

Table 2 Salivary metabolite marker candidates ($P < 0.05$ Steel–Dwass test) and the ratio of the relative area of oral, breast and pancreatic cancers, and periodontal diseases to controls ($n = 215$)

Metabolite	m/z		Identification (*1)	HMDB (*2)	p - Value	Control vs				
	Theoretical	Measured				Error (ppm)	cancer	breast cancer	pancreatic cancer	periodontal disease
$C_2H_6N_2$	59.0604	59.0616	21		0.260	$4.17 \times 10^{-6}***$	$2.67 \times 10^{-4}***$	$3.46 \times 10^{-4}***$	$1.51 \times 10^{-8}***$	
$C_{32}H_{48}O_{13}$ (*4)	214.4438	214.4440	1.0		0.834	$0.00385**$	0.271	0.907	$4.75 \times 10^{-4}***$	
$C_3H_7NO_2$	90.0550	90.0553	3.7		0.955	0.167	$1.40 \times 10^{-6}***$	0.0242*	0.154	
$C_4H_{12}N_5$	131.1165	131.1174	6.8		0.686	$0.00290**$	$3.37 \times 10^{-6}***$	0.0251*	0.323	
$C_4H_9NO_2$	104.0706	104.0706	-0.35		0.620	0.913	$5.78 \times 10^{-6}***$	0.779	1.00	
Cadaverine	103.1230	103.1231	1.4	Standard	0.0422*	$0.00889**$	$0.00104**$	$0.00100**$	0.993	
$C_3H_{11}NO_2$ (10.05 min.)	118.0863	118.0864	0.86		0.910	0.408	0.192	0.0308*	0.416	
$C_{30}H_{62}N_{19}O_5S_3$ (*3)	409.2321	409.2312	-2.2		0.367	$2.39 \times 10^{-5}***$	$4.64 \times 10^{-4}***$	0.0141*	0.0354*	
$C_{30}H_{55}N_{27}O_5S$ (*3)	437.7424	437.7442	4.2		0.428	0.0305*	0.0114*	0.672	0.532	
$C_{18}H_{32}N_6O_6$ (*3)	215.1264	215.1269	2.2		0.891	0.386	0.0387*	0.997	0.143	
Alpha-Aminobutyric acid	104.0706	104.0705	-1.0	Standard	0.00256**	0.0269*	$4.37 \times 10^{-6}***$	0.00796**	1.00	
Alanine	90.0550	90.0551	1.1	Standard	$2.45 \times 10^{-4}***$	0.0277*	$4.57 \times 10^{-6}***$	0.00647**	0.956	
Putrescine	89.1073	89.1075	1.5	Standard	0.890	$0.00108**$	$1.91 \times 10^{-4}***$	0.0227*	0.138	
Methylimidazoleacetic acid (*5)	141.0659	141.0660	0.02	Predicted	0.183	0.390	$5.32 \times 10^{-4}***$	0.213	1.00	
Trimethylamine (*5)	60.0808	60.0814	11	Predicted	0.609	0.921	$4.97 \times 10^{-4}***$	$1.99 \times 10^{-4}***$	1.00	
$C_3H_{14}N_5$	145.1322	145.1332	6.9		0.0212*	0.0700	$2.28 \times 10^{-4}***$	0.0825	0.998	
Piperidine	86.0964	86.0966	2.4	Standard	0.00119**	0.934	0.0347*	0.137	0.165	
Taurine	126.0219	126.0221	1.0	Standard	$3.57 \times 10^{-5}***$	$5.94 \times 10^{-6}***$	$2.36 \times 10^{-4}***$	$6.70 \times 10^{-5}***$	$6.82 \times 10^{-10}***$	
Piperidine (*5)	84.0808	84.0807	-0.41	Predicted	2.83 $\times 10^{-4}***$	0.0128*	0.569	3.19 $\times 10^{-4}***$	$2.79 \times 10^{-6}***$	
Pipecolic acid	130.0863	130.0861	-0.93	Standard	1.87 $\times 10^{-4}***$	0.264	0.943	0.00948**	$3.47 \times 10^{-4}***$	
C_4H_9N	72.0808	72.0813	7.8		$2.02 \times 10^{-7}***$	0.729	$3.37 \times 10^{-6}***$	0.949	$5.62 \times 10^{-4}***$	
C_8H_9N	120.0808	120.0801	-5.9		$2.64 \times 10^{-5}***$	$4.60 \times 10^{-4}***$	1.00	0.0385*	$1.02 \times 10^{-8}***$	
Pyroline hydroxycarboxylic acid (*5)	130.0499	130.0498	-0.49	Predicted	$1.28 \times 10^{-5}***$	0.141	0.228	0.176	$1.36 \times 10^{-5}***$	
Betaine	118.0863	118.0864	0.86	Standard	0.0162*	0.719	0.987	0.576	0.0183*	
C_4H_7N	70.0651	70.0655	5.8		0.101	0.637	0.997	0.502	0.0349*	
$C_6H_6N_2O_2$	139.0502	139.0500	-1.8		0.00270**	0.265	$1.65 \times 10^{-10}***$	0.127	0.948	
Leucine + Isoleucine	132.1019	132.1019	-0.30	Standard	$1.56 \times 10^{-5}***$	0.00136**	$7.47 \times 10^{-9}***$	0.00150**	1.00	
Phenylalanine	166.0863	166.0863	0.46	Standard	0.00333**	0.0149*	$2.95 \times 10^{-7}***$	0.0198*	1.00	
Tyrosine	182.0812	182.0812	0.16	Standard	0.0253*	0.00908**	$1.00 \times 10^{-5}***$	0.0279*	0.926	

Table 2 continued

Metabolite	m/z		Identification (*1)	HMDB (*2)	p-Value					
	Theoretical	Measured (ppm)			Control vs oral cancer	Control vs breast cancer	Control vs pancreatic cancer	Control vs periodontal disease	Oral cancer vs breast cancer	
Histidine	156.0768	156.0766	-1.2	Standard	HMDB00177	$6.87 \times 10^{-4}***$	0.350	0.00141**	0.246	0.805
Proline	116.0705	116.0705	-0.19	Standard	HMDB00162	0.968	0.00165**	0.00115**	0.0598	0.0299*
Lysine	147.1128	147.1127	-1.0	Standard	HMDB00182	0.0779	$9.56 \times 10^{-4}***$	$5.22 \times 10^{-6}***$	0.459	0.426
Glycine	76.0393	76.0397	4.8	Standard	HMDB00123	1.00	0.0140*	0.00216**	0.193	0.0753
Ornithine	133.0972	133.0971	-0.22	Standard	HMDB00214	0.293	0.0114*	0.0220*	0.0801	0.652
C ₁₇ H ₂₈ N ₄ O ₅	367.1976	367.1977	0.21			0.981	0.0990	0.0309*	0.0561	0.430
Pro-Gly-Pro or Pro-Pro-Gly (*6)	270.1449	270.1452	1.1	Predicted		0.143	0.635	0.00165**	0.425	0.990
C ₇ H ₁₂ N ₂ O ₃	173.0921	173.0919	-1.1			0.892	0.854	0.00289**	0.902	0.999
Butiramide (*5)	213.1168	213.1233	30	Predicted		0.999	0.266	0.0439*	0.950	0.224
Ethanolamine	62.0600	62.0601	1.1	Predicted	HMDB00149	0.684	0.0483*	$2.43 \times 10^{-7}***$	0.00187**	0.597
Gamma-Aminobutyric acid	104.0706	104.0706	-0.35	Standard	HMDB00112	0.833	0.139	$2.10 \times 10^{-7}***$	0.0133	0.897
Aspartic acid	134.0448	134.0447	-0.34	Standard	HMDB00191	0.287	0.00564**	$1.56 \times 10^{-7}***$	0.403	0.416
Valine	118.0863	118.0862	-0.72	Standard	HMDB00883	$7.31 \times 10^{-5}***$	0.0204*	$1.94 \times 10^{-8}***$	0.0138	0.990
Tryptophan	205.0972	205.0972	-0.091	Standard	HMDB00929	$6.13 \times 10^{-5}***$	0.402	$2.99 \times 10^{-8}***$	0.0113*	0.229
Beta-Alanine	90.0550	90.0551	1.1	Standard	HMDB00056	0.0407*	0.722	$3.56 \times 10^{-6}***$	0.268	0.842
Citrulline	176.1030	176.1028	-0.80	Standard	HMDB00904	0.148	0.0509	0.00420**	0.130	0.960
Glutamic acid	148.0604	148.0603	-1.1	Standard	HMDB00148	$4.95 \times 10^{-4}***$	0.00528**	$1.49 \times 10^{-8}***$	0.0757	1.00
Threonine	120.0655	120.0654	-0.64	Standard	HMDB00167	$1.18 \times 10^{-4}***$	0.00790**	$5.72 \times 10^{-8}***$	$1.80 \times 10^{-4}***$	1.00
Serine	106.0499	106.0499	0.33	Standard	HMDB00187	0.0197*	0.0119*	$3.42 \times 10^{-7}***$	0.00699**	0.846
Glutamine	147.0764	147.0764	-0.43	Standard	HMDB00641	0.0327*	0.368	$1.07 \times 10^{-6}***$	0.111	0.975
Hypoxanthine	137.0458	137.0457	-0.96	Standard	HMDB00157	0.207	0.616	$4.93 \times 10^{-6}***$	0.00279**	0.107
Choline (*5)	104.1070	104.1070	0.22	Predicted	HMDB00097	$2.30 \times 10^{-5}***$	1.00	$1.91 \times 10^{-4}***$	0.0580	0.0115*
Carnitine	162.1125	162.1124	-0.38	Standard	HMDB00062	$7.60 \times 10^{-4}***$	0.956	0.00341**	0.996	0.247
C ₅ H ₁₁ NO ₂ (13.42 min.)	118.0863	118.0863	0.33			0.0606	0.958	$3.73 \times 10^{-4}***$	0.915	0.754
Glycerophosphocholine	258.1107	258.1121	5.5	Standard	HMDB00086	0.287	$4.53 \times 10^{-5}***$	0.00263**	0.0322*	$7.05 \times 10^{-6}***$
C ₇ H ₈ O ₃ S	173.0267	173.0285	10			0.962	$1.88 \times 10^{-5}***$	0.00553**	0.0154*	$2.71 \times 10^{-4}***$
C ₄ H ₈ N ₂ O ₁₁ P	288.9704	288.9691	-4.3			0.0421*	0.0815	0.0776	0.256	$5.17 \times 10^{-6}***$

Table 2 continued

Metabolite	p-Value		Ratio of relative average to controls						
	Oral cancer vs pancreatic cancer	Oral cancer vs periodontal disease	Breast cancer vs pancreatic cancer	Breast cancer vs periodontal disease	Pancreatic cancer vs periodontal disease	Oral cancer	Breast cancer	Pancreatic cancer	Periodontal disease
C ₂ H ₆ N ₂	2.34 × 10 ^{-6***}	6.26 × 10 ^{-6***}	0.995	0.936	0.999	0.577	3.96	4.42	5.28
C ₃₂ H ₄₈ O ₁₃ (*4)	0.137	0.444	0.951	0.467	0.970	0.954	2.21	2.03	1.11
C ₃ H ₇ NO ₂	3.68 × 10 ^{-4***}	0.0662	0.0350*	0.748	0.744	1.24	1.62	3.00	2.16
C ₄ H ₁₂ N ₅	0.00636**	0.469	0.132	0.928	0.661	2.69	2.46	6.98	2.34
C ₄ H ₉ NO ₂	0.00504**	0.995	0.0138*	0.995	0.214	1.73	1.88	5.16	1.71
Cadaverine	0.449	0.488	0.569	0.781	1.00	5.20	2.21	6.15	2.71
C ₅ H ₁₁ NO ₂ (10.05 min.)	0.270	0.169	0.976	0.543	0.897	1.51	1.73	2.10	2.35
C ₃₀ H ₆₂ N ₁₉ O ₂ S ₃ (*3)	0.0812	0.420	1.00	0.992	0.973	2.18	10.3	6.65	3.40
C ₃₀ H ₅₅ N ₂₇ O ₃ S (*3)	0.225	0.993	0.954	0.980	0.790	2.06	5.47	5.56	2.11
C ₁₈ H ₃₂ N ₆ O ₆ (*3)	0.0247*	0.914	0.657	0.917	0.578	1.34	2.39	6.70	0.942
Alpha-Aminobutyric acid	0.0543	0.811	0.0321*	0.885	0.299	3.44	2.26	4.01	2.46
Alanine	0.0968	0.945	0.0447*	0.543	0.573	3.91	1.94	3.67	1.92
Putrescine	0.0399*	0.444	0.652	0.999	0.717	3.53	2.80	3.98	2.10
Methylimidazoleacetic acid (*5)	0.286	0.973	0.365	0.903	0.632	4.19	2.46	4.64	2.18
Trimethylamine (*5)	0.146	0.0643	0.0256*	0.00788**	0.977	4.80	1.36	4.13	3.75
C ₃ H ₁₄ N ₅	0.471	0.994	0.211	0.889	0.839	8.31	2.20	5.35	2.69
Piperidine	1.00	0.972	0.190	0.730	0.992	18.4	1.20	3.73	1.46
Taurine	5.82 × 10 ^{-7***}	1.10 × 10 ^{-6***}	0.994	0.612	0.793	2.35	0.358	0.376	0.00
Piperidine (*5)	0.00226**	2.91 × 10 ^{-5***}	0.740	0.304	0.0296*	2.43	0.509	0.665	0.113
Pipecolic acid	0.175	1.97 × 10 ^{-4***}	0.330	0.529	0.0161*	2.25	0.833	1.11	0.358
C ₄ H ₉ N	0.996	0.00766**	0.00286**	0.992	0.00416**	4.81	1.18	2.70	0.846
C ₈ H ₉ N	0.0123*	2.94 × 10 ^{-4***}	0.00358**	0.995	0.0379*	3.23	0.209	0.817	0.168
Pyroline hydroxycarboxylic acid (*5)	0.867	0.00267**	0.0170*	0.947	0.0400*	4.15	0.454	2.28	0.184
Betaine	0.133	0.0668	0.998	0.999	0.913	2.04	0.864	0.822	0.637
C ₄ H ₇ N	0.853	0.0858	0.825	0.998	0.632	2.21	0.742	1.14	0.517
C ₆ H ₆ N ₂ O ₂	0.0715	0.974	0.0460*	0.827	0.429	1.61	1.44	2.89	1.42
Leucine + Isoleucine	7.44 × 10 ^{-4***}	0.868	0.00272**	0.855	0.00574**	4.65	3.05	7.71	2.19
Phenylalanine	0.00351**	0.961	0.0122*	0.967	0.0165*	2.25	1.78	3.54	1.66
Tyrosine	0.0286*	0.969	0.235	1.00	0.215	1.84	1.99	2.90	1.49
Histidine	0.698	0.997	0.275	0.990	0.717	1.70	1.35	2.02	1.29
Proline	0.0171*	0.291	0.825	0.995	0.689	1.63	2.48	3.99	1.58
Lysine	0.00513**	1.00	0.226	0.730	0.0190*	1.84	2.96	3.97	1.22
Glycine	0.0122*	0.352	0.731	0.998	0.689	1.38	2.32	3.10	1.67
Ornithine	0.516	0.912	0.986	1.00	0.941	1.69	2.13	1.97	1.43

Table 2 continued

Metabolite	p-Value											
	Oral cancer vs pancreatic cancer	Oral cancer vs periodontal disease	Breast cancer vs pancreatic cancer	Breast cancer vs periodontal disease	Breast cancer vs pancreatic cancer	Breast cancer vs periodontal disease	Pancreatic cancer vs periodontal disease	Pancreatic cancer vs periodontal disease	Oral cancer	Breast cancer	Pancreatic cancer	Periodontal disease
C ₁₇ H ₂₆ N ₄ O ₅	0.248	0.357	0.917	0.997	0.913	0.997	0.913	0.913	2.19	2.63	2.69	1.52
Pro-Gly-Pro or Pro-Pro-Gly (*6)	0.413	1.00	0.205	0.972	0.660	0.972	0.660	0.660	4.25	2.65	4.27	2.00
C ₇ H ₁₂ N ₂ O ₃	0.0627	0.995	0.238	1.00	0.602	1.00	0.602	0.602	1.80	1.65	2.48	1.48
Burimamide (*5)	0.0675	0.841	0.779	0.961	0.631	0.961	0.631	0.631	1.68	2.32	2.56	1.42
Ethanolamine	2.34 × 10 ^{-4***}	0.132	0.00404**	0.486	0.401	0.486	0.401	0.401	1.66	1.84	4.35	2.23
Gamma-Aminobutyric acid	0.00108**	0.367	0.0140	0.712	0.215	0.712	0.215	0.215	1.82	1.62	3.31	2.07
Aspartic acid	4.37 × 10 ^{-5***}	0.980	0.0114*	0.932	0.00672**	0.932	0.00672**	0.00672**	1.53	1.70	4.10	1.26
Valine	0.00325**	0.999	0.00549**	0.967	0.00785**	0.967	0.00785**	0.00785**	4.53	2.64	5.92	1.75
Tryptophan	0.0461*	1.00	1.83 × 10 ^{-4***}	0.319	0.0424*	0.319	0.0424*	0.0424*	4.26	1.59	6.47	2.44
Beta-Alanine	0.156	0.999	0.00471**	0.730	0.375	0.730	0.375	0.375	2.22	1.51	3.04	1.74
Citrulline	0.291	0.880	0.865	0.987	0.992	0.987	0.992	0.992	1.98	2.30	3.10	1.65
Glutamic acid	0.00312**	1.00	0.00214**	1.00	0.00574**	1.00	0.00574**	0.00574**	2.87	2.12	4.80	1.54
Threonine	3.08 × 10 ^{-4***}	0.0829	0.00141**	0.166	0.162	0.166	0.162	0.162	2.15	1.71	4.75	2.19
Serine	1.16 × 10 ^{-4***}	0.435	0.00920**	0.868	0.0868	0.868	0.0868	0.0868	1.74	1.66	4.34	1.63
Glutamine	0.00228**	0.998	0.00167**	0.903	0.0251*	0.903	0.0251*	0.0251*	2.35	1.59	4.96	1.39
Hypoxanthine	0.0917	0.452	1.94 × 10 ^{-4***}	0.00237**	0.952	0.00237**	0.952	0.952	2.97	0.839	3.35	2.30
Choline (*5)	0.871	0.983	0.00374**	0.265	0.429	0.265	0.429	0.429	2.98	1.05	2.63	1.51
Carnitine	0.652	0.670	0.0895	1.00	0.335	1.00	0.335	0.335	2.10	1.21	1.88	1.11
C ₅ H ₁₁ NO ₂ (13.42 min.)	0.0687	0.930	0.0161*	0.999	0.0544	0.999	0.0544	0.0544	2.13	1.06	2.17	1.03
Glycerophosphocholine	7.33 × 10 ^{-4***}	0.0138*	1.00	1.00	1.00	1.00	1.00	1.00	2.78	0.00	0.00	0.00
C ₇ H ₈ O ₃ S	0.0176*	0.0307*	0.864	1.00	0.860	1.00	0.860	0.860	0.972	0.494	0.592	0.467
C ₄ H ₅ N ₂ O ₁₁ P	0.628	0.808	6.92 × 10 ^{-4***}	0.0116*	1.00	0.0116*	1.00	1.00	0.418	1.96	0.00	0.00

* $P < 0.05$; ** $P < 0.01$; *** $P < 0.001$ (Steel-Dwass test)

(*1) The peaks marked "standard" were identified based on the matched m/z values and normalized migration times of the corresponding standard compounds. The peaks marked "predicted" were obtained from Kyoto Encyclopedia of Gene and Genomics (KEGG) database (<http://www.genome.jp/kegg/>) and HMDB (<http://www.hmdb.ca/>) by matching the composition formula and migration times that were calculated from the isotope distribution and molecular structure, respectively

(*2) HMDB; accession numbers for the Human Metabolome Database

(*3) The peaks were detected as bivalent ions

(*4) The peaks were detected as trivalent ions

(*5) The normalized measured and predicted migration times were 8.16 min and 7.78 min (0.38 min) for ethanolamine, 11.61 min and 10.42 min (1.19 min) for burimamide, 8.75 min and 7.78 min (0.97 min) for choline, 8.16 min and 7.78 min (0.38 min) for ethanolamine, 10.012 min and 10.10 min (0.98 min) for methylimidazoleacetic acid, 8.78 min and 7.84 min (0.94 min) for piperidine, 12.90 min and 13.09 min (0.19 min) for pyrroline hydroxycarboxylic acid, and 7.61 min and 8.26 min (0.65 min) for trimethylamine. The parent values are differences between the predicted and the measured time

(*6) Because we did not find any candidate compounds for the 270.1454 m/z and 12.23 min peaks in the databases, we considered Pro-Gly-Pro, Pro-Pro-Gly or Gly-Pro-Pro as possible candidates ([M + H]⁺ = 270.3044 m/z in all cases). The predicted migration time of Pro-Gly-Pro and Pro-Pro-Gly was 11.33 min and that of Gly-Pro-Pro was 10.98 min; therefore, Gly-Pro-Pro was excluded as a candidate

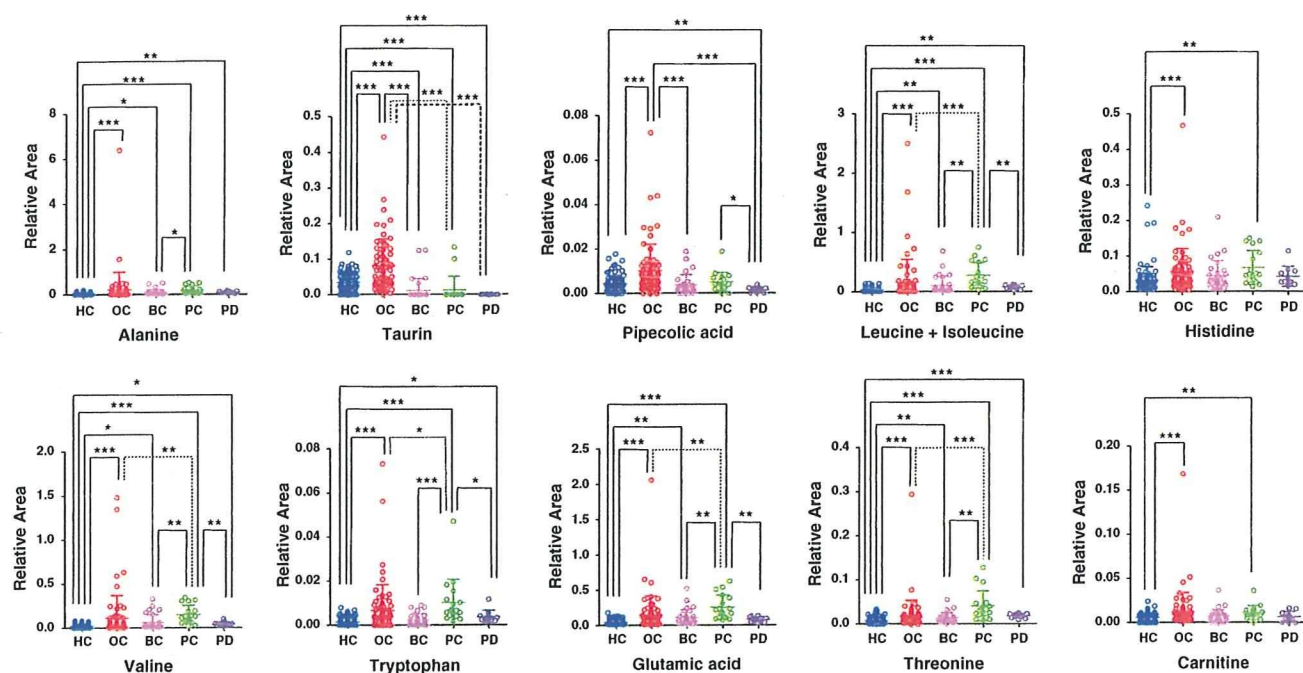
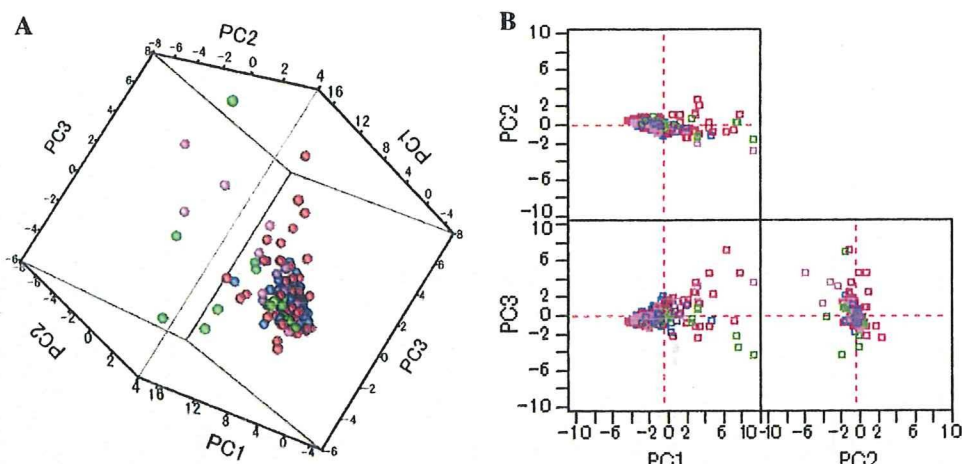


Fig. 2 Representative dot plots for the relative area of detected metabolites in samples from all groups. The colored dots denote healthy controls (blue), oral (red), breast (pink), pancreatic cancer (green), and periodontal disease (purple). The Y- and X-axes denote the relative peak area (no units) and the group name, respectively. The horizontal, center long bars and the short top/bottom bars indicate the

means and standard deviations, respectively. The stars indicates * $P < 0.05$, ** $P < 0.01$, and *** $P < 0.001$ (Steel-Dwass test). Only metabolites showing a significant difference between oral cancer and controls at $P < 0.001$ and matched with standard library are displayed. The dot plots of the other metabolites are shown in Supplementary Fig. S1

Fig. 3 Score plots of principal components (PC) analyses. The subjects in all groups are shown in 3-dimensional (a) and 2-dimensional (b) plots without outliers. The cumulative proportions of the first, second and third PCs (PC1, PC2, and PC3) were 44.8, 57.6 and 67.0%. The same analyses presented for all datasets are shown in Supplementary Fig. S2



concentration of putrescine and cadaverine are decreased in cancer patients undergoing radiotherapy but remain higher than those in healthy individuals (Khuhawar et al. 1999). There were no significant differences in urinary polyamine levels between the healthy individuals and breast cancer patients; however, the levels of putrescine, spermine and other metabolites were significantly higher in patients with breast cancer (Byun et al. 2008). Oral polyamine levels are also affected by periodontitis and gum healing (Silwood et al. 2002). We found that the levels of ornithine and

putrescine were higher in patients with breast or pancreatic cancer, and were markedly higher in patients with oral cancer, than in our healthy controls, while there was no significant difference between patients with periodontal disease and the controls. Although the quantitative level of polyamines is associated with regulation of tumor growth and with periodontitis, our results indicate that salivary polyamines are affected by the cancer type and by periodontitis, and that their levels were markedly higher in patients with oral cancer.

Fig. 4 ROC curve analysis of the ability of salivary metabolites to discriminate between samples from patients with **a** oral ($n = 69$), **b** breast ($n = 30$) or **c** pancreatic cancer ($n = 18$), and **d** samples from patients with periodontal diseases ($n = 11$) and the controls ($n = 87$). The *solid (red)* and *dotted (blue)* ROC curves were obtained using the complete data as a training set and with a tenfold cross-validation, respectively. Using a cut-off probability of 50%, the calculated area under the ROC curves were 0.865 (0.810) for oral, 0.973 (0.881) for breast and 0.993 (0.944) for pancreatic cancer, and 0.969 (0.954) for periodontal diseases. The non-parenthetic values were obtained with the full-training data and parenthetic values by tenfold cross-validation

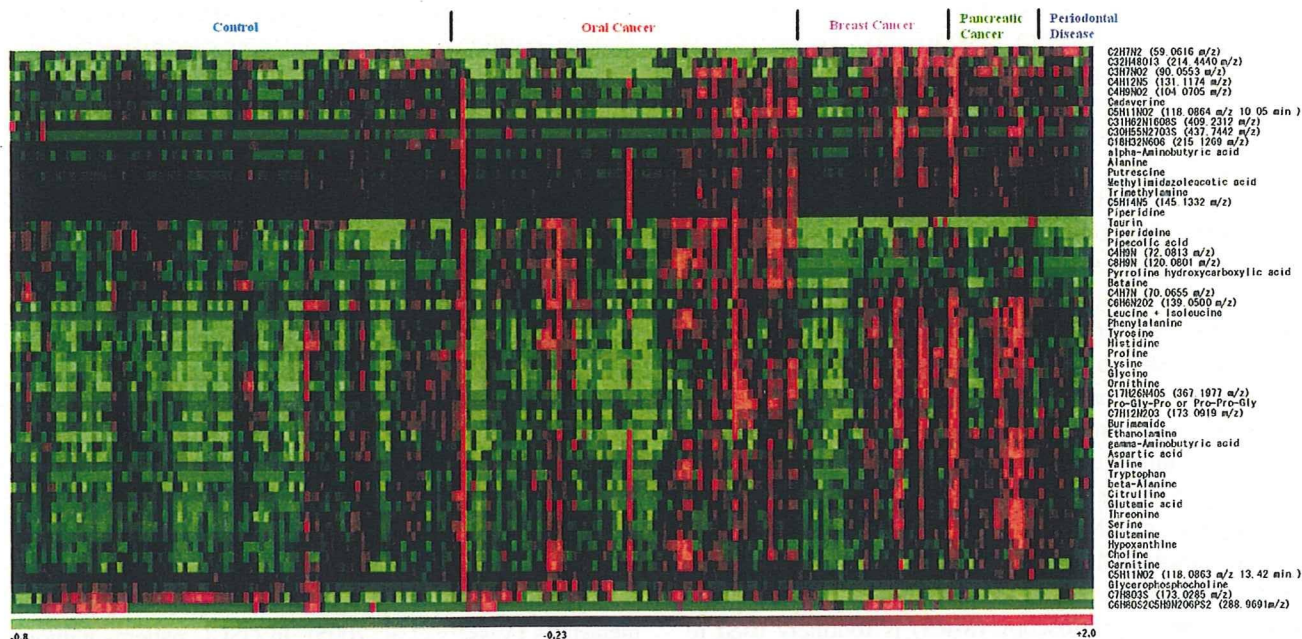
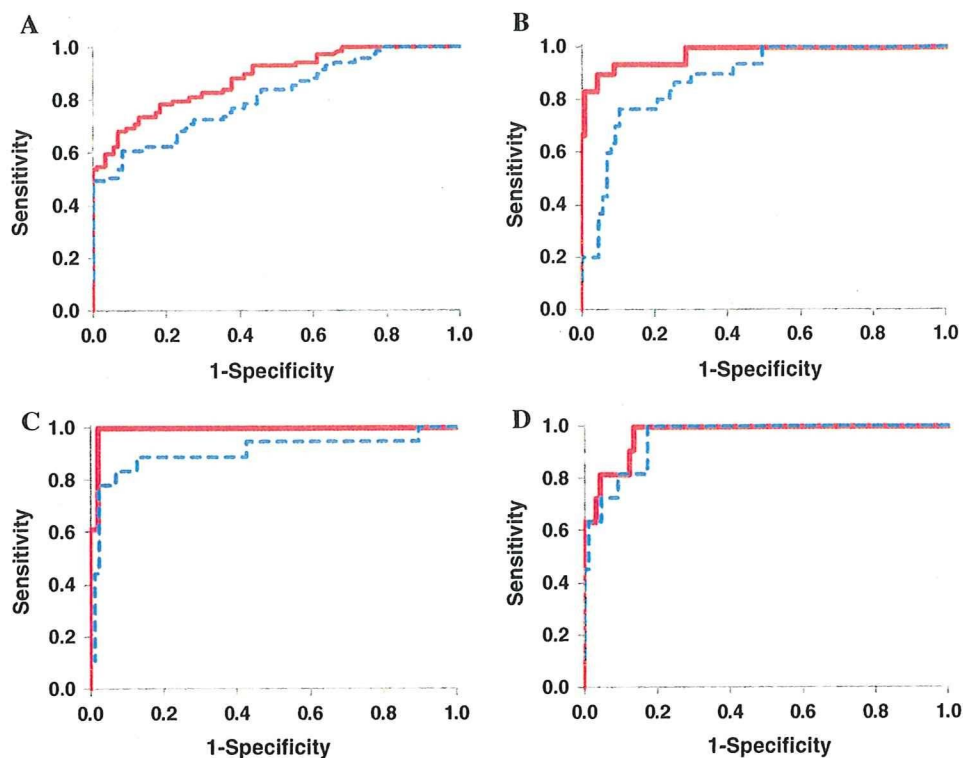


Fig. 5 Heat map of 57 peaks showing significantly different levels ($P < 0.05$; Steel–Dwass test) between control samples ($n = 87$) and samples from patients with at least one disease ($n = 128$). Each row

shows data for a specific metabolite and each column shows an individual. The colors correspond to the relative metabolite areas that were converted to Z-scores

In addition to polyamines, the level of tryptophan (Carlin et al. 1989), which is increased in oral and pancreatic cancer, is a direct marker for tumor development. In terms of an indirect connection between the detected metabolites and human cancer, the repeat peptide Pro-Pro-

Gly, which is expressed at high levels in breast cancer, is an inhibitor of matrix metalloproteinase-2 (MMP-2, gelatinase A), which plays an important role in tumor invasion and metastasis (Jani et al. 2005). The expression levels of the amino acid transporters ACST2 and LAT1 are elevated

in primary human cancers, and cancer cells optimize their metabolic pathways by activating the extra- to intracellular exchange of amino acids. Peptides and acids are derived from various sources, such as fragmented proteins, and the saliva metabolome profiles comprising these compounds may reflect the integrated results.

A significantly decreased level of arginine was observed in plasma samples from several cancers including breast, colonic and pancreatic cancer, which might be due to increased uptake of arginine by tumor tissues with high arginase activity (Vissers et al. 2005). However, salivary arginine was hardly changed, and there were no differences among the groups (Supplementary Fig. S3 and Table S2). A trend for decreasing levels of amino acids, including leucine, isoleucine, valine and alanine, has been reported in pancreatic cancer samples (Fang et al. 2007). The levels of amino acids in breast cancer tissue samples showed similar patterns, with low levels of isoleucine, leucine, lysine and valine (Yang et al. 2007). The decreased amino acid levels appear to be the result of enhanced energy metabolism or upregulation of the appropriate biosynthetic pathways, and required cell proliferation in cancer tissues. However, the observed salivary amino acid levels showing significant differences in the cancer groups (in Table 2) were higher than in the controls. The heterogeneous systems that transport amino acids from blood to saliva via the salivary gland, such as kinetic differences, or the dependence or independence of small ions such as potassium and sodium (Mann and Yudilevich 1987), altered the concentration of these ions because of water movement through the paracellular route (Melvin 1999) or channels (Ishikawa and Ishida 2000). Metabolism in the salivary gland itself might also play a major contribution to the differences in profiles between saliva and blood. Further validation of these findings by comparing saliva profiles with blood and tissue profiles is needed to understand the reason for the different saliva amino acid profiles.

Choline, a quaternary amine, is an essential nutrient that is predominantly supplied by the diet, and choline-containing metabolites are important constituents of phospholipid metabolism of cell membranes and are associated with malignant transformation, such as breast, brain and prostate cancers (Ackerstaff et al. 2003). Magnetic resonance spectroscopy (MRS) is routinely used to quantify choline-based metabolism in malignancies such as head and neck cancer and breast cancers (Bolan et al. 2003). Choline is highly metabolized in tumors to phosphocholine and is also highly oxidized to betaine; hence, the low concentration of choline and high concentrations of phosphocholine and betaine (Katz-Brull et al. 2002) were observed. Furthermore, the levels of choline metabolites were higher in tumors than in benign lesions or normal tissues (reviewed in Haddadin et al. 2009). In tumor cells,

an excessive increase in plasma choline levels in patients with breast cancer was also shown (Katz-Brull et al. 2001). Aberrant choline metabolism can be explained as a result of enhanced membrane synthesis and degradation, which represent excessive proliferation of cancer cells. Pancreatic cancer tissue had a unique profile showing decreased levels of phosphocholine and glycerophosphocholine, but not choline (Fang et al. 2007). We found that the levels of phosphocholine (Supplementary Fig. S3 and Table S2) and glycerophosphocholine (Table 2) were increased in the saliva samples from oral cancer patients and were decreased in the other groups.

Creatine phosphate acts as a store for high-energy phosphates. Therefore, its concentration might be altered in energy-demanding tissues (Maheshwari et al. 2000). Previous studies showed an increase in the choline-creatinine ratio in tumor tissues and in the serum of patients with OSCC (Maheshwari et al. 2000; Tiziani et al. 2009). Creatine is converted to creatine phosphate by creatine kinase. Increased creatine phosphate levels were also found in other tumors, such as breast and gastrointestinal tract tumors. In our study, the salivary choline level was significantly higher in subjects with oral and pancreatic cancers ($P = 2.30 \times 10^{-5}$ and $P = 1.91 \times 10^{-4}$, respectively; Steel–Dwass test), but not in the other groups. Therefore, the salivary choline–creatinine ratio showed oral cancer-specific elevation (Supplementary Table S2 and Fig. S3). However, this finding needs to be interpreted with care because choline is a nutrient present in most foods.

Compared with oral cancer, breast and pancreatic tumors are physically remote from the oral cavity. Therefore, it can be questioned why salivary metabolite profiles reflect the aberrant localized tumour metabolism. Systemic biofluids, such as blood and lymph fluid, are one of the routes that readily bypass these tumors and the salivary gland, which blends saliva with contaminating blood. Several metabolites in tumor tissues, such as lactate, which is derived from tumor exposed to hypoxia, were altered both with and without metastasis (Hirayama et al. 2009; Walenta et al. 2000). Although abnormal arginine levels in breast cancer without metastasis were observed, the same metabolic changes were shown in a pooled group of patients with colonic and pancreatic cancer with/without metastasis (Vissers et al. 2005). In OSCC patients without metastasis from the primary tumor, cancer-specific changes in serum and salivary mRNA levels (Li et al. 2006; Pickering et al. 2007) and blood metabolome levels (Tiziani et al. 2009; Zhou et al. 2009) were shown. Although this does not constitute direct proof that the aberration in salivary metabolites is attributed to a remote tumor, evidence that the salivary metabolite profiles reflects the systemic and localized tumor status or its response to chemotherapies, such as breast and lung cancer, has accumulated

(Emekli-Alturfan et al. 2008; Gao et al. 2009; Harrison et al. 1998; Streckfus et al. 2006, 2008). Although previous studies have demonstrated an increase in choline metabolites in blood in various cancers, the increase in choline metabolites in oral cancer patients in study indicate that the transportation of these metabolites from the blood to the saliva through the salivary gland is low, even though their levels in blood are elevated. Alternatively, these metabolites were diffused from the oral malignancy to the salivary gland via a route other than the blood vessel. We acknowledge that the current study merely mined the data and showed that the changes in salivary metabolites had cancer-specific features. Further biological studies to compare the metabolite profiles obtained concurrently from saliva, blood and cancer tissue is needed to provide rational evidence for the systemic metabolite links.

3.3 Bias derived from clinical parameters

We evaluated the metabolite bias introduced by relevant clinical parameters (age, gender, race and ethnicity). The PCA score plots showed poor separation between male and female subjects among healthy controls and patients with oral cancer (Supplementary Fig. S4). Statistical comparisons of the relative area are presented in Supplementary Table S3. Takeda et al. (2009) measured the gender-specific differences in salivary metabolites and found that formate, lactate, propionate and taurine were significantly higher in males. Compared with these metabolites, the gender-specific level of taurine, which was the only metabolite observed in our measurement condition, showed little difference between the subjects in the control and oral cancer groups. By contrast, in the control group, tyrosine and a metabolite at 214.4440 m/z were significantly higher in females than in males ($P = 0.0492$ and $p = 0.0261$, respectively; Mann–Whitney test). In the oral cancer group, threonine and serine were significantly higher in males and piperidine was higher in females ($P = 0.0340$, $P = 0.0462$, and $P = 0.0221$, respectively; Mann–Whitney test). Takeda et al. (2009) discussed that these gender-specific differences might be attributed to dental care, hormones such as estrogen, and oral pathogenesis carriers such as microflora. Indeed, infection of the oral environment with viruses such as human papillomavirus or microorganisms is known to be a risk factor for the development of oral cancer (Meurman and Uittamo 2008). Although we found that the gender-specific differences in metabolic profiles differed between the tumor types, the number of metabolites showing significant differences was low, which implies that the disease-specific variation is predominantly embedded in the 57 metabolites identified here.

In the control and oral cancer groups, the PCA based on race and ethnicity were visualized using score plots

(Supplementary Fig. S5) and the statistical analytical results are presented in Supplementary Table S4. In the control group, there were no significant differences between African-Americans and Caucasians, or African-Americans and Hispanics. Meanwhile, 11 and 12 significantly different ($P < 0.05$; Steel–Dwass test) metabolites were observed between African-Americans and Asians, and Asians and Caucasians, respectively. Similarly, the profiles between Asians and Hispanics, and Caucasians and Hispanics revealed three and seven significantly different metabolites ($P < 0.05$; Steel–Dwass test). Of particular note, levels of putrescine, proline, glycine and unannotated metabolites at 118.0864 m/z and 10.05 min were low in Asians, while the level of burimamide was high in African-Americans. A country-dependant bias in human urinary metabolite profiles has also been reported elsewhere (Holmes et al. 2008). In their study, positively charged metabolites, such as alanine-related metabolites, showed discriminative characteristics and were correlated with several dietary factors such as energy intake, dietary cholesterol and alcohol intake. However, in our study, there were no differences in alanine levels in either the control or the oral cancer subjects. In the control group, there were no differences in 34 out of 57 marker candidates among the race or ethnic groups. In subjects with oral cancer, only a metabolite at 211.4440 m/z showed a significant difference ($P = 0.0386$; Steel–Dwass test). Although biases based on race or ethnicity were found in the 57 metabolic profiles, the number of the metabolites showing significant differences were less than the number of peaks showing significant differences in cancer-specific profiles, which implies that this bias might be more moderate than disease-specific differences.

Age-related differences have been reported in a transcriptome study of the salivary gland (Srivastava et al. 2008). The coefficients of regression lines for age and relative area for all 57 metabolite markers are presented in Supplementary Table S5. It has been reported that other commonly used methods for standardization of metabolites in biofluid yield different statistical results (Schnackenberg et al. 2007); therefore, consistent decreases or increases in levels of metabolites among subjects with correlated clinical parameters should be accounted for. In the control subjects and patients with pancreatic cancer, there was a positive correlation between metabolites and age, whereas the opposite was true for patients with oral or breast cancer or periodontal diseases. Accordingly, it is unlikely that age is correlated with the concentrations of salivary metabolites.

Several limitations in this study need to be acknowledged. First, the metabolite profiles in saliva might fluctuate to similar or greater levels compared with other *omics* profiles, such as the proteome and transcriptome, in

response to systemic conditions such as stress, and oral conditions including gingival crevicular fluid and oral microbiota (reviewed in Fabian et al. 2008). Therefore, the reproducibility of the sample collection protocol used in this study should be rigorously verified under various conditions. Circadian rhythms in salivary flow rate and components have been reported (Dawes 1972). Levels of putrescine and cadaverine, which correlate with oral mal-odor, were markedly altered during waking time, even in healthy donors (Cooke et al. 2003). Although, the samples were collected within a limited period of time in the morning, levels of these metabolites were generally higher in patients with most types of cancer in the present study. The variance in the concentrations of these metabolites should be validated in future studies. Another external factor that alters saliva contents is the time-course of fluoride concentration, which has been tracked, and the changes in concentrations continued for 30 min after eating food (Hedman et al. 2006). Therefore, the 1-h period before sample collection should be evaluated in terms of food intake. Smoking is also known to affect salivary metabolites such as citrate lactate, pyruvate and sucrose (Takeda et al. 2009). The metabolites identified in this study could not be compared with these metabolites because they were not positively charged in our measurement condition. Therefore, the profiles of positively charged metabolites should be explored in further analyses.

Second, the sample sizes, particularly the number of patients with breast or pancreatic cancer or periodontal diseases, were relatively small. A larger cohort, including samples from an independent institute, would allow for statistical comparisons with greater power and a more rigorous validation. In addition, samples from patients with systemic diseases showing similar symptoms, such as oral leukoplakia and oral cancer (Zhou et al. 2009), chronic pancreatitis and pancreatic cancer (Fang et al. 2007; Kojima et al. 2008), should be compared with evaluate the sensitivity and specificity of the detected metabolites. In this study, the patients' age was collected for all samples and only a few additional parameters, namely sex and race, or ethnic group, were collected for the control and oral cancer group. Analyses and validation studies taking into account the complete clinical and pathological parameters, including menopausal status, estrogen and progesterone receptors for breast cancer, and risk factors including smoking and alcohol drinking for oral cancers are essential before actual diagnostic application of the classification model obtained in this study. In this study, although we used stepwise feature selection and an MLR model to identify classifiers, other feature selection and classification methods are also applicable, such as regression tree models (Li et al. 2004, 2006) and concurrent use of ANN with SVM (Ayers et al. 2004). Instead of developing a

classification model only based on the salivary metabolome profiles of matched subjects, the construction of a marker model incorporating related clinical features or risk factors and biomarkers can be used to visualize the probability of a specific diseases status; for example, nomograms are a commonly used strategy (Brennan et al. 2004; Gross et al. 2008; Katz et al. 2008).

A metabolomic study using serum samples from patients with oral cancer showed stage-specific profiles (Tiziani et al. 2009). The profiles obtained in this study were simply categorized into the type of cancer. Therefore, future studies are needed that integrate histological and clinical features. Simultaneous analyses of the metabolic profiles in blood and tissue collected from the same patients are also needed to track the biological sources of the disease-specific signatures in salivary metabolite profiles. Although there are still several limitations to be addressed, the methodology used in this study to detect salivary metabolite profiles are not limited to early diagnosis but offer the potential to aid the characterization of malignant neoplasms or tumors by integrating histological or clinical features, such as staging.

4 Concluding remarks

This is the first study to comprehensively analyze salivary metabolites and to identify metabolic profiles specific to oral, breast and pancreatic cancers. A larger number of patient samples, particularly those from different institutes, and additional clinical variables are needed for further validation and future clinical application of our method. In addition, integrating the knowledge obtained from other *omics* studies may help us to understand the biological basis of these disease-specific metabolic profiles.

In conclusion, our study has demonstrated that CE-TOF-MS can readily and effectively be applied to salivary metabolomics. We have proposed an alternative use for salivary diagnosis to be applied for the detection of oral, breast and pancreatic cancers.

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Measurement of internal body time by blood metabolomics

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Detection of internal body time (BT) via a few-time-point assay has been a longstanding challenge in medicine, because BT information can be exploited to maximize potency and minimize toxicity during drug administration and thus will enable highly optimized medication. To address this challenge, we previously developed the concept, “molecular-timetable method,” which was originally inspired by Linné’s flower clock. In Linné’s flower clock, one can estimate the time of the day by watching the opening and closing pattern of various flowers. Similarly, in the molecular-timetable method, one can measure the BT of the day by profiling the up and down patterns of substances in the molecular timetable. To make this method clinically feasible, we now performed blood metabolome analysis and here report the successful quantification of hundreds of clock-controlled metabolites in mouse plasma. Based on circadian blood metabolomics, we can detect individual BT under various conditions, demonstrating its robustness against genetic background, sex, age, and feeding differences. The power of this method is also demonstrated by the sensitive and accurate detection of circadian rhythm disorder in jet-lagged mice. These results suggest the potential for metabolomics-based detection of BT (“metabolite-timetable method”), which will lead to the realization of chronotherapy and personalized medicine.

chronotherapy | circadian | metabolome | jet-lag | LC-MS

In the 18th century, the Swedish botanist Karl von Linné designed a “flower clock” comprising a series of plant species arranged according to the respective time of the day their flowers open or close. Watching this flower clock, one can estimate the time of the day by noting the pattern of flower opening and closing. Since Linné’s early times, it has been a well known fact that plants have an internal clock and thereby can open or close their flowers at a precise time of the day. Similarly, animals possess an internal molecular mechanism, a “circadian clock,” which underlies endogenous, self-sustained oscillations with a period of ≈ 24 h manifest in diverse physiological and metabolic processes (1). In mammals, several clock genes, including *Clock*, *Bmal1*, *Per1*, *Per2*, *Cry1*, *Cry2*, *RevErbA*, *Rora*, *Csnk1e*, *Csnk1d*, and *Fbxl3*, regulate, at least in part, gene expression in central and/or peripheral clock tissues (2–4). Reflecting the temporal changes in gene expression in central and peripheral clock tissues (5–8), the potency and/or toxicity of administered drugs depend on the individual’s present body time (BT) (9–13). It has been suggested that administering a drug at a specific BT improves the outcome of pharmacotherapy by maximizing its potency and minimizing its toxicity (14). In contrast, administering a drug at an inappropriate BT can result in severe side effects (15). Despite the importance of such BT-dependent therapy (also known as “chronotherapy”) (9–13), its application to clinical practice has been obstructed by a lack of clinically feasible methods for measuring BT.

To overcome this problem, we previously developed the concept of a “molecular-timetable method (16),” which was originally

inspired by Linné’s flower clock. In Linné’s flower clock, one can estimate the time of the day by watching the opening and closing pattern of various flowers. Similarly, in molecular-timetable method, one can measure the BT of the day by profiling the up and down pattern of substances in the molecular timetable. This concept was proven using the expression profile of clock-controlled genes in a target organ (16). However, estimates of BT from the expression profile of oscillating substances within a target organ (in this case, the liver) are hard to apply directly to clinical situations. To make the molecular-timetable method more clinically relevant, we decided to determine BT from blood samples, which are more available in clinical practice.

In the blood of mammals, several small chemical substances such as metabolites and hormones have been reported to exhibit circadian oscillations. For example, the concentration of the steroid hormone, corticosterone, is rhythmically controlled by circadian clock with a peak in the evening (17), and an amine-derived hormone, melatonin, show circadian rhythm with a peak in the early morning in mice (18). In humans, several peptide hormone levels show daily variations; growth hormone increases during sleep (19), leptin increases during the evening (20), and prolactin increases at night (21). Concentrations of amino acids, including tryptophan, tyrosine, phenylalanine (22), methionine (23), cysteine, glutathione (24), and homocysteine (25), also exhibit daily variations in human blood plasma. Despite these findings, comprehensive profiling of circadian dynamics of chemical substances in mammalian blood has not yet been reported, and until now a comprehensive molecular timetable of such chemical substances has not been constructed.

Metabolomics technology aims to comprehensively identify and/or quantify the dynamic chemical substances present in biological samples. It is gaining interest in the fields of drug discovery, disease diagnostics, and treatment (26–28). The present metabolomics technology was developed rapidly by coupling advanced separation technology with highly sensitive and selective mass spectrometry–gas chromatography mass spectrometry (GC/MS) (29–31), liquid chromatography mass spectrometry (LC-MS) (32–34), and capillary electrophoresis mass spectrometry (CE-MS) (35, 36). To construct the molecular timetable from clinically available samples, we have performed blood metabolome analysis in this study. Using the LC-MS technique, we quantified hundreds of

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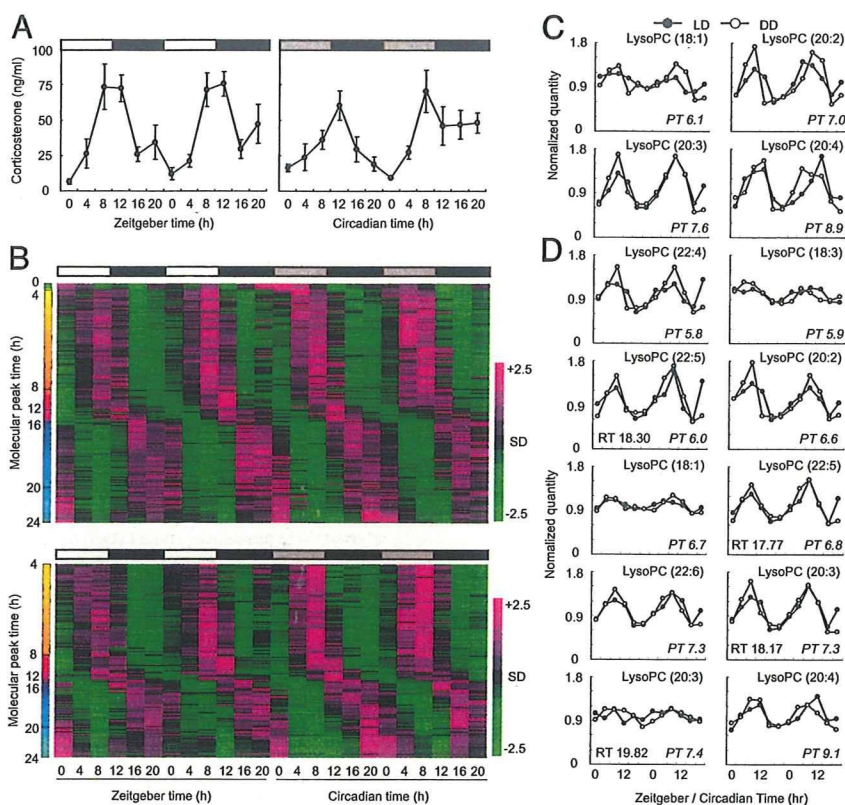


Fig. 1. Circadian patterns of metabolites in mouse plasma. (A) Circadian changes in corticosterone levels in the plasma of CBA/N mice under LD (Left) and DD, Center) conditions. All values are mean \pm SEM. The white bars above the graph indicate day, gray bars indicate subjective day, and black bars indicate night/subjective night. ZT0 is the time of light on, and CT0 is the time the light used to be turned on. (B) Circadian oscillatory metabolites in the plasma of CBA/N mice [negative ions (up); positive ions (down)]. On the heat maps, magenta tiles indicate a high quantity of substances and green tiles indicate a low quantity in plasma. Metabolites are sorted according to their molecular peak time (molecular peak times are indicated as colors). (C and D) Identified oscillatory peaks measured by negative ion mode (C) and positive ion mode (D). Mean value was set to 1.0.

clock-controlled metabolites in mouse plasma and successfully constructed a molecular timetable of blood metabolites. This metabolite timetable allowed us to measure individual BT under various conditions and was robust enough to be used in mice with different genetic backgrounds, sex, age, and feeding conditions. It was also sensitive and accurate enough to detect circadian rhythm disorders in jet-lagged mice. Our preliminary results suggest that other metabolomics techniques such as CE-MS can also be applied to the molecular-timetable method, demonstrated by the quantification of hundreds of clock-controlled metabolites, the identification of substantial portion of these metabolites, and successful measurement of BT from independent blood samples. Thus, metabolomics-based measurement of BT will contribute to the potential areas of chronotherapy and personalized medication regimens.

Results

Construction of the Metabolite Timetable from Blood Plasma by LC-MS. Samples of blood plasma were taken from young male CBA/N mice every 4 hours over 2 days during light–dark (LD) or constant darkness (DD) conditions. Plasma corticosterone was used as a quality control because it exhibits a clear circadian oscillation when quantified by radio immunoassay (Fig. 1A). Small chemical substances in the plasma were quantified by LC-MS analysis to construct the metabolite timetable. LC-MS analysis detected 695 negative ion and 938 positive ion peaks in the plasma. Of these, 176 negative and 142 positive ion peaks exhibited significant circadian oscillations in LD and DD conditions [Fig. 1B; false discovery rate (FDR) < 0.01 ; see also *Materials and Methods*]; these peaks accounted for the $\approx 19.5\%$ of the peaks detected in mouse plasma. These substances served as “time-indicating metabolites,” because they oscillate considerably even under constant environmental conditions (DD). For instance, at zeitgeber time 0 (ZT0; the beginning of day) or circadian time 0 (CT0; the beginning of a subjective day), concentrations of dawn-indicating metabolites, which peak at approximately ZT0 or CT0 (Fig. 1B, green color bars

in the molecular peak time), are high, whereas those of dusk-indicating metabolites, which peak at approximately ZT12 or CT12 (Fig. 1B, red color bars in the molecular peak time), are low. Conversely, at ZT12 or CT12, concentrations of dawn-indicating metabolites are low, whereas those of dusk-indicating metabolites are high; this suggests that time-indicating metabolites can represent BT (BT), the endogenous state of circadian clock. In fact, the oscillations of these time-indicating metabolites are directly or indirectly controlled by circadian clock, because the disruption of functional molecular clock in *Cry1*^{-/-}, *Cry2*^{-/-} mice (37) results in the alteration of circadian oscillations of these metabolites (Fig. S1). We used these LC-MS data to construct the molecular timetable of time-indicating metabolites (a “metabolite timetable”) in mouse plasma (Table S1 online). We also note that, among these time-indicating metabolites (i.e., oscillatory peaks detected by LC-MS), 14 oscillatory peaks were identified as various types of lysophosphatidylcholines with different unsaturated fatty acids (Fig. 1C and D).

Measurement of BT from Independent Samples. To verify whether the metabolite timetable was a good indicator of BT, we attempted to estimate the BT from the metabolite profiles of independently sampled mice. We collected fresh blood plasma from individual young male CBA/N mice every 4 h over 24 h under both LD or DD conditions because of the possibility that sampling time and/or light conditions would affect the accuracy of BT estimation. LC-MS analysis was performed to profile the time-indicating metabolites in the plasma samples (Fig. 2A and B). After measured profile of the time-indicating metabolites was normalized by using the metabolite timetable, we filtered out outliers, fitted the normalized profile to cosine curve, and calculated the significance of its fitness (see also *Materials and Methods*). This metabolite-timetable method could successfully detect significant circadian rhythmicity in all metabolite profiles of these samples ($P < 0.01$, Fig. 2A and B). The estimated BT closely matched with the environmental time when sampled (ZT under LD condition or CT under LD condition) with estima-

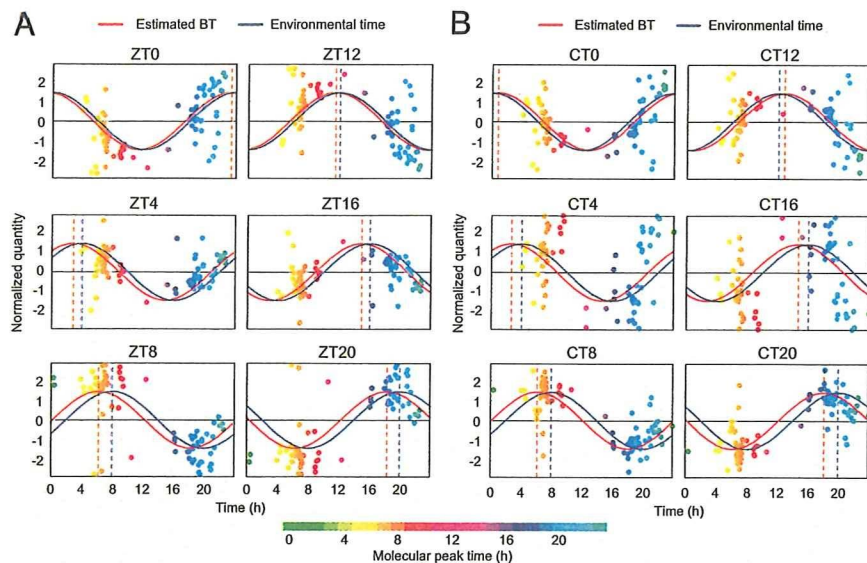


Fig. 2. BT estimation. BT measurements of mice kept under LD (A) and DD (B) conditions. Colors of the dots indicate the molecular peak times of each substance (Table S1). Peak time of the red cosine curves indicates estimated BT and peak time of the blue indicates the time the sample was taken (“environmental time”). The greater the degree of overlap of the red and blue curves, the greater the accuracy of the measurement. The dashed vertical lines show the BT (red) or environmental time (ZT/CT, blue). See Table S2 for statistics.

tion errors of 1.0 ± 0.49 h for LD and 1.3 ± 0.45 h for DD (mean \pm SD, Table S2). Estimation error was here defined as time difference between estimated BT and sampling time (environmental time). These results suggest that BT can be accurately determined from the metabolite profiles of independently sampled mice.

Differences in Genetic Backgrounds. In clinical situations, methods for BT detection must apply to populations with heterogeneous genetic backgrounds. To demonstrate the suitability of the metabolite-timetable method for individuals with different genetic backgrounds, we applied the method to other inbred mouse strain with genetic backgrounds that differed from the original CBA/N strain. We chose C57BL/6, because C57BL/6 and CBA/N are genetically remote from each other and classified into 2 completely different clusters among 55 mice strains according to SNP-based study (38). We collected the blood plasma samples from individual young male C57BL/6 mice every 4 h over 24 h under LD and DD conditions and quantified the time-indicating metabolites in the plasma by LC-MS (Fig. 3 A and B). The metabolite-timetable method detected significant circadian rhythmicity ($P < 0.01$) in all metabolite profiles both under LD (Fig. 3A) and DD conditions (Fig. 3B) even if we

used the metabolite timetable constructed from CBA/N mice. The estimated BT closely matched with the environmental time with the estimation errors of 1.6 ± 0.36 h for LD and 1.7 ± 0.24 h for DD (mean \pm SD, Table S2). These results suggest that BT can be accurately determined from the metabolite profiles of mice with heterogeneous genetic backgrounds.

Differences in Age and Sex. We constructed the metabolite timetable from young male mice only, so it was possible that age and sex factors might affect the accuracy of the metabolite-timetable method. To determine the influence of age and sex, we also applied the metabolite-timetable method to aged male and young female mice of the same strain. Blood plasma from individual aged male or young female CBA/N mice was sampled at 2 time points, ZT0 (the beginning of the day, i.e., time of light on) and ZT12 (the end of the day, i.e., time of the light off) under LD condition. These time points were considered as 2 “noisiest” time points, because light conditions were dramatically changed at these points. Time-indicating metabolites in the plasma were quantified by LC-MS (Fig. 4A) and significant circadian rhythmicity ($P < 0.01$) was detected in all metabolite profiles of both the aged male mice and

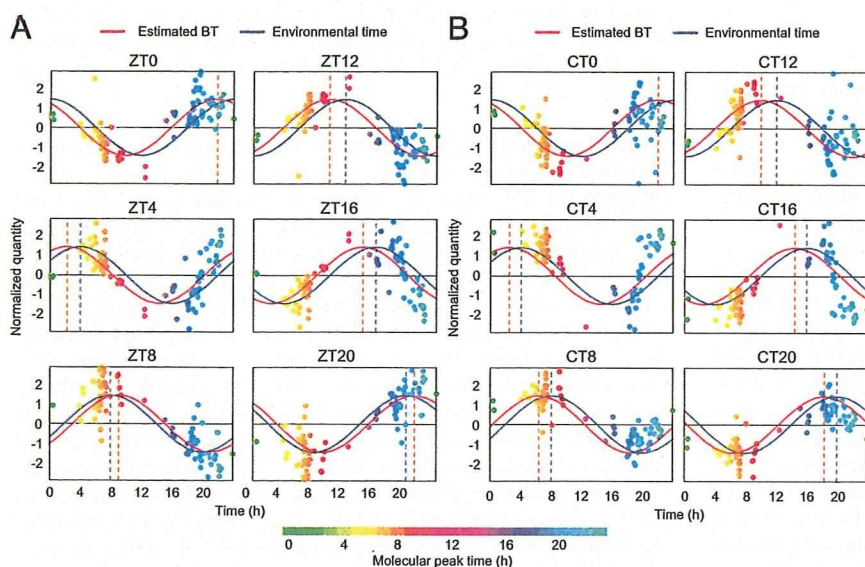


Fig. 3. Genetic background. BT measurement using C57BL/6 mice plasma collected under LD (A) and DD (B) conditions. Colors of the dots indicate the molecular peak times of each substance (Table S1). Peak time of the red cosine curves indicate estimated BT and peak time of the blue indicate the environmental time. The dashed vertical lines show the BT (red) or environmental time (ZT/CT, blue). See Table S2 for statistics.

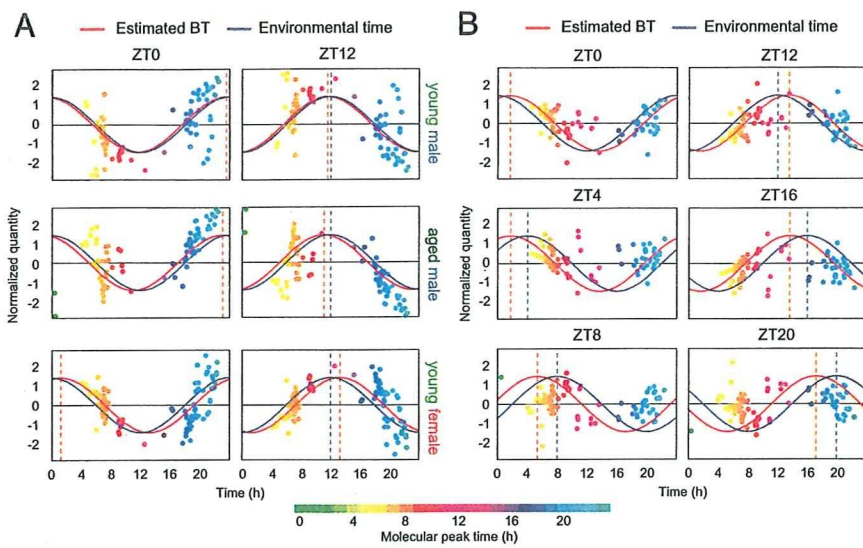


Fig. 4. Age, sex differences and feeding condition. (A) BT measurement of young male (Top), aged male (Middle), and young female mice plasma (Bottom) harvested at ZT0 (Left) and ZT12 (Right). (B) BT measurement of young male mice kept under food-deprivation conditions. Colors of the dots indicate the molecular peak times of each substance (Table S1). Peak time of the red cosine curves indicates estimated BT, and peak time of the blue indicates the environmental time. The dashed vertical lines show the BT (red) or environmental time (ZT, blue). Results for young male mice (A) are replotted from Fig. 2.A for comparison. See also Table S2 for statistics.

the young female CBA/N mice (Fig. 4A). The estimated BT from individual mice sampled at ZT0 and ZT12 were BT23.0 and BT11.0 in aged male mice and BT1.2 and BT13.2 in young female mice (Table S2). These results demonstrate that BT can be accurately determined from the metabolite profiles of mice of different age and sex.

Differences in Feeding Conditions. The circadian rhythmicity of food intake is well known (39); therefore, feeding conditions may severely affect the accuracy of the metabolite-timetable method. To validate the use of the metabolite timetable in individuals with different feeding conditions, we applied the metabolite-timetable method to CBA/N mice deprived of food (food deprivation). This feeding condition differed greatly from the original feeding condition where CBA/N mice were allowed ad lib feeding. We collected the blood plasma from individual young, male, food-deprived CBA/N mice every 4 hours over 24 h under LD condition. LC-MS analysis was performed to quantify the plasma metabolites (Fig. 4B). The metabolite-timetable method detected significant circadian rhythmicity ($P < 0.03$) in all metabolite profiles. The estimated BT matched with the environmental time with the estimation errors

of 2.2 ± 0.50 h (mean \pm SD, Table S2). These results suggest that BT can be determined from the metabolite profiles of mice even under severe feeding conditions.

Detection of Jet Lag. The final stage of the study was to evaluate the use of the metabolite-timetable in the diagnosis of circadian rhythm disorders. Jet lag is a common disorder of circadian rhythm, in which there is a difference between the internal BT and environmental time. To mimic jet lag, we kept the mice for 2 weeks under normal LD conditions and then rapidly advanced the lighting schedule by 8 h (40). Plasma samples were analyzed at 2 time points (ZT0 and ZT12 of the original LD cycle, termed as “Time 1” and “Time 2”) on 3 separate days: on day 1 (before entrainment to the new cycle), day 5 (during entrainment), and day 14 (after entrainment) (Fig. 5A and B). On day 1, the estimated BTs were 23.8 h (Time 1) and 11.8 h (Time 2), suggesting that the internal BTs still follow the original LD cycle. By day 14, estimated BTs were 8.8 h (Time 1) and 20.8 h (Time 2), suggesting that the internal BTs had shifted by ≈ 8 h from the original LD cycle and had therefore become entrained completely to the advanced cycle. Notably, on day 5, estimated BTs were 3.5 h (Time 1) and 15.5 h (Time 2), a shift

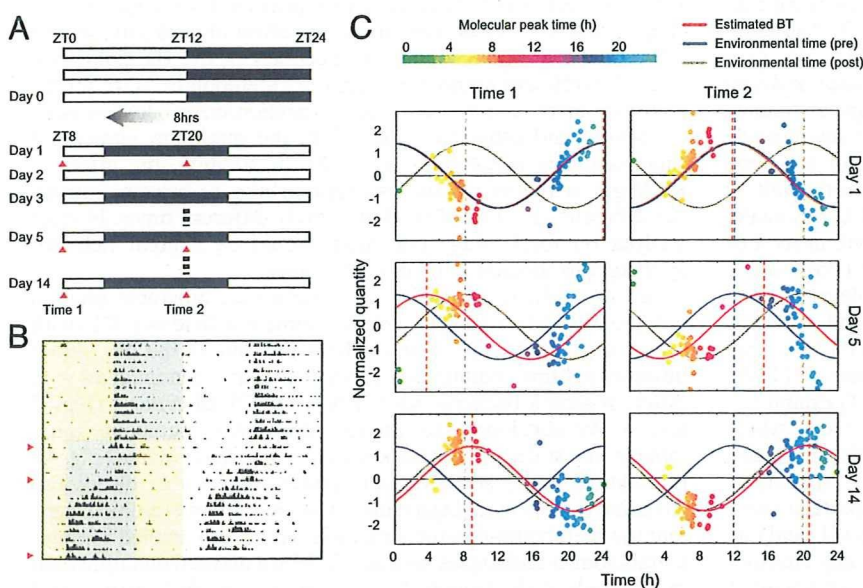


Fig. 5. Detection of jet lag. (A) Schematic view of lighting conditions. White bars indicate light on, and black bars indicate light off. On day 1, the light was turned off 8 h earlier. Samples were collected at 2 time points on days 1, 5, and 14 after LD shift (day 14, Lower). (B) The actogram of a single mouse, showing that it was experiencing “jet lag” induced by the LD shift. Yellow shading indicates periods of light on, and gray shading indicates periods of light off. The red triangles indicate days 1 (Top), 5 (Middle), and 14 (Bottom). (C) BT measurement from mouse plasma collected before (day 1, Top), during (day 5, Middle), and after entrainment to the new LD cycle (day 14, Lower). Colors of the dots indicate the molecular peak time of each substance (Table S1). The red cosine curve is the estimation, the blue curve is the environmental time (pre LD condition shift), and the brown cosine curve is the environmental time (post shift). See also Table S2 for statistics.

of 3.5 h from the original LD cycle, indicating incomplete entrainment to the advanced cycle, i.e., jet lag (Fig. 5C and Table S2). These results suggest that the metabolite-timetable method can accurately detect circadian rhythm disorders. Another set of BT estimation data for jet-lagged mice supports this conclusion (Fig. S2 and Table S2).

General Applicability of the Metabolite-Timetable Method: CE-MS-Based Method. As described above, the metabolite-timetable method based on LC-MS analysis can accurately measure the individual's BT and sensitively diagnose circadian rhythm disorders such as jet lag. The method depends entirely on the oscillations of numerous chemical substances and time-indicating metabolites; therefore, it can be applied to other metabolomics technologies such as CE-MS analysis. With CE-MS analysis, it is possible to separate charged compounds. So this is a complementary technology to LC-MS analysis (35). To demonstrate the general applicability of the metabolite-timetable method to other metabolomics technology, blood plasma was sampled and pooled from young male CBA/N mice every 4 hours under LD or DD conditions over 2 days. Small positively charged chemical substances in these samples were quantified by CE-MS, which detected 953 peaks. Of these peaks, 153 exhibited significant circadian oscillations under LD and DD conditions (Fig. S3A). We used these CE-MS data to construct the metabolite timetable in mouse plasma (Table S3). Notably, 28 peaks (18.3% of the total) were identified as known metabolites (Fig. S3B).

To confirm whether the CE-MS-based metabolite timetable was a good indicator of an individual's BT, we estimated BT from the metabolite profiles of independently sampled mice. Fresh plasma from young male CBA/N mice was collected every 4 hours over 24 h both under LD and DD conditions. CE-MS analysis was performed to profile the time-indicating metabolites (Fig. S3 C and D). The CE-MS-based metabolite-timetable method detected significant circadian rhythmicity in all metabolite profiles of these samples ($P < 0.01$, Fig. S3 C and D). The estimated BT was close to the environmental time with estimation errors of 0.6 ± 0.29 h for LD and 0.6 ± 0.54 h for DD (mean \pm SD, Table S4). These results suggest the metabolite-timetable method is generally applicable to other metabolomics technologies such as CE-MS. See Fig. S3 E–H and *SI Text* for the result at a more stringent criterion (FDR < 0.01).

Discussion

We identified 14 and 28 oscillatory peaks in mouse blood as known metabolites in LC-MS and CE-MS analysis, respectively. For example, a different type of lysophosphatidylcholines exhibit significant circadian oscillations in LC-MS analysis (Fig. 1 C and D). The genes for key enzymes synthesizing lysophosphatidylcholine are *Lcat* (lecithin:cholesterol acyltransferase), *Lipc* (hepatic lipase), and *Lipg* (endothelial lipase). Among these genes, *Lcat* and *Lipc* are mainly expressed in liver. We found that *Lipc* mRNA are rhythmically expressed in the mouse liver with the peak time around peak time (PT) 5 (8), which slightly proceeds with the peak time of identified oscillatory lysophosphatidylcholines (PT5.8–9.1). In CE-MS analysis, many amino acids exhibit significant circadian rhythmicity. For example, glutamine (Gln), threonine (Thr), proline (Pro), valine (Val), phenylalanine (Phe), methionine (Met), isoleucine (Ile), leucine (Leu), and tryptophan (Trp) peak at around midnight (\approx PT18), whereas glycine (Gly) peaks in the evening (PT12.1) (Fig. S4). In the urea cycle, metabolites such as ornithine (PT18.6), citrulline (PT19.9), and 4-guanidino-butyrate (PT20.1) exhibit significant circadian rhythmicity (FDR < 0.1). Arginine (Arg), which plays an important role in the urea cycle, also exhibits suggestive circadian rhythmicity (FDR = 0.215; PT0.6) in our CE-MS data. It is also noteworthy that the final product—Urea—is reported to vary over 24 h in the blood of certain species such as rabbits (41) and rats (42). Interestingly, Reddy et al. (43) showed that 3 key enzymes involved in the urea cycle, carbamoyl-phosphate synthetase 1

(CPS1), argininosuccinate synthetase 1 (ASS1), and arginase 1 (ARG1), show circadian rhythms in the liver, the center of the urea cycle and urea formation (43). In the creatine pathway and neighboring glycine and threonine metabolism, metabolites such as guanidoacetate (PT6.2), Creatine (PT14.7), creatinine (PT18.7), sarcosine (PT18.0), and dimethylglycine (PT16.5) exhibit significant circadian rhythmicity (FDR < 0.1). Arginine (PT0.6) first converts to guanidoacetate. Guanidoacetate (CT6.2) then converts to creatine. Creatine (PT14.7) finally converts to creatinine (PT18.7) or sarcosine (PT18.0), which is also converted from dimethylglycine (PT16.5). The differences in the peak times of these metabolites may reflect successive processing throughout the day in the creatine pathway and neighboring glycine and threonine metabolism (Fig. S4).

Our results suggest that metabolite-timetable method can detect circadian rhythm disorders in vivo. In a normal situation, patients live under the zeitgeber (e.g., light). Notably, our method successfully diagnoses the jet-lag state under LD conditions (Fig. 5 and Fig. S2), and this strongly suggests that endogenous abnormal clock state can be diagnosed by our method, even if there is external time information such as light. Circadian rhythm disorders are caused by environmental factors (such as jet lag) and/or inherited factors (as in familial advanced sleep-phase syndrome). Brown et al. (44, 45) reported detecting circadian rhythm disorders by characterizing the feature of molecular circadian clocks in the isolated cells. They collected skin samples from human subjects, cultured the cells, and transfected clock-controlled reporter into the cells. The features of the molecular circadian clock in the isolated cells correlated with the chronotypes (i.e., the feature of organismal circadian clock) of the subjects, suggesting that the method should also allow detection of inherited circadian rhythm disorders. Our method can detect both inherited and acquired circadian rhythm disorders but cannot distinguish between them, whereas Brown et al.'s (44, 45) method can detect inherited but not acquired disorders. These 2 methods are therefore complementary for detecting circadian rhythm disorders.

Although our results suggest the metabolite-timetable method can successfully estimate BT, keeping MS and hiring a specialized operator in each hospital seems difficult. Establishing a special center for "detecting BT" performing MS analysis is 1 possibility to solve this. Another possibility is detecting time-indicating metabolites in a specific way (e.g., making an ELISA kit for detecting BT using a specific antibody for target time-indicating metabolites). To achieve the latter possibility, the assignment of oscillatory peaks to known metabolites is important, and we already identified 14 (LC-MS) and 28 (CE-MS) oscillatory peaks as known metabolites (Figs. 1 and S3). We also examined the effect of peak numbers on BT estimation. Fig. S5 shows the accuracy of the BT estimation using the different number of oscillatory metabolites. If we set the statistical error rate $P < 0.05$ and estimation error between environmental and estimated time < 2 h, the minimum number of time-indicating metabolites was ≈ 20 . In addition, the effect of feeding is an important issue, especially in humans, because humans eat different amounts of food at entirely different times. Further analysis on food intake conditions would be a great help for applying this method in clinical situations.

In this study, we showed that a metabolite-timetable method based on LC-MS analysis is able to estimate individuals' BTs with a high degree of accuracy throughout the time of the day, under different lighting conditions (LD and DD), and in individuals with different genetic backgrounds (CBA/N and C57BL/6 mice) (Figs. 2 and 3). We also found that the LC-MS-derived metabolite timetable is robust despite differences in age, sex, and feeding (Fig. 4); in addition, it is a sensitive and accurate detector of disordered circadian rhythm in jet-lagged mice. Our preliminary results suggest that the metabolite-timetable method can be also applied to other metabolomics techniques such as CE-MS; it allowed quantification of hundreds of clock-controlled metabolites, of which many could

be identified, enabling successful measurement of BT from independent blood samples. The next step is to construct a metabolite timetable for human blood plasma, which will help measurement of BTs for humans and diagnosis of circadian rhythm disorders and facilitate the development of chronotherapy and tailored medication regimens.

Materials and Methods

BT Measurement. Metabolomics-based measurement of BT is performed as described for expression-based measurement of BT (16), except that 2 samples are used for an estimation of BT. In the metabolite-timetable method, we used 2 samples with 12-h sampling time interval (e.g., ZT0 and ZT12 are used for 1 measurement of BT) to calibrate measurement-to-measurement experimental fluctuations of detection sensitivity, which usually differs among metabolites. We define the area in a certain sample as A_{ij} and the mean areas of 2 samples (of 12-h time interval) as M_{ij} for metabolite i . We also define the mean, standard deviation, and peak time of metabolite i in the timetable as M_{ti} , S_{ti} , and p_{ti} , respectively. For estimation of BT, we did not use outlying metabolites that do not satisfy the condition $(|M_{ti} - M_{ij}|/S_{ti}) < 2\sqrt{2}$. By changing b to 0, 0.1, ..., 23.9, we searched for b with a maximum Pearson's correlation between $\{\sqrt{2} \cos(2\pi(P_{ij} - b)/24)\}$ and

$\{(A_{ij} - M_{ij})/S_{ti}\}$, and we predicted b as BT of the target sample. To estimate the P value of the prediction, we applied a permutation test to the maximum correlations.

Ethics. All experiments were performed with the permission of Kobe Animal Experiment Supervisory Panel (permission IDs are AH15-10 and AH18-01).

Supporting Information. More Materials and Methods information is available in *SI Text*.

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Quantitative Metabolome Profiling of Colon and Stomach Cancer Microenvironment by Capillary Electrophoresis Time-of-Flight Mass Spectrometry

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Abstract

Most cancer cells predominantly produce energy by glycolysis rather than oxidative phosphorylation via the tricarboxylic acid (TCA) cycle, even in the presence of an adequate oxygen supply (Warburg effect). However, little has been reported regarding the direct measurements of global metabolites in clinical tumor tissues. Here, we applied capillary electrophoresis time-of-flight mass spectrometry, which enables comprehensive and quantitative analysis of charged metabolites, to simultaneously measure their levels in tumor and grossly normal tissues obtained from 16 colon and 12 stomach cancer patients. Quantification of 94 metabolites in colon and 95 metabolites in stomach involved in glycolysis, the pentose phosphate pathway, the TCA and urea cycles, and amino acid and nucleotide metabolisms resulted in the identification of several cancer-specific metabolic traits. Extremely low glucose and high lactate and glycolytic intermediate concentrations were found in both colon and stomach tumor tissues, which indicated enhanced glycolysis and thus confirmed the Warburg effect. Significant accumulation of all amino acids except glutamine in the tumors implied autophagic degradation of proteins and active glutamine breakdown for energy production, i.e., glutaminolysis. In addition, significant organ-specific differences were found in the levels of TCA cycle intermediates, which reflected the dependency of each tissue on aerobic respiration according to oxygen availability. The results uncovered unexpectedly poor nutritional conditions in the actual tumor microenvironment and showed that capillary electrophoresis coupled to mass spectrometry-based metabolomics, which is capable of quantifying the levels of energy metabolites in tissues, could be a powerful tool for the development of novel anticancer agents that target cancer-specific metabolism. [Cancer Res 2009;69(11):4918–25]

Introduction

Most cancer cells are exposed to chronic hypoxia from the early stage of carcinogenesis. Indeed, the measurement of oxygen tension in tumors confirms severe hypoxia in many types of cancer (1). However, cancer cells' predominant use of glycolysis

rather than oxidative phosphorylation for energy production, irrespective of oxygen availability (Warburg effect; ref. 2), is widely acknowledged. This indicates that tumor hypoxia is caused not by the excessive oxygen consumption of cancer cells, but rather the inadequate blood supply that results from structurally and functionally defective angiogenesis. In addition, intrinsic characteristics of cancer cells and their constitutive expression of hypoxia-inducible transcription factors activate the genes that encode glycolytic enzymes and glucose transporters (3, 4) and, therefore, jointly hyperactivate glycolysis, to replenish ATP for their continuous growth and proliferation. Nevertheless, the cancer cells' intense use of energy-inefficient glycolysis in the hypovascular microenvironment may deplete glucose from the surrounding tissues. In this manner, the nutritional conditions of the tumor microenvironment may be extremely unfavorable from the perspective of energy metabolism, and significantly different from those that we generally expect from the observation of overgrowing cancer cells.

Although little is known concerning the actual concentrations of glucose and resultant metabolic intermediates in human cancer tissues, the recent development of metabolomics technologies, which are typically based on gas chromatography mass spectrometry (GC-MS; ref. 5), liquid chromatography mass spectrometry (LC-MS; ref. 6), and nuclear magnetic resonance (NMR; ref. 7) is suitable for the large-scale measurement of metabolite levels in tumor and normal tissues. This provides not only direct information on energy metabolism but also the potential reciprocal relationship between metabolic networks and the underlying mechanisms of carcinogenesis.

Recently, metabolome analysis has been applied to the characterization of cancer-cell-specific metabolism. Yang and colleagues (8) applied a computational flux analysis to compare breast cancer and normal human mammary epithelial cell lines by using two-dimensional NMR and GC-MS. Their finding of significant increases in the glycine and proline biosynthesis in cancer cells is interesting, yet may be limited for *in vitro* environment due to its high dependency on culture conditions. Chan and colleagues (9) compared the metabolic profile of biopsied colorectal tumors and their matched normal mucosae obtained from 31 colorectal cancer patients using high-resolution magic angle spinning-NMR and GC-MS and obtained 31 marker metabolites that distinguish normal from malignant samples and further colon from rectal cancers. Moreover, applying GC-MS-based metabolomics, Denkert and colleagues (10) compared the metabolic profiles between invasive ovarian carcinomas and the borderline tumors and between colorectal tumor and pairwise normal tissues (11) and showed that differentially expressed metabolic phenotypes could be exploited

Note: Supplementary data for this article are available at Cancer Research Online (<http://cancerres.aacrjournals.org/>).

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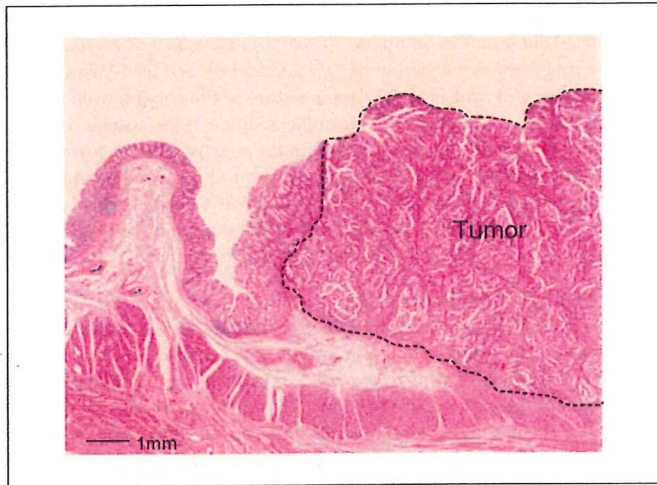


Figure 1. Representative microscopic image of an organ excised from a patient with well-differentiated colorectal adenocarcinoma. Samples were collected from the tumor region (surrounded by a dotted line) and nontumor region (considered normal).

to distinguish tumors from the others with high accuracies. However, little has been reported on the quantitation of metabolic intermediates involved in global-scale energy metabolism, including glycolysis, pentose phosphate pathway, and tricarboxylic acid (TCA) cycle, in human cancer and normal tissues. This is mainly due to the lack of effective methodology that allows comprehensive analysis of these metabolites. Most compounds involved in energy metabolism display common properties characterized by high polarity, nonvolatility, and poor detectability, which thus complicates the analysis. We recently developed a state-of-the-art metabolome analysis tool based on capillary electrophoresis coupled to mass spectrometry (CE-MS; refs. 12, 13). The major advantages of CE-MS analysis include its extremely high resolution, versatility to analyze metabolic profiles of various organisms, and ability to simultaneously quantify virtually all the charged low-molecular weight compounds in a sample (12, 14), which makes CE-MS best suited for the comprehensive analysis of energy metabolism in cells, tissues and biological fluids.

In the present study, we applied capillary electrophoresis time-of-flight mass spectrometry (CE-TOFMS; ref. 15) to the metabolome

profiling of human colon and stomach cancers, and compared the metabolite levels in tumor and normal tissues obtained by surgery. The results clearly showed that the tumor microenvironment is far from ideal for cell growth from the viewpoint of energy metabolism, and showed the versatility of CE-MS-based metabolomics for global-scale analysis of energy metabolism in tissues.

Materials and Methods

Sample Collection and Metabolite Extraction

We conducted all the experiments according to the study protocol approved by the Institution Review Board of the National Cancer Center upon obtaining informed consent from all the subjects. Tumor and surrounding grossly normal-appearing tissues (Fig. 1) were obtained from 16 colon and 12 stomach cancer patients after surgical treatment. Patient and tumor stage information are listed in Table 1. The excised tissues were cut into <1-cm³ pieces, immediately frozen in liquid nitrogen, and stored at -80°C until metabolite extraction.

To extract metabolites, preweighed deep-frozen samples (~50 mg each) were completely homogenized by a cell disrupter (MS-100R; TOMY) at 2°C, after adding 625 µL of methanol that contained internal standards [20 µmol/L each of methionine sulfone and 2-(N-morpholino)-ethanesulfonic acid]. The homogenate was then mixed with Milli-Q water and chloroform in a volume ratio of 5:2:5 and centrifuged at 9,000 g for 15 min at 4°C. Subsequently, the aqueous solution was centrifugally filtered through a 5-kDa cutoff filter (Millipore) to remove proteins. The filtrate was centrifugally concentrated and dissolved in 50 µL Milli-Q water that contained reference compounds (200 µmol/L each of 3-aminopyrrolidine and trimesate) immediately before CE-TOFMS analysis.

Reagents

Ophthalmate was purchased from BACHEM AG; glycerol-3-phosphate from Nakalai Tesque; sedoheptulose 7-phosphate from GlycoTeam; tyramine, CoA, and NADH from MP Biomedicals; and fructose 1,6-bisphosphate, glucose 1-phosphate, ribose 5-phosphate, and ribulose 5-phosphate from Fluka. γ-Glu-Cys and γ-Glu-2-aminobutyrate were synthesized at the Toray Research Center. All other reagents were obtained from either Wako or Sigma-Aldrich. Stock solutions (1–100 mmol/L) were prepared in either Milli-Q water, 0.1 mol/L HCl, or 0.1 mol/L NaOH. All chemical standards were analytic or reagent grade. A mixed solution of the standards was prepared by diluting stock solutions with Milli-Q water immediately before CE-TOFMS analysis.

Analytic Condition for Metabolome Analysis

Instruments. All CE-TOFMS experiments were performed using an Agilent CE Capillary Electrophoresis System equipped with an Agilent TOFMS, an Agilent 1100 isocratic HPLC pump, an Agilent G1603A CE-MS adapter kit, and

Table 1. Patient information and tumor stages

	Characteristic	Colon		Stomach	
		Number	%	Number	%
Position of colon tumor	Rectum	10	63		
	Ascending colon	3	19		
	Transverse colon	1	6		
	Sigmoid colon	2	13		
Stage	I	5	31	2	17
	II	4	25	3	25
	III	6	38	5	42
	IV	1	6	2	17
Sex	Male	11	69	7	58
	Female	5	31	5	42