcinogens including aristolochic acid^{22,23} and polyaromatic hydrocarbons²⁴ preferentially form dA adducts,

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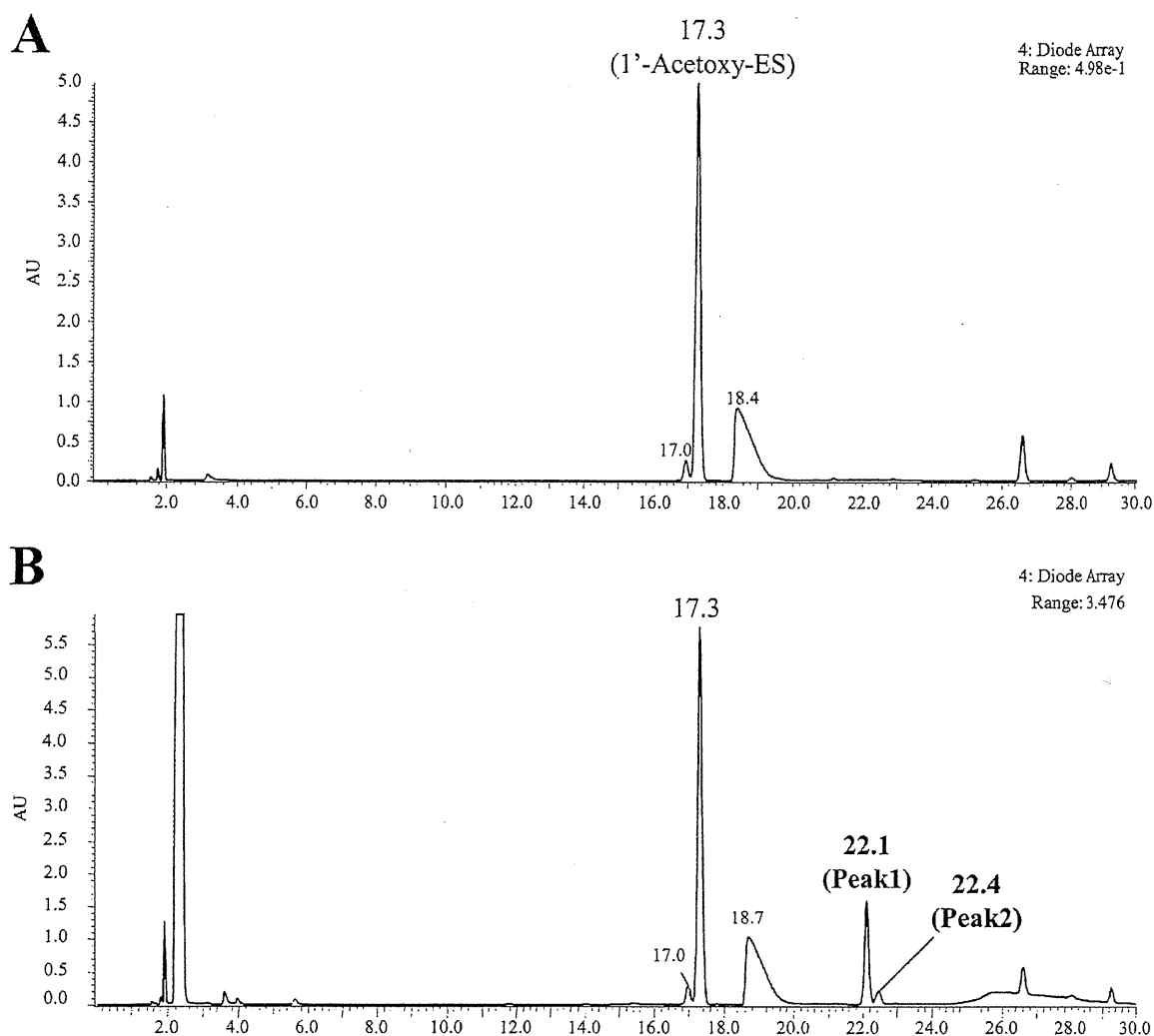


Figure 1. Typical LC-PDA chromatograms (280 nm) in the reaction of (A) 1'-acetoxy-ES alone and (B) 1'-acetoxy-ES with dA. LC-PDA conditions are described in the Materials and Methods.

thereby predominantly inducing dA mutations.^{25,26} Thus, quantitative analysis for each ES-modified base is necessary to investigate the role of specific DNA adducts in ES carcinogenesis.

Although the exact chemical structure of three dG adducts has already been confirmed by mass spectrometry (MS) techniques, that of the dA adduct has not been identified. In this study, ES-dA adduct formation by reaction of dA with 1'-acetoxy-ES (used as a 1'-sulfoxy-ES mimic) was investigated by liquid chromatography (LC)-electron spray ionization (ESI)/MS analysis. The precise chemical structures of the detected adducts were identified using nuclear magnetic resonance (NMR). Subsequently, a quantitative analytical method using LC-ESI/tandem mass spectrometry (MS/MS) for ES-specific dG and dA adducts was developed. After the evaluation of applicability to *in vivo* samples, our LC-ESI/MS/MS method was applied to quantify ES-specific DNA adducts in the livers of rats treated with 600 mg/kg bw of ES for 4 weeks.

MATERIALS AND METHODS

Chemicals and Reagents. ES, *p*-anisaldehyde, vinylmagnesium bromide, tetrahydrofuran (THF), dG, dA, and alkaline phosphatase

were purchased from Sigma Chemical Co. (St. Louis, MO). Nuclease P1 was from Yamasa Shoyu Co. (Chiba, Japan). Stable isotope-labeled [¹⁵N₅]-dG and [¹⁵N₅]-dA were obtained from Cambridge Isotope Laboratories (Cambridge, MA). Acetic anhydride ammonium carbonate, dimethylsulfoxide (DMSO), dichloromethane, *N,N*-dimethylformamide (DMF), diethyl ether, isopropyl alcohol, and DNA extractor TIS kit were purchased from Wako Pure Chemicals (Tokyo, Japan). All other chemicals used were of specific analytical or HPLC grade.

Synthesis of 1'-Hydroxyestragole. 1'-Hydroxy-ES was synthesized from *p*-anisaldehyde as described by Punt et al.⁷ The synthesis of 1'-hydroxy-ES encompassed a Grignard reaction, using vinylmagnesium bromide as the Grignard reagent (1 M solution in THF). Briefly, *p*-anisaldehyde (0.0165 mol) was dissolved in 10 mL of dry THF, and this solution was added dropwise over a period of 30 min to the Grignard reagent (0.035 mol) while stirring at 50 °C in anhydrous conditions under a nitrogen atmosphere. The reaction mixture was further incubated for 90 min, and the resulting solution was added to a solution of 4.5 g of ammonium chloride in 200 mL of ice cold water. The emulsion was stirred for several minutes, and 1'-hydroxy-ES was extracted with diethyl ether. The organic solution was dried with magnesium sulfate, and the yield was 94%. The identity of the product was confirmed by ¹H NMR. ¹H NMR spectra were recorded with a Varian 500 MHz NMR system

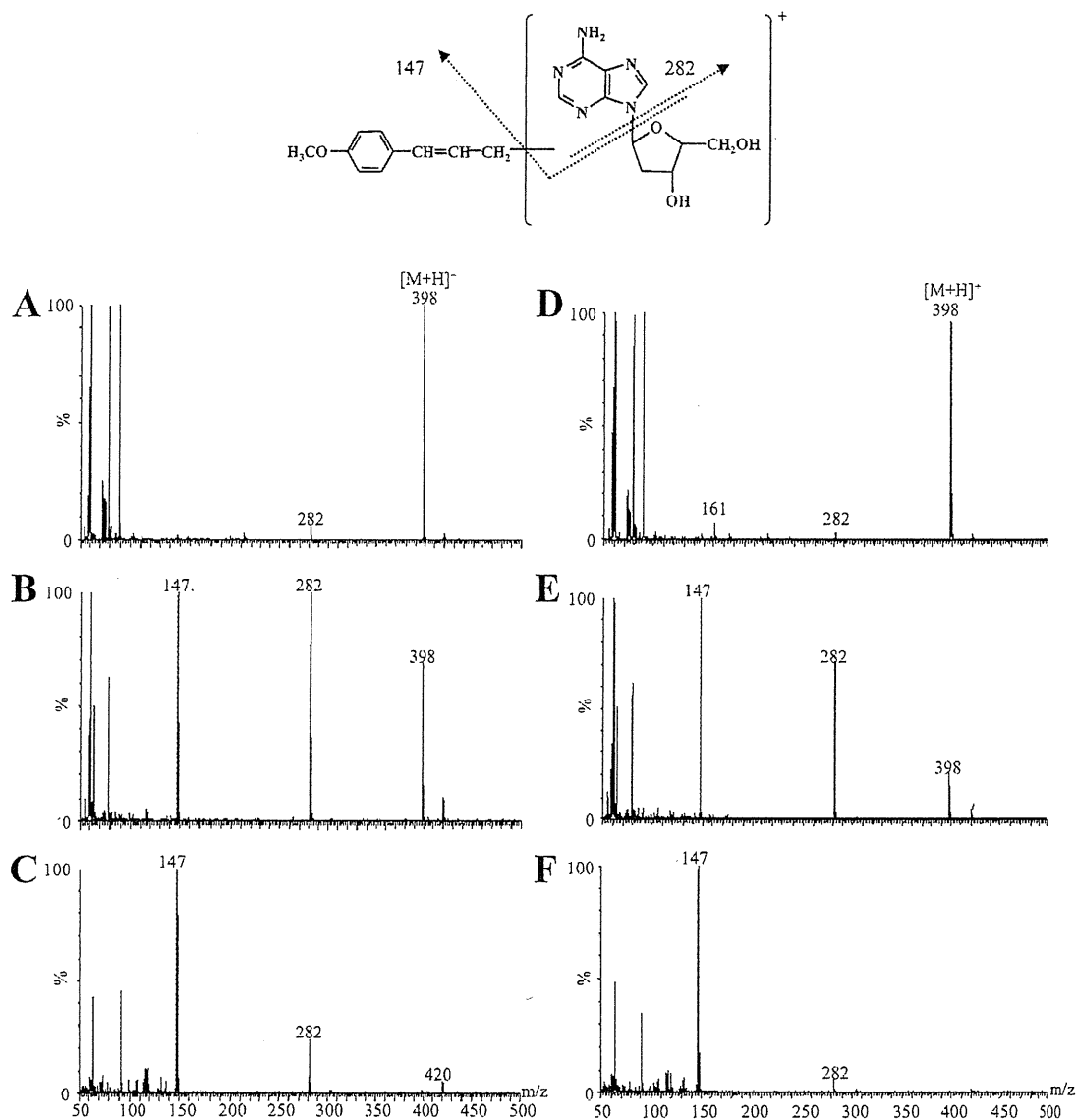


Figure 2. Mass spectra of peaks 1 (A–C) and 2 (D–F) obtained from LC-ESI/MS analysis for the reaction of 1'-acetoxy-ES with dA. Mass analysis was performed in scan mode. The cone voltages were 20 (A and D), 40 (B and E), and 60 V (C and E) in positive ion mode. LC-ESI/MS conditions are described in the Materials and Methods.

(Varian Inc. Corp.). 1'-Hydroxy-ES: $^1\text{H NMR}$ ($\text{DMSO}-d_6$): δ 7.22–7.29 (m, 2H, Ar), 6.82–6.89 (m, 2H, Ar), 5.98–6.10 (m, 1H, $-\text{CHCH}_2$), 5.26–5.35 [m, 2H, $-\text{C}(\text{OH})\text{CH}-$], 5.10–5.20 (m, 2H, $-\text{CHCH}_2$), 3.79 (s, 1H, $-\text{OCH}_3$), 2.24 [s, 1H, $-\text{C}(\text{OH})-$].

Synthesis of 1'-Acetoxyestragole. 1'-Acetoxy-ES was synthesized from 1'-hydroxy-ES as described by Punt et al.⁷ and Drinkwater et al.² In brief, 1'-hydroxy-ES (50 mg) was dissolved in 200 μL of pyridine. Acetic anhydride (33 μL) was added dropwise to this solution, and the reaction mixture was stirred for 5 h at room temperature after which 400 μL of dichloromethane was added. The reaction mixture was extracted several times with aliquots of 200 μL of 1 N HCl. When the aqueous phase reached pH 2–3, the organic layer then immediately was extracted with 400 μL of 1 M sodium carbonate solution (pH 7.6). The organic solution was dried with magnesium sulfate, and the solvent was evaporated in a nitrogen atmosphere. The identity of the product was confirmed by $^1\text{H NMR}$. 1'-Hydroxy-ES: $^1\text{H NMR}$ ($\text{DMSO}-d_6$): δ 7.26–7.28 (m, 2H, Ar), 6.92–6.95 (m, 2H, Ar), 5.99–6.07 (m, 1H,

$-\text{CHCH}_2$), 5.26–5.27 [m, 2H, $-\text{CH}(\text{OH})\text{CH}-$], 5.19–5.23 (m, 2H, $-\text{CHCH}_2$), 3.75 (s, 1H, $-\text{OCH}_3$), 2.05 (s, 1H, $-\text{CO}-\text{CH}_3$).

Determination of ES-Specific dA Adduct. ES-specific dA adducts were determined from reactions between 1'-acetoxy-ES and dA based on the protocol of Phillips et al.⁸ The reaction products containing 1'-acetoxy-ES were diluted 50-fold in DMF from which 200 μL was added to 1.8 mL of 10 mM dA solution in 50 mM sodium phosphate buffer (pH 7.4). The reaction mixture was stirred for 24 h at 37 $^\circ\text{C}$. The reaction mixture was then passed through an HLC-DISK syringe filter (Kanto Chemical Co Inc., Tokyo) and was separated on a LC-photodiode array (PDA)-ESI/MS system. LC-PDA-ESI/MS analyses were performed using an Alliance HT model 2695 liquid chromatographic system coupled to a 996 PDA detector and Micromass ZQ, a single quadrupole, MS system (Waters Corp., Milford, MA) equipped with an ESI source through a splitter. Twenty microliters of the reaction mixture was injected directly onto a reverse-phase C_{18} column (Mightysil RP-18, 4.6 mm \times 150 mm, 5 μm , Kanto Chemical Co.,

Table 1. ¹H NMR Chemical Shifts of Peak 1^a

H-1'	6.34 (1H, m)
H-2'	2.70 (1H, m)
H-2''	2.25 (1H, m)
H-3'	4.40 (1H, m)
H-4'	3.87 (1H, m)
H-5'	3.61 (1H, m)
H-5''	3.51 (1H, m)
OH-3'	5.31 (1H, s)
OH-5'	5.22 (1H, bs)
H-2	8.21 (1H, bs)
H-8	8.37 (1H, s)
N ⁶ -H	8.07 (1H, bs)
ESH-1'	6.42 (1H, m)
ESH-2'	6.21 (1H, m)
ESH-3'	4.22 (2H, bs)
ESH-2,6	7.30 (2H, m)
ESH-3,5	6.85 (2H, m)
ES-3,5	3.75 (3H, m)

^a m, multiplet; s, singlet.

Inc.) maintained at 40 °C. Solvent A was water, solvent B was methanol, and solvent C was 0.1% formic acid. The column was equilibrated with a mixture of solvent A/solvent B/solvent C (70/10/20, v/v). A linear gradient was applied from 30 to 90% methanol over 40 min, kept at 90% for 10 min, lowered to 20% over 2 min, and re-equilibrated at the initial conditions for 15 min. Mass analysis was performed in scan mode. The cone voltages were 20, 40, and 60 V in the positive ion mode.

Synthesis of ES-Derived dA Adduct. ES-dA adducts were synthesized from a reaction between 1'-acetoxy-ES and dA. Four milliliters of 30 mM 1'-acetoxy-ES solution in dichloromethane was diluted 50-fold in DMF from which 200 mL was added to 1800 mL of 10 mM dG or dA solution in 50 mM sodium phosphate buffer (pH 7.4). The reaction was stirred for 24 h at 50 °C. The yield of peak 1 was 86 mg (7.1%). The reaction was repeated several times as needed to acquire enough products for NMR analysis.

Purification was performed using the combined reaction mixtures that were evaporated and reconstituted in 50% methanol/water. The concentrated reaction mixture was separated by a LC system equipped with a UV detector (LC-UV) (PU-2080 Plus Intelligent HPLC Pump, AS-2057 plus Intelligent Sampler, CO-966 Intelligent Column Thermostat, and UV-970 Intelligent UV/vis Detector; Jasco Co., Tokyo). Two milliliters of sample was injected directly on to a reverse phase C₁₈ column (Mightysil RP-18 GP, 20 mm × 250 mm, 5 μm, Kanto Chemical Co., Inc.) maintained at 40 °C. Solvent A was 0.01% formic acid, and solvent B was methanol containing 0.01% formic acid. The column was equilibrated with a mixture of solvent A/solvent B (70/30, v/v). A linear gradient was applied from 30 to 90% methanol over 30 min, kept at 90% for 10 min, lowered to 30% over 2 min, and equilibrated at the initial conditions for 15 min. Products eluting at 22.5 and 31.2 min (UV absorbance at 280 nm; flow rate, 10 mL/min) were collected. The fractions were dried in vacuo and weighed. The identity of the product was confirmed by LC-ESI/MS and ¹H NMR. ¹H NMR spectra were recorded with a Varian 600 MHz NMR system (Varian Inc. Corp., Palo Alto, CA). Chemical shifts are expressed in ppm downfield shift from trimethylsilane (TMS) (δ scale). The synthesis of stable isotopically labeled surrogate standards was also performed on a small scale by the same method. The reaction products were purified using the LC-UV system.

Standard Solutions. ES-3'-C8-dG and ES-3'-N²-dG were synthesized from 1'-acetoxy-ES and dG as described by Punt et al.⁷ The yields

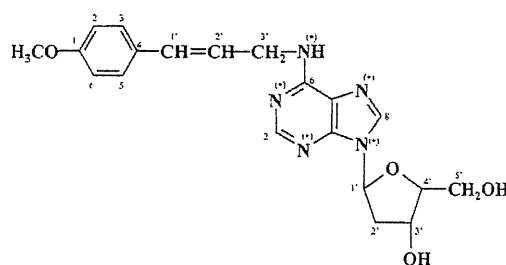


Figure 3. Chemical structure of ES-3'-N⁶-dA adducts and the stable isotopically labeled compounds. The asterisk (*) indicates nitrogen [¹⁵N]-labeling.

of ES-3'-C8-dG and ES-3'-N²-dG were 63.4 (5.2%) and 103.5 mg (8.56%), respectively, in the reaction. The syntheses of [¹⁵N₅]-ES-3'-8-dG, [¹⁵N₅]-ES-3'-N²-dG, and [¹⁵N₅]-ES-3'-N⁶-dA as stable isotopically labeled surrogate standards were also performed on a small scale by the same method. ¹⁵N₅-labeled ES-dG adducts including four isomers and ES-3'-N⁶-dA were repeatedly purified using LC-UV system until 98% and over.

One millimolar solutions of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA were prepared in methanol and immediately diluted with methanol/HPLC grade water (50/50, v/v) to 10 μM (stock solution). Working solutions for calibration (0.01–10 nM for ES-3'-C8-dG, ES-3'-N²-dG, and 0.003–10 nM for ES-3'-N⁶-dA) were prepared by the addition of an adequate amount of surrogate standard and diluted with methanol/HPLC grade water (50/50, v/v) to appropriate concentrations.

LC-ESI/MS/MS Conditions. LC-ESI/MS/MS analyses were performed using a Quattro Ultima (Micromass) coupled to a Hewlett Packard 1100 series (G1322A, degasser; G1312A, bin pump; G1316A, Colcom; G1329A, ALS; Agilent Technologies, Palo Alto, CA). The mass spectrometer was operated using an ESI source in the positive ion mode for multiple reaction monitoring (MRM). An aliquot (20 μL) of the sample was injected into a Mightysil C18-GP (2.0 mm × 150 mm, 5 μm; Kanto Chemical Co., Tokyo, Japan) maintained at 40 °C. Solvent A was 0.001% formic acid, and solvent B was 0.001% formic acid containing acetonitrile. The column was equilibrated with a mixture of solvent A/solvent B (75/25, v/v). The mobile phase consisted of a mixture of 0.001% formic acid/0.001% formic acid containing acetonitrile at an initial ratio of 75/25, employing a linear gradient to a final ratio of 30/70 (v/v) over 30 min, at a constant flow rate of 0.2 mL/min.

In the assay for ES-3'-N²-dG and ES-3'-C8-dG, the precursor ion ([M + H]⁺) was *m/z* 414, and the selected product ion [M + H - glycoside]⁺ was *m/z* 298. Correspondingly, for [¹⁵N]-labeled ES-3'-N²-dG and ES-3'-C8-dG, the precursor ion was *m/z* 419, and the selected product ion was *m/z* 303. The cone voltage used was 18 V, and the collision energy was 10 eV. In the assay for ES-3'-N⁶-dA, the precursor ion ([M + H]⁺) was *m/z* 398, and the selected product ion [M + H - glycoside]⁺ was *m/z* 282. Correspondingly, for [¹⁵N]-labeled ES-3'-N⁶-dA, the precursor ion was *m/z* 403, and the selected product ion was *m/z* 287. The cone voltage used was 12 V, and the collision energy was 18 eV. The source block temperature was 120 °C, and the desolvation temperature was 400 °C. The flow rate of the cone gas was set at 150 L/h, while that of the desolvation gas was set at 600 L/h. Under these conditions, the standard retention times were 11.8, 12.7, and 15.1 min for ES-3'-N²-dG, ES-3'-C8-dG, and ES-3'-N⁶-dA, respectively.

Recovery. The recovery was evaluated by calculating the mean of the response at each concentration. The spiked concentrations (low, middle, and high dose) of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA were determined from the concentrations of each compound in the liver DNA of control rats, using LC-ESI/MS/MS. A standard sample was

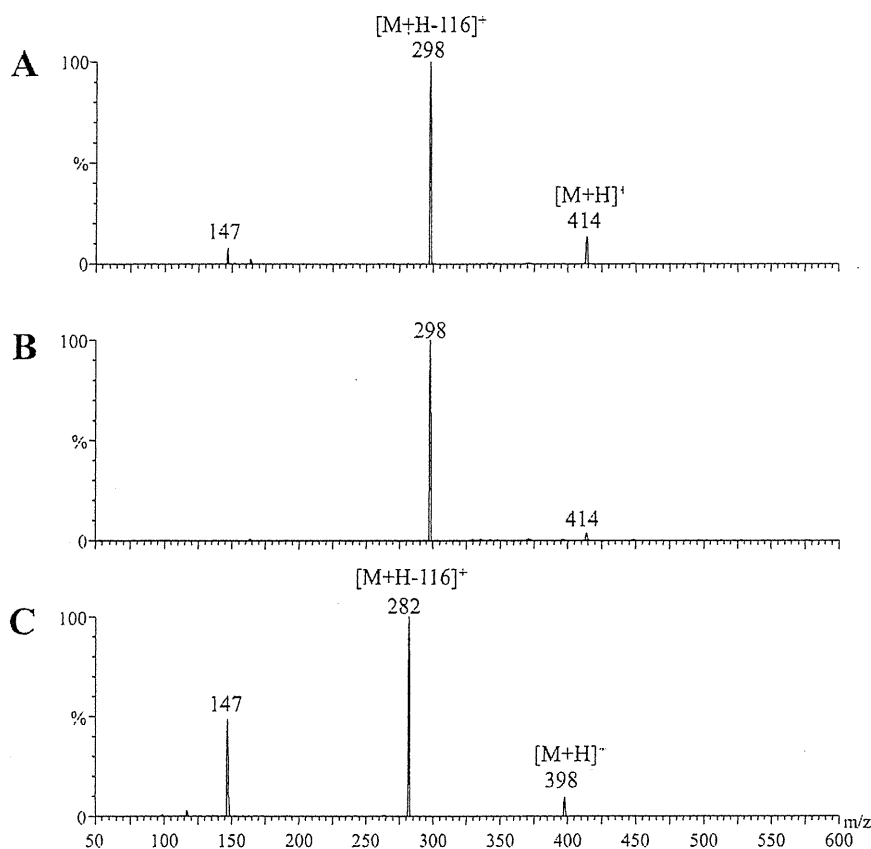


Figure 4. Product ion spectra of (A) ES-3'-C8-dG, (B) ES-3'-N²-dG, and (C) ES-3'-N⁶-dA. The cone voltages and collision energies were set at the optimum conditions for each compound in negative ion mode. LC-ESI/MS/MS conditions are described in the Materials and Methods.

added together with an adequate amount of surrogate standards to 20 mM sodium acetate buffer (pH 4.2) for DNA digestion so that the final concentration might be set to 0.05, 0.5, and 5 nM. The extracted DNA pellets of rat liver were redissolved in this buffer and digested according to the protocol. The sample was analyzed by LC-ESI/MS/MS, and the recovery rates were calculated.

Animal and Treatment. The protocol for this study was approved by the Animal Care and Utilization Committee of the National Institute of Health Sciences (Tokyo, Japan). Five week old male F344 rats were obtained from Japan SLC (Shizuoka, Japan). Ten rats were housed in polycarbonate cages (five rats per cage) with hardwood chips for bedding in conventional temperature (23 ± 2 °C), humidity ($55 \pm 5\%$), air change (12 times per hour), and lighting (12 h light/dark cycle) and were given free access to CRF-1 basal diet (Oriental Yeast Co., Ltd., Tokyo, Japan) and tap water. Groups of 10 rats were given ES by gavage in corn oil at 600 mg/kg bw per day, 5 days/week. All rats were killed at 4 weeks by exsanguination under ether anesthesia, and the livers were immediately removed and weighed. Samples were frozen with liquid nitrogen and stored at -80 °C until measurement of ES-specific DNA adducts.

DNA Isolation and Enzymatic Digestion. DNA extraction and digestion were performed according to the method of Nakae et al.²⁷ and our previous report.²⁸ The samples were homogenized with lysis buffer including commercial DNA isolation kit. The mixture was centrifuged at 10000g for 20 s at 4 °C. The deposit was dissolved in 200 μ L of enzyme reaction buffer. After treatment with RNase and protease K, the DNA pellet was obtained by washing with 2-propanol and ethanol and centrifugation.

The dried DNA pellet was dissolved in surrogate standard containing 20 mM sodium acetate buffer, pH 4.8, and was incubated with 4 μ L of nuclease P1 (2000 U/mL) at 70 °C for 15 min. Then, 20 μ L of 1.0 M Tris-HCl buffer, pH 8.2, was added, and the sample was incubated with 4 μ L of alkaline phosphatase (2500 U/mL) at 37 °C for 60 min. After the addition of 3.0 M sodium acetate buffer, pH 5.1, the digested DNA samples were used for adduct analysis and base analysis. Then, 50 μ L of the digested sample for dG and dA analysis was passed through 100000 NMWL filter (Millipore, Bedford, MA) and injected into the LC-UV. One hundred microliters of digested sample for adduct analysis was diluted with an equal volume of methanol and injected into the LC-ESI/MS/MS.

LC-UV Analysis for dG and dA. dG and dA were determined by an LC-UV system (Jasco Co.: PU-980 Intelligent HPLC Pump, AS-950-10 Intelligent Sampler, CO-1560 Intelligent Column Thermostat, MD-1515 Multiwavelength Detector, Tokyo, Japan). Two milliliters of sample was injected directly on to a reversed phase C18 column (Ultrasphere ODS, 4.6 mm \times 250 mm, 5 μ m, Beckman Coulter, Inc.) maintained at 40 °C. Solvent A was 0.01% formic acid containing water, and solvent B was 0.01% formic acid containing methanol. The column was equilibrated with a mixture of solvent A/solvent B (98/2, v/v). The compounds were eluted at a flow rate of 1.0 mL/min. A linear gradient was applied from 2 to 10% methanol over 20 min, kept at 10% for 5 min, lowered to 2% over 2 min, and equilibrated at these initial conditions for 15 min. The wavelength of the UV detector was set at 280 nm for the detection of dG and dA.

RESULTS

Identification of Luc-N⁶-dA. The reaction mixtures for 1'-acetoxy-ES with/without dA were separated by LC-PDA-ESI/MS.

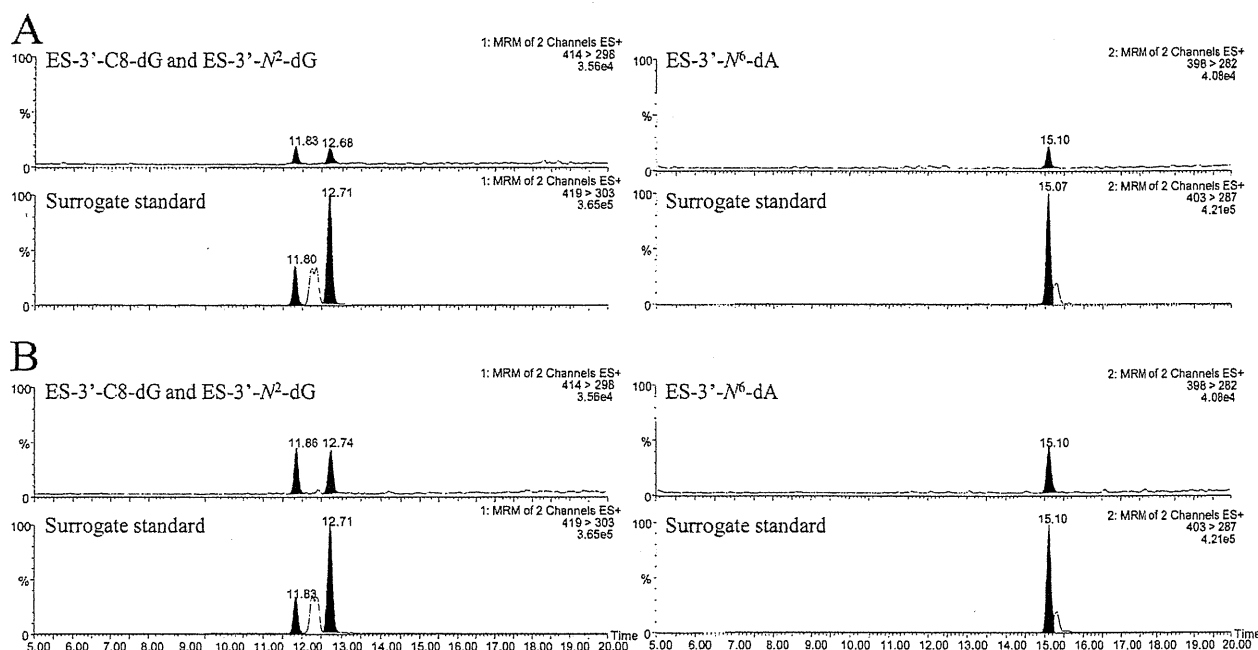


Figure 5. MRM chromatograms of LOD (A) and LOQ (B) levels of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA adducts and each surrogate standard from extracted DNA samples.

Table 2. Recoveries of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA Adducts in Rat Liver Samples

compounds	added (pmol/L)	concentration (pmol/L)	recovery (%)	RSD (%)
ES-3'-C8-dG (n = 5)	50	50.0 ± 2.3	98.0 ± 4.6	4.7
	500	501.1 ± 26.5	100.3 ± 5.3	5.3
	5000	4938.8 ± 135.5	98.8 ± 2.7	2.7
ES-3'-N ² -dG (n = 5)	50	50.6 ± 3.7	101.3 ± 7.5	7.4
	500	497.2 ± 18.6	99.4 ± 3.7	3.7
	5000	4981.2 ± 79.2	99.6 ± 1.6	1.6
ES-3'-N ⁶ -dA (n = 5)	50	51.7 ± 0.77	103.5 ± 1.5	1.5
	500	509.9 ± 7.29	102.0 ± 1.5	1.4
	5000	4988.0 ± 109.8	99.8 ± 2.2	2.2

Typical LC-PDA chromatograms are shown in Figure 1A,B. Two unknown peaks were observed in the PDA chromatogram (280 nm) after the reaction of 1'-acetoxy-ES with dA (Figure 1B). Simultaneously, mass spectra and UV-vis absorption spectra were obtained by ESI/MS (cone voltage 20, 40, and 60 V) and PDA. MS spectra of all unknown peaks were analyzed according to the ES and dA structure, and then, two peaks were analyzed in the chromatograms. One of the two peaks (peak 1) was observed at 22.1 min in the chromatogram of dA reaction (Figure 1B). The UV-vis absorption spectrum of peak 1 had λ_{\max} at 260 nm. In the mass spectra of peak 1 when the cone voltage was set at 20 V (Figure 2A), the precursor ion ($[M + H]^+$, m/z 398) and product ion (m/z 282), corresponding to an ES-adenine adduct following glycoside bond cleavage, were clearly observed. A product ion (m/z 147) corresponding to 2-allyl-4-methoxybenzene structure was observed in the mass spectra at 40 and 60 V (Figure 2B,C). The other peak (peak 2) was observed at 22.4 min in the chromatogram for the reaction of 1'-acetoxy-ES and dA (Figure 1B). The UV-vis absorption spectrum was virtually identical to peak 1 and also had at λ_{\max} at

260. In the mass spectra of peak 2 when the cone voltage was set at 20 and 40 V (Figure 2D,E), the precursor ion ($[M + H]^+$, m/z 398) and product ion (m/z 282), corresponding to ES-adenine adduct following glycoside bond cleavage, were clearly observed. A product ion (m/z 147) corresponding to 2-allyl-4-methoxybenzene structure was also observed in the mass spectra of 40 and 60 V (Figure 2E,F).

1'-Acetoxy-ES was used as an electrophilic synthon in large scale synthesis to identify the chemical structures of peaks 1 and 2 from dA, respectively. These adducts were synthesized repeatedly by reaction with each base until the quantity was sufficient for ¹H NMR after HPLC-UV chromatography. However, peak 2 could not be collected in sufficient amounts for structural analysis. ¹H NMR chemical shifts of peak 1 redissolved in DMSO-*d*₆ are shown in Table 1. On the basis of these data, we judged that peak 1 was ES-3'-N⁶-dA. Chemical structures of ES-3'-N⁶-dA and stable isotopically labeled compound as a surrogate standard for LC-ESI/MS/MS analysis are shown in Figure 3.

Optimal Conditions for LC-MS/MS Detection. Figure 4 shows the product ion spectra for ES-3'-C8-dG, ES-3'-N²-dG,

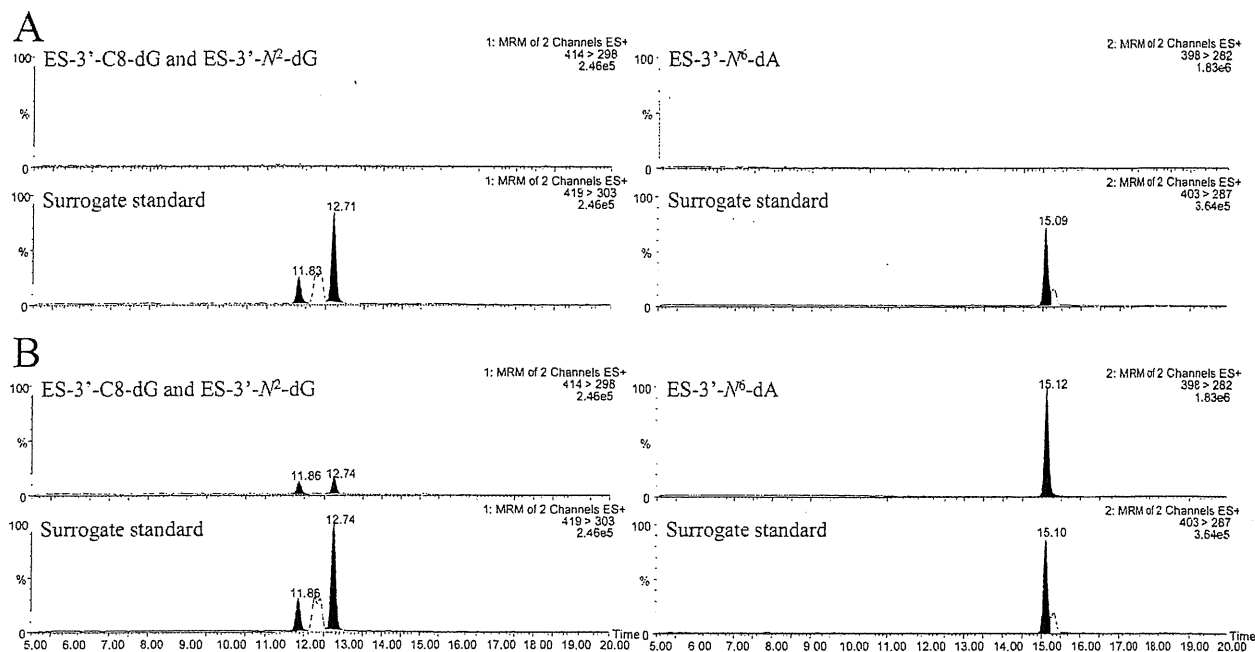


Figure 6. MRM chromatograms of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA adducts and each surrogate standard in rat liver DNA samples. (A) Liver from rat administrated with vehicle as a control group. (B) Liver from rat orally administrated with 600 mg/kg ES for 4 weeks.

Table 3. LC-MS/MS Analysis of Liver DNA Samples Obtained from Rats Treated with 600 mg/kg bw ES for up to 4 Weeks

group	sample no.	ES-3'-C8-dG/10 ⁶ dG	ES-3'-N ² -dG/10 ⁶ dG	ES-3'-N ⁶ -dA/10 ⁶ dA
control	1	ND ^a	ND	ND
	2	ND	ND	ND
	3	ND	ND	ND
	4	ND	ND	ND
	5	ND	ND	ND
	average			
ES (600 mg/kg)	6	3.02	4.16	18.96
	7	3.53	4.28	20.05
	8	3.42	4.37	21.22
	9	4.12	6.18	22.89
	10	3.33	5.07	20.47
	average	3.48 ± 0.40	4.81 ± 0.84	20.47 ± 1.61

^a ND, not detected.

and ES-3'-N⁶-dA. The mass spectrometer equipped with an ESI source using a crossflow counter electrode was run in positive ion mode to detect MRM of the transitions 414 > 298, 414 > 298, and 398 > 282, respectively.

The crucial parameters affecting LC-ESI/MS/MS, namely, cone voltage, collision energy, and mobile phase, were investigated. To establish the optimum cone voltage and collision energy for the detection of these adducts, the signals at *m/z* 414 and 419 precursor ions versus cone voltage were investigated, respectively. The optimal cone voltages were 18, 18, and 12 V in the negative ion mode for ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA standard solutions, respectively. Then, the *m/z* 298 and 282 product ion signals versus collision energy were investigated, respectively. The optimal collision energies were 10, 10, and 18 eV for ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA standard solutions, respectively. The ionization of the samples at

the LC-MS interface was affected by the mobile phase; hence, a mobile phase containing a volatile acid or salt was generally used. In this study, the responses were measured using 0–0.1% formic acid in water–acetonitrile (v/v) as the mobile phase. The responses of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA were increased by the addition of formic acid to the mobile phase. The increase in response reached a maximum and leveled off when 0.001% formic acid was added.

Validation of LC-MS/MS. The calculated instrument detection limit (IDL) of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA of the standard solutions was 3.0, 3.0, and 1.0 pM, respectively, for LC-MS/MS detection at the ratio of the compound's signal to the background signal (S/N) of 3. In addition, the instrument quantification limit (IQL) was calculated when S/N = 10 was 10, 10, and 3.0 pM, respectively. The limit of detection (LOD) and limit of quantification (LOQ) in the real sample were the same as IDL and

Table 4. Interday Precision for the Determination of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA Adduct Levels in the Rat Liver DNA Samples on Five Different Days Using LC-MS/MS

repeat analysis of ES-3'-8-dG adduct (nM)							
sample no.	day 1	day 2	day 3	day 4	day 5	mean (nmol/L)	RSD (%)
6	0.510	0.518	0.491	0.506	0.509	0.507 ± 0.010	1.96
7	0.431	0.443	0.450	0.432	0.459	0.443 ± 0.012	2.68
8	0.462	0.470	0.482	0.483	0.445	0.468 ± 0.016	3.39
9	0.474	0.490	0.480	0.458	0.476	0.476 ± 0.011	2.39
10	0.484	0.483	0.483	0.511	0.447	0.482 ± 0.023	4.74
repeat analysis of ES-3'-N ² -dG adduct (nM)							
sample no.	day 1	day 2	day 3	day 4	day 5	mean (nmol/L)	RSD (%)
6	0.701	0.711	0.733	0.703	0.727	0.715 ± 0.014	1.97
7	0.522	0.597	0.597	0.586	0.546	0.569 ± 0.034	5.91
8	0.590	0.626	0.608	0.638	0.605	0.613 ± 0.019	3.06
9	0.711	0.706	0.705	0.678	0.661	0.692 ± 0.022	3.15
10	0.737	0.676	0.717	0.706	0.718	0.711 ± 0.022	3.13
repeat analysis of ES-3'-N ⁶ -dA adduct (nM)							
sample no.	day 1	day 2	day 3	day 4	day 5	mean (nmol/L)	RSD (%)
6	3.100	2.967	3.098	3.044	2.949	3.032 ± 0.071	2.36
7	2.286	2.333	2.236	2.339	2.250	2.289 ± 0.047	2.04
8	2.699	2.617	2.738	2.797	2.766	2.723 ± 0.069	2.55
9	2.444	2.455	2.546	2.506	2.526	2.495 ± 0.044	1.77
10	2.736	2.735	2.846	2.880	2.830	2.805 ± 0.066	2.36

IQL, respectively. The peak area ratio with respect to each surrogate standard was plotted, and the response was found to be linear (ES-3'-C8-dG, $y = 1.9707x - 0.0096$ and $y = 2.0318x - 0.0989$; ES-3'-N²-dG, $y = 0.6359x + 0.0048$ and $y = 0.6336x - 0.0263$; and ES-3'-N⁶-dA, $y = 1.8567x + 0.012$ and $y = 1.8205x + 0.065$) over the calibration range, from LOQ to 1.0 nM (low range) and from 0.1 to 10 nM (high range), with a correlation coefficient (r) of over 0.999. The average retention times of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA standards were 11.8 (RSD = 0.08%, $n = 5$), 12.7 (RSD = 0.12%, $n = 5$), and 15.1 min (RSD = 0.06%, $n = 5$), respectively. MRM chromatograms of the mixture of three adduct and their corresponding surrogate standards in 50% methanol at LOD and LOQ levels are shown in Figure 5.

As shown in Table 2, the average recoveries of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA from DNA sample in the livers of nontreated rats were 99.0, 100.1, and 101.7% for each adduct.

Analysis of ES-N²-dG, C8-dG, and N⁶-dA in the Livers DNA Extracted from Rats Treated with ES. The potential formation of ES adducts in livers of rats exposed to ES was assessed using the isotope dilution LC-ESI/MS/MS method. Typical selected ion recording (SIR) chromatograms are shown in Figure 6. No peaks indicative of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA were observed in liver DNA extracted from control rats. In contrast, ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA were detected at 11.8, 12.7, and 15.1 min in the liver DNA MRM chromatograms extracted from rats treated with 600 mg/kg ES for up to 4 weeks, respectively. Quantitative data are summarized in Table 3. The ES-3'-C8-dG, ES-3'-N²-dG/10⁶dG, and ES-3'-N⁶-dA/10⁶dA ratios were 3.5 ± 0.4 , 4.8 ± 0.8 , and 20.5 ± 1.6 ,

respectively. The interday sample variation of ES-3'-C8-dG, ES-3'-N²-dG, and ES-3'-N⁶-dA adduct levels analyzed on different days resulted in an average coefficient of variation (CV) of 3.0, 3.4, and 2.2% for different samples that were analyzed (Table 4). All of these adducts were not detected in the control group.

DISCUSSION

Phillips et al. have suggested the chemical structure of four ES-specific DNA adducts, ES-3'-N²-dG, ES-3'-C8-dG, ES-1'-N²-dG, and ES-3'-N⁶-dA, using radio isotope-labeled nucleosides. The formation of these adducts in mouse liver treated with [¹⁴C]-labeled ES was demonstrated by radioactive detection after HPLC fractionation.⁸ Subsequently, LC-MS analysis by Punt et al. demonstrated the formation of three ES-dG adducts, ES-3'-N²-dG, ES-3'-C8-dG, and ES-1'-N²-dG, after in vitro nucleoside reaction.⁷ However, the existence of the dA adduct reported previously and its chemical structure have not been confirmed using MS technique. In the present study, we examine dA adduct formation using LC-ESI/MS by taking advantage of reactive carbocation formation generated by acetylation of ES proposed by Punt et al.⁷ In the reaction of dA with 1'-acetoxy-ES, two peaks including ions characteristic for the ES-dA adduct were observed by total ion chromatography. Subsequently, ¹H NMR analysis of large-scale synthesized and HPLC purified samples led to the conclusion that the major adduct was ES-3'-N⁶-dA, in line with the previous report.⁸ Although a minor and unknown ES-dA adduct was also found, its precise chemical structure could not be elucidated due to the low yield.

ES has been reported to have genotoxicity and carcinogenicity in the livers of mice of both sexes.² ES treatment at a dose of 600 mg/kg by gavage for 16 weeks induced significant developments of glutathione S-transferase placenta form (GST-P) foci in the liver of rats (unpublished data). To examine the precise quantity of ES-specific adducts in ES-hepatocarcinogenesis, we used interim samples (4 weeks) for the carcinogenicity study mentioned above. On the basis of previous reports, we attempted to detect ES-3'-N²-dG, ES-3'-C8-dG, and ES-3'-N⁶-dA. Because appropriate sample preparations and a highly sensitive analytical method are necessary to detect modified DNA bases in genomic DNA, we developed a new analytical method by an isotope dilution LC-ESI/MS/MS method using SIR. ³²P-postlabeling analysis without internal standards for identifying the products neither provides structural characterization of adducts nor has sufficient selectivities. In addition, this assay requires radioactive γ -³²P-labeled ATP in the analytical process, which raises the necessity of rigorous handling. The LC-MS/MS technique using stable isotope adducts as internal standards achieves accurate quantification of DNA adducts along with structural characterization. The LOQs of ES-3'-N²-dG, ES-3'-C8-dG, and ES-3'-N⁶-dA in our method were determined to be 2–5 adducts/10⁸ unmodified dG or dA bases from a 150 mg liver sample. In addition, the high recoveries of these adducts in the wide range from LOQ level indicated that our new method enables precise adduct determination with the use of surrogate standards and is applicable to the detection of these compounds in animal tissue samples. As a result, our method was able to quantify these three adducts in liver DNA of rats treated with 600 mg/kg ES for 4 weeks, and we observed that the ES-3'-N⁶-dA adduct is predominant. Phillips et al. have suggested that ES-3'-N²-dG was a major adduct in the livers of mice treated with single dose of 1'-hydroxy-ES (12 μ mol/mouse), a metabolite of ES. Considering that the present experiment was performed under carcinogenic conditions, it is highly probable that the status of DNA adduct formation observed in the present study may be reflected in the ES carcinogenesis. The modification at the N⁶-position of dA has been reported following treatment with other potent mutagens including aristolochic acid,^{22,23} polyaromatic hydrocarbon, benzo[*c*]phenanthrene,^{29,30} and 5,6-dimethylchrysene.³¹ It is likely that the formation of various types of chemical-specific base modifications is dependent on accessibility of the chemical (or the proximate form) to the reactive amino group between dG and dA in the DNA helical structure.³² Those mutagens predominantly induce AT-TA transversion and AT-GC transition mutations.^{25,33,34} Therefore, our unpublished data that the AT-GC transition mutation was predominant in ES-treated rat livers allow us to hypothesize that N⁶-dA adducts might play a key role in ES mutagenicity. Thus, information regarding the precise concentrations of ES-specific DNA adducts discovered in the present study would be very helpful for further research on ES hepatocarcinogenesis.

Minor adducts of dG (ES-1'-N²-dG) and dA (not identified) were hardly detected in *in vivo* samples even though such adducts are detectable in the reaction of 1'-acetoxy-ES with deoxynucleoside.⁷ Dissociation of the ester group from an ester of 1-hydroxy-ES generates an electrophilic ion in which the positive charge may reside on the double bond between the 2-, 3- and 1-, 2-positions on the allyl chain. Nucleophilic attack by the purine bases of DNA at the 1'-position of ES would account for ES-1'-N²-dG formation, while attack at the 3'-position would account for ES-3'-N²-dG. Because dG modification at the 3'-position of

ES far from the benzene ring is predominant both *in vivo* and *in vitro*, it is conceivable that the priority of these reactions is determined by steric hindrance between DNA bases and ES. The suggestion that unknown minor dA adducts also might result from modification at the 1'-position of ES is in line with this hypothesis. Therefore, the fact that genomic DNA possesses more steric hindrance than nucleosides might explain the lack of detection of these minor adducts *in vivo*.

In conclusion, dA modified by ES was determined to be EG-3'-N⁶-dA as the major adduct. ES-3'-N²-dG, ES-3'-C8-dG, and ES-3'-N⁶-dA adducts can be identified and quantified by this new method, which may prove useful in other related studies.

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■ ABBREVIATIONS

ES, estragole; dG, deoxyguanosine; dA, deoxyadenosine; LC, liquid chromatography; ESI, electron spray ionization; MS, mass spectrometry; NMR, nuclear magnetic resonance; MS/MS, tandem mass spectrometry; SIR, selected ion recording; LOQ, limit of quantification; SULT, sulfotransferase; THF, tetrahydrofuran; DMSO, dimethylsulfoxide; DMF, dimethylformamide; TMS, trimethylsilane; MRM, multiple reaction monitoring; PDA, photodiode array; IDL, instrument detection limit; IQL, instrument quantification limit; LOD, limit of detection; GST-P, glutathione S-transferase placenta form.

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