

Fig. 1. Transfection efficiency after intravitreal injection of naked pDNA (a), pDNA/DOTMA/DOPE liposomes (b) or pDNA/DOTMA/Chol liposomes (c) at pDNA doses of 40, 60, and 85  $\mu\text{g}$  in rabbits. pDNA was complexed with each cationic liposome at a charge ratio (-:+) of 1.0:2.0. Luciferase activities were measured 3 days post-injection in the cornea ( $\square$ ), aqueous humor ( $\blacksquare$ ), iris-ciliary body ( $\square$ ), lens ( $\blacksquare$ ), vitreous body ( $\blacksquare$ ), and retina ( $\blacksquare$ ), respectively. Each bar represents the mean  $\pm$  S.D. of three or four experiments. \* Significant differences ( $P < 0.05$ ) at pDNA (85  $\mu\text{g}$ ).

in the ocular tissues at 0.25 day (6h). The gene expression in ocular tissues was transient, peaking at 3 days, and dropping to less than 10% of the peak level at 7 days. Transfection efficacy in the cornea and vitreous body at 3 days was significantly higher than that at 7 days ( $P < 0.05$ ).

### 3.2. Effect of the charge ratio of pDNA/cationic liposome complexes on gene expression

Fig. 3 shows the effect of the charge ratio on transfection efficiency 3 days after intravitreal injection of pDNA (85  $\mu\text{g}$ ) complexed with DOTMA/Chol

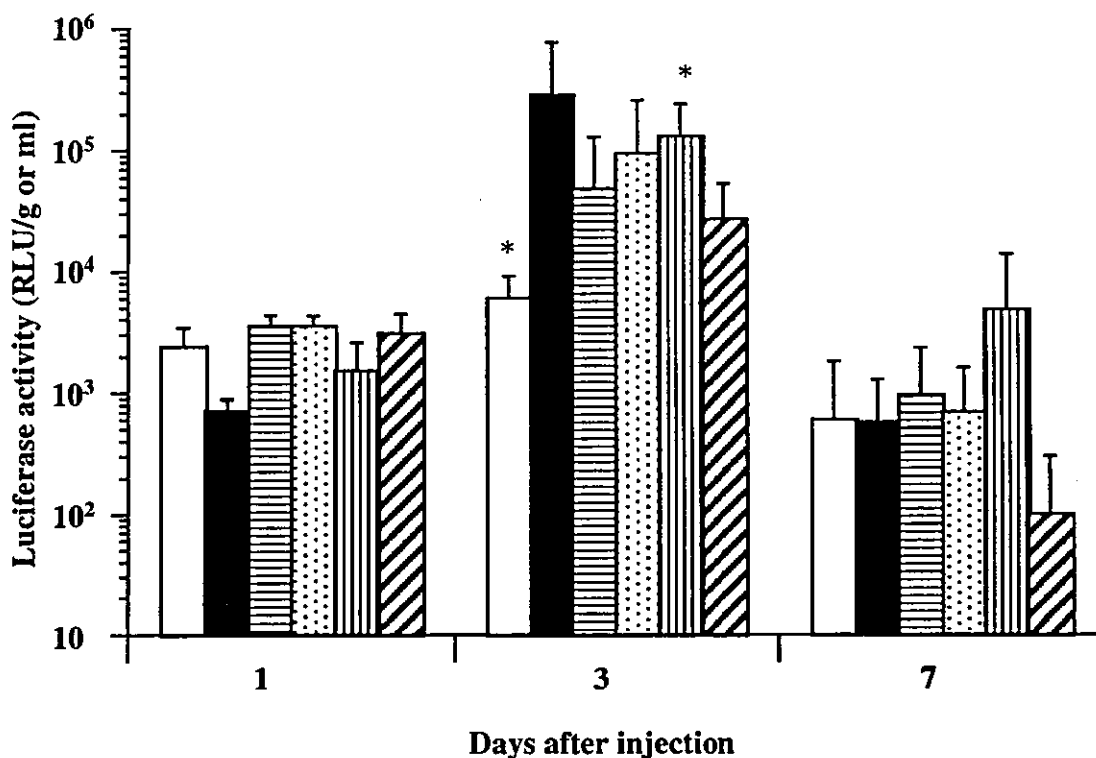


Fig. 2. Time courses of transfection efficiency after intravitreal injection of pDNA complexed with DOTMA/Chol liposomes at a pDNA dose of 85 mg in rabbits. Luciferase activities were measured 1, 3, and 7 days post-injection in the cornea (□), aqueous humor (■), iris-ciliary body (▨), lens (▩), vitreous body (▧), and retina (▧), respectively. Each bar represents the mean  $\pm$  S.D. of three or four experiments. \* Significant differences ( $P < 0.05$ ) at the value of 7 days.

liposomes in rabbits. pDNA (85  $\mu$ g) was complexed with various amounts of DOTMA/Chol liposomes at the charge ratio (–:+) of 1.0:1.5, 1.0:2.0, or 1.0:2.5. Among these charge ratios, with the charge ratio (–:+) of 1.0:2.0 resulted in the highest transfection efficiency.

#### 4. Discussion

DOPE is often used as the neutral lipid for in vivo ocular gene transfection using cationic liposomes (Matsuo et al., 1996; Masuda et al., 1996; Urtti et al., 2000; Abul-Hassan et al., 2000; Pleyer et al., 2001) because of its pH-sensitive ability to destabilize the lysosomal membranes following endocytosis. Although Chol does not have such a pH-sensitive property, it is well known that liposomal membranes are stabilized by Chol in vivo (Semple et al., 1996;

Murao et al., 2002). Recently, Li et al. (1999) reported that DOPE-containing liposome/pDNA complexes were more markedly aggregated by serum exposure than Chol-containing liposome/pDNA complex. It is possible that this ability to stabilize liposomal membranes might also result in stabilization of the pDNA in the vitreous body, resulting in enhancement of the transfection efficiency in the eye. As shown in Fig. 1, pDNA (85 mg) in DOTMA/Chol liposome complexes showed higher transfection efficiency than pDNA in other liposome complexes. These results suggest the potential usefulness of pDNA complexed with DOTMA/Chol liposomes for ocular gene transfection after intravitreal injection. This observation is consistent with previously reported results about systemic administration of pDNA/cationic liposome complexes (Huang and Li, 1997; Sakurai et al., 2001).

In the present study, we found that pDNA complexed with DOTMA/Chol liposomes at a charge ratio

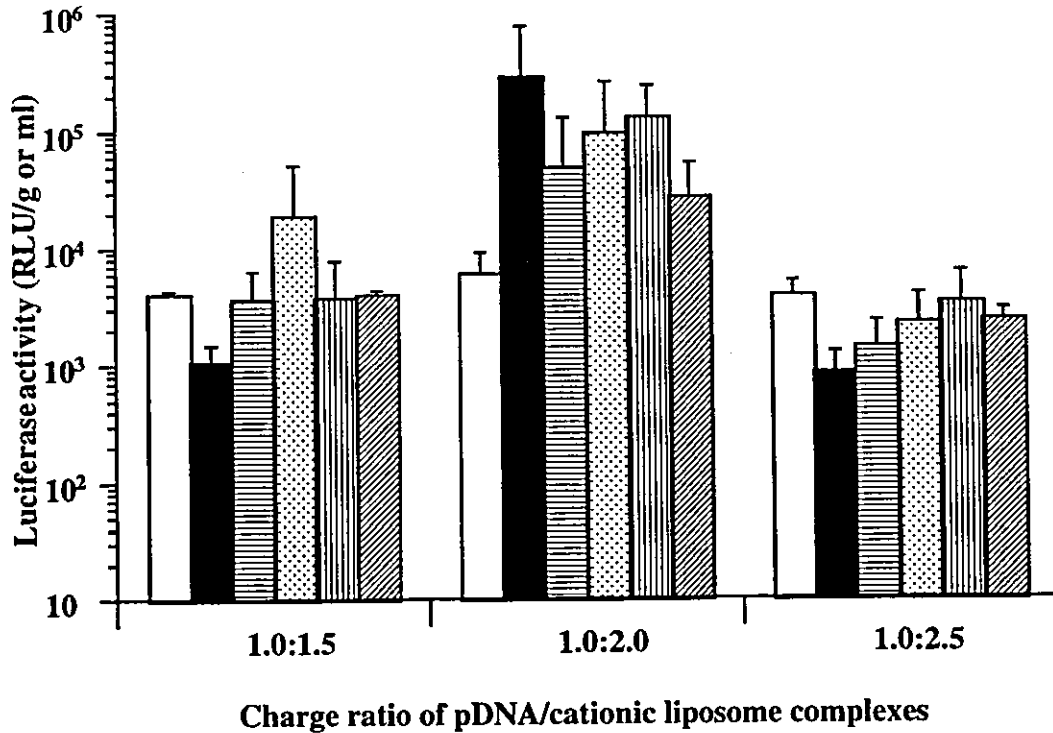


Fig. 3. Effect of charge ratio of pDNA complexed with DOTMA/Chol liposomes on transfection efficiency after intravitreal injection of pDNA/DOTMA/Chol liposomes complex at a pDNA dose of 85 mg in rabbits. pDNA was complexed with cationic liposomes at a charge ratio (-:+) of 1.0:1.5, 1.0:2.0, and 1.0:2.5, respectively. Luciferase activities were measured 3 days post-injection in the cornea (□), aqueous humor (■), iris-ciliary body (▨), lens (▩), vitreous body (▧), and retina (▦), respectively. Each bar represents the mean  $\pm$  S.D. of three or four experiments.

(-:+) of 1.0:2.0 produced maximal gene expression compared with complexes with other charge ratios (Fig. 3), indicating that the charge of complexes is an important factor with respect to the *in vivo* ocular transfection efficiency. Cationic liposomes can interact with various types of negatively charged biological components such as serum proteins, probably due to their positive charge after administration into the systemic circulation (Liu et al., 1997). Yang and Huang (1997) reported that the highest transfection efficiency was observed when pDNA was complexed with cationic liposomes at a pDNA-cationic liposome charge ratio (-:+) of 1.0:2.0 in the absence of protein and more than 1.0:4.0 in the presence of protein. Following the intravenous administration of liposome-DNA complexes into mice at a charge ratio of more than 1.0:4.0, high-level, reproducible transgene expression was brought about in various tissues. The discrepancy between the results obtained with

intravenous administration and intravitreal administration may be partly due to the fact that the protein content in the vitreous body (about 0.1%) (Ueno et al., 1991) is much lower than that in the blood (about 10%); thus, the characteristics of the transfection efficiency after intravitreal injection might be compatible with those in the absence of protein.

The selection of administration route could be an important issue regarding the efficiency of the transfection of genes into ocular tissues. Thus, various routes of administration such as instillation onto the eye, injection into the anterior chamber, intravitreal injection, and subretinal injection were studied using the pDNA/cationic liposome complexes. Generally, topical administration on the eye is expected to be a convenient and safe method for ophthalmic therapy. It was reported that expression of a transfected gene after topical administration on the eye was found in the corneal epithelium, conjunctival

epithelium, ciliary epithelium, iris stroma, and retinal ganglion cells using pDNA complexed with *N*-( $\alpha$ -trimethylammonioacetyl)-didodecyl-D-glutamate (TMAG)/DOPE/dilauroylphosphatidylcholine (DLPC) liposomes or  $3\beta$ [*N,N'*-dimethylaminoethane)-carbamoyl]cholesterol (DC-Chol)/DOPE liposomes (Matsuo et al., 1996; Masuda et al., 1996). However, our preliminary experiments showed that little luciferase activity was detected even in cornea epithelium after topical administration of 25 mL of pDNA (10 mg) complexed with DOTMA/DOPE or DOTMA/Chol liposomes on to the eye. The cornea, considered to be a major pathway for ocular penetration of topically applied drugs to the eye, is an effective barrier to drug penetration (Sasaki et al., 1996), because the corneal epithelium has annular tight junctions (zonula occludens) that completely surround and effectively seal the superficial epithelial cells. Also, liposomes (Kawakami et al., 2001) like most drugs, are rapidly eliminated from the precorneal area by tear fluid after topical administration to the eye (Yamamura et al., 1999). These features might limit the transgene expression in the eyes following the topical administration of pDNA complexed with DOTMA/DOPE and/or DOTMA/Chol liposomes.

The gene expression in the ocular tissues was transient, with a peak level at 3 days (Fig. 2), even though the pDNA/cationic liposomes were stabilized by the application of DOTMA/Chol liposomes; therefore, more prolonged transgene expression is a current goal for clinical gene therapy because repeated intravitreal administration is painful for the patients. This observation leads us to believe that further studies are needed on the issue of prolonging the gene expression in the eyes.

In conclusion, superior gene transfection in the ocular tissues was observed with the intravitreal administration of pDNA/DOTMA/Chol liposomes, which were an effective gene carrier when delivered by intravenous administration. The use of appropriate neutral lipids and the optimal charge ratio (–:+) of the pDNA/cationic liposome complexes will improve the efficiency of *in vivo* ocular gene transfection. Our observations based on the physicochemical properties of pDNA/cationic liposome complex can be generalized, so this information will be valuable for designing gene transfection systems for the eye.

## Acknowledgements

This work was supported in part by a grant-in-aid for scientific research from the Ministry of Education, Culture, Sports, Science, and Technology, Japan, and by a grant-in-aid from the Uehara Memorial Foundation.

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## Prednisolone retention in integrated liposomes by chemical approach and pharmaceutical approach

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Received 25 August 2003; accepted 1 March 2004

### Abstract

The purpose of this study is to demonstrate a stable retention of prednisolone (PLS) in the unique liposomes integrated by lipophilic derivative approach and PEGylation approach. Palmitoyl prednisolone (Pal-PLS) was newly synthesized and used as a lipophilic derivative. The liposomes were composed of egg phosphatidylcholine (EggPC)/cholesterol (Chol) and L- $\alpha$ -distearoylphosphatidylcholine (DSPC)/Chol with or without L- $\alpha$ -distearoylphosphatidylethanolamine-polyethylene glycol 2000 (DSPE-PEG 2000) or -PEG 5000 (DSPE-PEG 5000). The retentions of PLS and Pal-PLS in the various liposomes were examined by ultrafiltration and gel filtration. Although PLS showed high trapping efficiency by all liposomes after ultrafiltration, low incorporation efficiency was observed in gel filtration. It indicates that PLS was released from the liposomes by a dilution with elution medium in gel filtration. Pal-PLS showed high incorporation into all liposomes after both ultrafiltration and gel filtration. The high incorporation of Pal-PLS into EggPC/Chol liposomes, however, was reduced by incubation with rat plasma in gel filtration. The reducing effect of rat plasma on drug incorporation into liposomes was inhibited by using DSPC and DSPE-PEGs. Thus, we systemically examined the drug retention in various liposomes and demonstrated the high retention of PLS in the liposomes integrated by lipophilic derivative approach and pharmaceutical approach using special lipids.

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**Keywords:** Drug delivery system; Prednisolone; Liposomes; Lipophilic derivative; Polyethylene glycol

### 1. Introduction

Glucocorticoids are highly potent antiinflammatory and immunosuppressive drugs. However, even at moderate doses, systematic administration of gluco-

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corticoids causes many side effects, such as diabetes, hypertension, Cushing syndrome, and osteoporosis [1]. It is reported that the therapeutic activity of glucocorticoids could be improved through application of liposomal drug carrier technology [2,3]. A major problem, however, exists on the utility of liposomes for delivering drugs because of a rapid removal of liposomes by the mononuclear phagocytic system uptake [4,5]. For improving the pharmacokinetics of liposomes, therefore, the surface-modified polyethylene glycol (PEG) liposomes containing synthetic polyethyleneoxide lipopolymers were introduced and a significant improvement in the liposomes half-lives in the blood could be achieved [6,7]. Even if PEG-modified liposomes exhibited a favorable in vivo disposition profiles, rapid release of incorporated drugs would lead to failure to achieve therapeutic potency. We previously reported that rapid release of incorporated drugs into liposomes was suppressed by chemical modification of drug to lipophilic derivative [8,9]. There has been, however, little information about the unique liposomes of prednisolone (PLS) integrated by chemical modification and pharmaceutical modification. The integrated liposomes of PLS would be useful for drug targeting because of its stable characteristics.

In the present study, we newly synthesized palmitoyl prednisolone (Pal-PLS) and prepared the liposomes with various lipids such as egg phosphatidylcholine (EggPC), 1- $\alpha$ -distearoylphosphatidylcholine (DSPC), cholesterol (Chol), 1- $\alpha$ -distearoylphosphatidylethanolamine-PEG 2000 (DSPE-PEG 2000) and -PEG 5000 (DSPE-PEG 5000). In addition, the drug retention in the various liposomes was examined by ultrafiltration and gel filtration.

## 2. Materials and methods

### 2.1. General procedure

Melting points were determined with a Yanagimoto micromelting point apparatus and are uncorrected.  $^1\text{H-NMR}$  spectra were recorded on Varian Gemini 300 spectrometers with tetramethylsilane as an internal standard and chemical shifts are given in  $\delta$  (ppm). Elemental analyses were performed by the Center for Instrumental Analysis, Nagasaki

University. Thin-layer chromatography (TLC) was carried out on TLC aluminum sheets precoated with a 0.2-mm layer of Silica gel 60 F<sub>254</sub> (Merck, Darmstadt, Germany), using the following solvent system: chloroform–methanol–acetic acid (95:4:1, v/v/v).

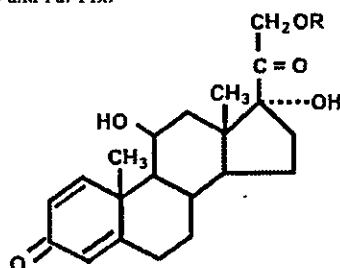
### 2.2. Materials

PLS was kindly supplied from Shionogi (Osaka, Japan). Palmitoyl chloride was obtained from Nacalai Tesque (Kyoto, Japan). DSPC (COATSOME MC-8080) and EggPC (average molecular weight 773, COATSOME NC-50) were purchased from Nippon Oil and Fats (Tokyo, Japan). DSPE-PEG was a gift from Nippon Oil and Fats. Average molecular weight of PEG chain in the DSPE-PEG was 2000 (SUNBRIGHT DSPE-20H) or 5000 (SUNBRIGHT DSPE-50H). Chol was obtained from Nacalai Tesque. Fluorescein isothiocyanate-dextran (average molecular weight 4400, FD-4) was purchased from Sigma (St. Louis, MO, USA). All other chemicals were of reagent grade and used as obtained commercially. Phosphate-buffered saline (PBS) was prepared by mixing an isotonic phosphate buffer (pH 7.4) with an equal volume of saline.

### 2.3. Synthesis of Pal-PLS

Palmitoyl chloride (1 ml, 3.3 mmol) was added to PLS (0.5 g, 1.4 mmol) in anhydrous pyridine (10 ml) and stirred at room temperature overnight. The reaction was monitored by thin-layer chromatography. The reaction mixture was dissolved in 1 M HCl (100 ml) and extracted with diethyl ether (50 ml) five times. The organic layer collected was dried, filtered, and evaporated in vacuo to give Pal-PLS as white solid crystals in 70% yield. The structure of synthesized derivative was confirmed by both elemental analysis and  $^1\text{H-NMR}$ . *Analysis*: Calculated for  $\text{C}_{37}\text{H}_{58}\text{O}_6 \cdot 1/4 \text{H}_2\text{O}$ : C (73.65), H (9.77), O (16.58). Found: C (73.62), H (9.84), O (16.54).  $^1\text{H-NMR}$  ( $\text{CDCl}_3$ )  $\delta$ : 0.72–0.98 (H, Pal- $\text{CH}_3$ ), 0.98 (3H, s,  $\text{C}_{13}$ - $\text{CH}_3$ ), 1.09–2.85 (18H, protons belong to B, C, and D rings), 1.36 [26H, s,  $-(\text{CH}_2)_{14}-$ ], 2.30–2.50 (2H, m,  $\text{C}_{23}$ - $\text{H}_2$ ), 4.52–4.46 (1H,  $\text{C}_{11}$ -OH), 4.93 (2H, AB-q,  $\text{C}_{21}$ - $\text{H}_2$ ), 6.01 (1H,  $\text{C}_4$ -H), 6.27 (1H, d,  $J=10$  Hz,  $\text{C}_2$ -H), 7.28 (1H, d,  $J=10$  Hz,  $\text{C}_1$ -H).

Table 1  
Structures and physicochemical properties of PLS and Pal-PLS



Compound	R	Molecular weight	Melting point (°C)	Lipophilic index ( $\log k'_0$ )	Relative lipophilicity
PLS	-H	360.5	235	3.2	1
Pal-PLS	-CO(CH <sub>2</sub> ) <sub>14</sub> CH <sub>3</sub>	598.9	52–56	10.3	$1.4 \times 10^7$

\* Lipophilic index was measured by reversed-phase HPLC systems.

#### 2.4. Lipophilic index

The relative lipophilic indices ( $\log k'$ ) were determined by high-performance liquid chromatography (HPLC) employing Eq. (1) [10];

$$\log k' = \log[(t_R - t_0)/t_0] \quad (1)$$

where  $t_R$  is the retention time of the solute and  $t_0$  is the elution time of the solvent. The lipophilic indices ( $\log k'_0$ ) were determined by extrapolation of the  $\log k'$  values to 0% organic solvent concentration. A mixture of 2-propanol and distilled water was used as the mobile phase at a flow rate of 1.0 ml/min.

#### 2.5. Preparation of liposomes

The liposomes were prepared by sonication method [11]. Both EggPC/Chol and DSPC/Chol liposomes contained phospholipid, Chol, and drug at a molar ratio of 3:2:0.5. In some cases, 1 mol% of DSPE-PEG 2000 or DSPE-PEG 5000 was also added to the lipid mixture. After each mixture of lipids in chloroform was placed in a round-bottom glass tube, chloroform was evaporated. The lipid film obtained was further dried in vacuo in a desiccator for 4 h. The lipid suspension was vortex-mixed followed by ultrasonic radiation for 3 min under nitrogen gas. The liposomes containing DSPC and EggPC were prepared at 60 and 0 °C, respectively. PBS was added to the resulted liposomes, then the lipid was allowed to hydrate for 24 h at

5 °C. FD-4 was selected as a tracer of liposomal fraction in gel filtration. Liposomal formulation was prepared at 20 mmol/l lipid in PBS. All lipid compositions are given as molar ratios unless otherwise indicated.

Table 2  
Trapping efficiency of PLS and Pal-PLS by various liposomes in ultrafiltration

Entry	Drug	Lipid composition (molar ratio)	Trapping efficiency (%)
1	PLS	EggPC:Chol (3:2)	97.3 ± 0.4
2	PLS	EggPC:Chol:DSPE-PEG 2000 (3:2:0.05)	98.6 ± 1.2
3	PLS	EggPC:Chol:DSPE-PEG 5000 (3:2:0.05)	96.3 ± 3.3
4	PLS	DSPC:Chol (3:2)	94.5 ± 2.0
5	PLS	DSPC:Chol:DSPE-PEG 2000 (3:2:0.05)	90.5 ± 1.3
6	PLS	DSPC:Chol:DSPE-PEG 5000 (3:2:0.05)	80.0 ± 3.8
7	Pal-PLS	EggPC:Chol (3:2)	99.8 ± 0.2
8	Pal-PLS	EggPC:Chol:DSPE-PEG 2000 (3:2:0.05)	99.8 ± 0.2
9	Pal-PLS	EggPC:Chol:DSPE-PEG 5000 (3:2:0.05)	99.9 ± 0.2
10	Pal-PLS	DSPC:Chol (3:2)	99.9 ± 0.2
11	Pal-PLS	DSPC:Chol:DSPE-PEG 2000 (3:2:0.05)	99.9 ± 0.1
12	Pal-PLS	DSPC:Chol:DSPE-PEG 5000 (3:2:0.05)	99.8 ± 0.1

Each value represents the average ± S.D. of at least three experiments.



## 2.6. Drug determination

PLS and Pal-PLS were determined using HPLC system (LC-10AD, Shimadzu, Kyoto, Japan) in the reversed-phase mode. The stationary phase used was Cosmosil 5C<sub>18</sub>-MS-II packed column (150 × 4.6 mm for PLS and Pal-PLS, Nacalai Tesque). A mixture of 2-propanol, acetonitrile, and water (42:38:20, v/v/v) was used as a mobile phase with a flow rate of 1.0 ml/min for assay of Pal-PLS. A mixture of methanol and ammonium acetate buffer (pH 4) (57.5:42.5, v/v) was used as a mobile phase with a flow rate of 0.55 ml/min for assay of PLS. Retention of drugs was monitored with a variable wavelength ultraviolet detector (wavelength at 240 nm, SPD-10A, Shimadzu).

## 2.7. Ultrafiltration

Ultrafiltration was performed using Ultrafree<sup>®</sup>-MC centrifugal filter units (Millipore, Bedford, MA,

USA). The drug concentrations in the filtrate and in the applied liposomes were assayed by HPLC. Trapping efficiency was calculated from Eq. (2) [12]:

$$\text{Trapping efficiency(\%)} = (D_T - D_F)/D_T \times 100 \quad (2)$$

where  $D_T$  is the drug amounts in applied liposomes and  $D_F$  is the drug amounts in filtrate, respectively.

## 2.8. Gel filtration

The plasma was withdrawn from male Wistar rats (230–250 g) after anesthetization with a sodium pentobarbital solution. After liposomes were incubated for 1 min at 25 °C with or without rat fresh plasma (20%, v/v), they were applied to a Sephacryl S-400 column (16 × 2.8 cm) and eluted with PBS at 25 °C.

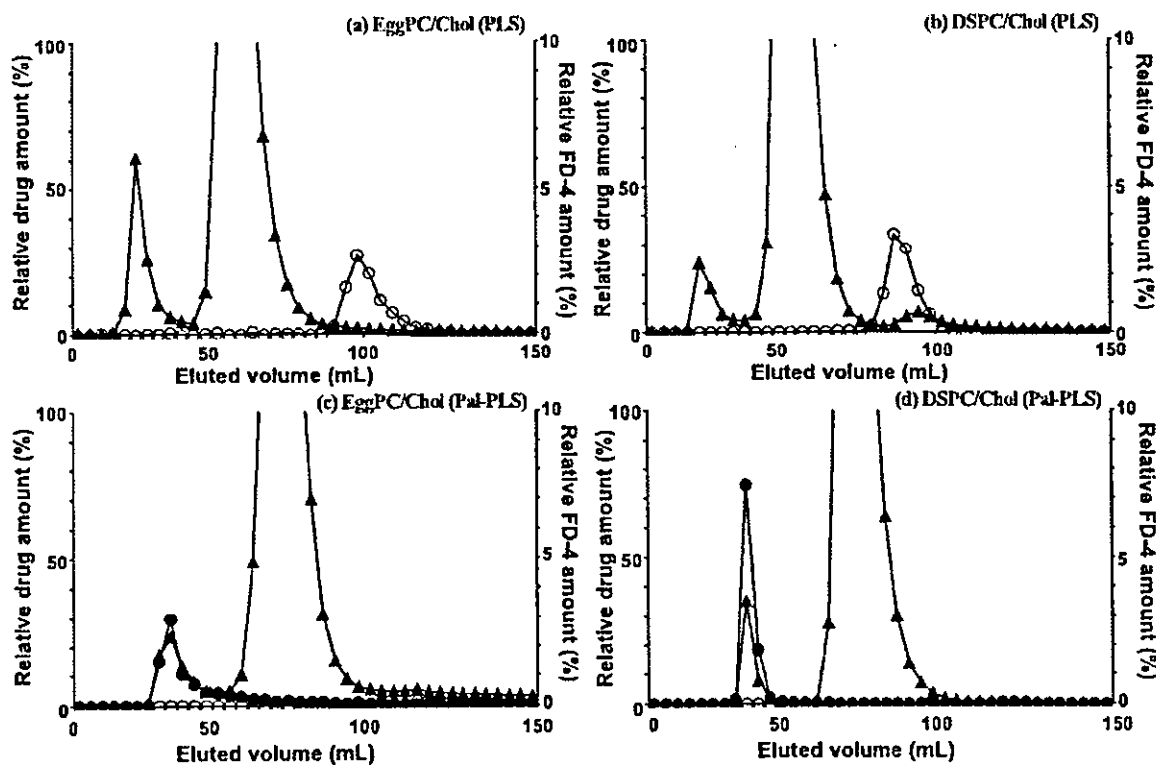


Fig. 1. Typical gel filtration profiles for EggPC/Chol (a and c) and DSPC/Chol (b and d) liposomes incorporating PLS (a and b) and Pal-PLS (c and d). Symbols: (▲) FD-4; (○) PLS; (●) Pal-PLS.

The incorporation efficiency was calculated from Eq. (3) [13];

$$\text{Incorporation efficiency}(\%) = D_e/D_t \times 100 \quad (3)$$

where  $D_e$  is the drug amounts recovered in the liposomal fractions and  $D_t$  is the drug amounts in applied liposomes to a gel column, respectively.

### 2.9. Statistical analysis

Statistical comparisons were performed by both analysis of variance and Tukey multiple comparison test.  $P < 0.05$  was considered to be indicative of statistical significance.

## 3. Results

### 3.1. Physicochemical properties of PLS and Pal-PLS

The structures and physicochemical properties of the compounds tested in the present investigation are summarized in Table 1. The lipophilicity of Pal-PLS

Table 3  
Incorporation efficiency of PLS and Pal-PLS into various liposomes in gel filtration

Entry	Drug	Lipid composition (molar ratio)	Incorporation efficiency (%)
1	PLS	EggPC:Chol (3:2)	0.1 ± 0.1
2	PLS	EggPC:Chol:DSPE-PEG 2000 (3:2:0.05)	1.2 ± 0.8
3	PLS	EggPC:Chol:DSPE-PEG 5000 (3:2:0.05)	2.4 ± 2.1
4	PLS	DSPC:Chol (3:2)	1.7 ± 1.0
5	PLS	DSPC:Chol:DSPE-PEG 2000 (3:2:0.05)	1.9 ± 1.5
6	PLS	DSPC:Chol:DSPE-PEG 5000 (3:2:0.05)	4.3 ± 3.0
7	Pal-PLS	EggPC:Chol (3:2)	71.3 ± 3.5
8	Pal-PLS	EggPC:Chol:DSPE-PEG 2000 (3:2:0.05)	82.7 ± 9.8
9	Pal-PLS	EggPC:Chol:DSPE-PEG 5000 (3:2:0.05)	88.9 ± 4.1
10	Pal-PLS	DSPC:Chol (3:2)	91.2 ± 7.1
11	Pal-PLS	DSPC:Chol:DSPE-PEG 2000 (3:2:0.05)	90.2 ± 1.8
12	Pal-PLS	DSPC:Chol:DSPE-PEG 5000 (3:2:0.05)	91.9 ± 3.8

Each value represents the average ± S.D. of at least three experiments.

Table 4

The effect of plasma on incorporation efficiency of Pal-PLS into various liposomes in gel filtration

Entry	Drug	Lipid composition (molar ratio)	Incorporation efficiency (%)
7	Pal-PLS	EggPC:Chol (3:2)	46.0 ± 5.5
8	Pal-PLS	EggPC:Chol:DSPE-PEG 2000 (3:2:0.05)	83.1 ± 3.6*
9	Pal-PLS	EggPC:Chol:DSPE-PEG 5000 (3:2:0.05)	83.0 ± 1.0*
10	Pal-PLS	DSPC:Chol (3:2)	88.4 ± 7.2*
11	Pal-PLS	DSPC:Chol:DSPE-PEG 2000 (3:2:0.05)	87.4 ± 6.8*
12	Pal-PLS	DSPC:Chol:DSPE-PEG 5000 (3:2:0.05)	88.0 ± 2.9*

Each value represents the average ± S.D. of at least three experiments.

\* Significantly different from Entry 7 ( $P < 0.05$ , Tukey multiple comparison test).

was confirmed by measuring  $\log k'_0$  in the HPLC system. Pal-PLS showed higher lipophilicity compared with PLS.

### 3.2. Ultrafiltration

The trapping efficiency of PLS and Pal-PLS by various liposomes was calculated after ultrafiltration and is shown in Table 2. More than 80% of not only PLS but also Pal-PLS were trapped by the liposomes consisting of various lipids. In the preliminary experiments, it was confirmed that there was little adhesion of drugs with ultrafiltrated membrane.

### 3.3. Gel filtration

The typical gel filtration profiles for EggPC/Chol and DSPC/Chol liposomes incorporating PLS and Pal-PLS are shown in Fig. 1. The first and second peaks of FD-4 are considered to be the incorporated drug into liposomes (liposomal fraction) and non-incorporated drug (free fraction), respectively. PLS was mostly existed in the free fraction, although Pal-PLS was mostly retained in liposomal fraction.

The incorporation efficiencies of PLS and Pal-PLS into various liposomes were calculated from gel filtration profiles and are summarized in Table 3. Although PLS showed high trapping efficiency by all liposomes after ultrafiltration, the low incorporation efficiency of PLS into all liposomes was observed

in the gel filtration. On the other hand, Pal-PLS showed high incorporation into all liposomes not only in ultrafiltration but also in gel filtration.

#### 3.4. Effect of plasma on Pal-PLS retention in liposomes

The effect of plasma on Pal-PLS retention in the liposomes was evaluated by gel filtration. After liposomes were incubated for 1 min with rat plasma, they were applied to a gel column and eluted with PBS. The incorporation efficiency of Pal-PLS was calculated from gel filtration profiles and is summarized in Table 4. The incorporation efficiency of Pal-PLS into EggPC/Chol liposomes reduced to approximately 65% after their incubation with plasma compared with the value without plasma. On the other hand, approximately 97% incorporation of Pal-PLS was observed in DSPC/Chol liposomes after their incubation with plasma. Addition of DSPE-PEG 2000 or DSPE-PEG 5000 to EggPC/Chol liposomes also showed high incorporation of Pal-PLS after their incubation with plasma.

#### 4. Discussion

Recently, drug delivery systems using liposomes as drug carriers have been well studied to achieve controlled and site-specific delivery of drugs [14–17]. Liposomes have various advantages as a drug carrier: biodegradability, low *in vivo* toxicity, and encapsulation of hydrophilic, lipophilic and amphipathic drugs. Introduction of thermosensitive lipid to liposomes [14] and modification of the liposomal membrane by galactose [15] have been reported to be useful for achievement of further drug targeting. In systemic circulation, however, drugs incorporated into liposomes were diluted by blood, followed by the interaction with plasma components [18–20]; consequently, incorporated drugs were rapidly released from liposomes [21].

In liposomes, lipophilic drugs were retained in the liposomal lipid bilayer, although water-soluble drugs were retained inside the aqueous phase [22]. We newly synthesized Pal-PLS as a lipophilic derivative of PLS to increase its affinity to lipid bilayer. The initial step in producing biological responses to ad-

ministered corticosteroids is the diffusion of unbound drug from plasma into cells for interaction with cytosolic receptors. Pal-PLS might be extremely less active than PLS because the chemically modified part of Pal-PLS seems to be important for its nuclear receptor binding. On the other hand, Pal-PLS was rapidly biodegraded to PLS in the presence of not only rat liver homogenate but also rat plasma in the preliminary experiments (data not shown). Therefore, it is considered that Pal-PLS released from liposomes must convert to PLS rapidly under biological condition and show therapeutic activity.

The retention property of PLS and Pal-PLS in various liposomes was investigated by ultrafiltration and gel filtration. PLS is lipophilic drug and its liposomes showed an increased therapeutic efficiency in IgA nephropathy model mice compared with PLS alone owing to enhancement of kidney delivery [2]. Actually, PLS was well incorporated into various liposomes as the results from ultrafiltration (Table 2). Ultrafiltration method could measure trapping efficiency of liposomes simply and immediately after their preparation. The incorporation of PLS into the liposomes, however, was dramatically decreased in the gel filtration (Table 3). The amounts of FD-4, hydrophilic marker, in liposomal fractions containing PLS had no significant difference from those in liposomal fractions containing Pal-PLS. This result indicated that PLS was released from liposomes by a dilution with elution medium without the collapse of liposomes in the process of gel filtration. As the results in the gel filtration, most of PLS must be released from any liposomes by dilution with blood circulation after their administration *in vivo*.

On the other hand, Pal-PLS showed highly incorporation efficiency not only in ultrafiltration but also in gel filtration (Table 3). It is demonstrated that the enhanced lipophilicity of drug itself is more important to improve the drug retention in liposomes compared with pharmaceutical approach using special lipids.

It is well known that liposomes were largely influenced by biological component under *in vivo* condition [23]. Therefore, the influence of rat plasma on retention of Pal-PLS in the liposomes was determined by gel filtration. As the results, rat plasma decreased incorporation efficiency of Pal-PLS into EggPC/Chol liposomes. DSPC/Chol liposomes

showed high incorporation efficiency of Pal-PLS even after their incubation with rat plasma (Table 4). DSPC and EggPC have been chosen as a typical lipid for preparing the liposomes that are commonly used as drug delivery [24,25]. Silvander et al. [24] reported that DSPC/Chol liposomes highly incorporated various fluorescent dyes compared with EggPC/Chol liposomes. Senior et al. [26] also demonstrated that DSPC/Chol liposomes showed longer half-lives than EggPC/Chol liposomes after their intravenous injection to mice. It is recognized that liposomes composed of unsaturated lipids with low phase transition temperatures display lower stability towards leakage than saturated analogues with higher phase transitions. In the experiments of the monolayer interactions of phospholipids and Chol, Demel et al. [27] reported that saturated phospholipids such as DSPC could condense with Chol partially by Van der Waals interaction. The stable retention of Pal-PLS in the DSPC/Chol liposomes after their incubation with rat plasma may be explained by the condensation effect and the high phase transition temperature of DSPC.

Surface modification of liposomes with the hydrophilic polymer PEG has provided a major advance in drug delivery applications due to the ability of this polymer to reduce protein binding and plasma elimination of liposomes [28]. The medical utility of these long circulation liposomes is displayed by the fact that several liposomal drugs are on the market or in late stage clinical trials [29,30]. In fact, PEGylation using DSPE-PEG 2000 and DSPE-PEG 5000 induced the low incorporation efficiency of EggPC/Chol liposomes after their incubation with rat plasma (Table 4). Addition of PEG-lipids on liposomes formed hydrophilic outer layer, and the hydrophilic surface protect liposomes from plasma components binding and subsequent mononuclear phagocytic system uptake [28]. It is valuable to note that not only water-soluble drug but also lipophilic drug such as Pal-PLS that may be inserted into liposomal lipid bilayer was stably retained in PEG-modified liposomes in the presence of plasma.

The activity of liposomal corticosteroids should depend on the various procedures such as drug release from liposomes in blood, uptake of liposome by target cells, and drug release from liposomes in cytoplasm. Strong retention of drugs into liposomes can not always improve its pharmacokinetics and activity.

Optimization of retention and release of drugs in liposomes is considered to be very important for therapeutic success. Our data indicated that adjustment of drug lipophilicity and lipid compositions including PEG-lipids is a promising approach to prepare adequate liposomes of PLS. On the other hand, gel filtration method could be useful for liposomal formulations to simultaneously evaluate both their dilution by blood circulation and interaction with biological components under in vivo condition.

## 5. Conclusion

We systemically examined about the drug retention from various liposomes in the ultrafiltration and gel filtration. The enhanced lipophilicity of drug was more important to improve the drug retention in liposomes than lipid compositions of liposomes. The release of lipophilic derivative from the liposomes by plasma was efficiently protected by selection of lipids and PEGylation.

## Acknowledgements

This work was supported in part by a Grant-in-Aid for Scientific Research from the Ministry of Education, Culture, Sports, Science, and Technology, Japan. The authors thank Naomi Takada for technical assistance.

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## Ocular Pharmacokinetic/ Pharmacodynamic Modeling for Bunazosin After Instillation into Rabbits

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Received September 3, 2003; accepted January 29, 2004

**Purpose.** To develop a pharmacokinetic/pharmacodynamic (PK/PD) model for an  $\alpha_1$ -blocker (bunazosin) after instillation. The PK/PD model can predict both the drug concentrations in various ocular tissues and the hypotensive effect.

**Methods.** Bunazosin concentrations were determined with High Performance Liquid Chromatography (HPLC) in tear fluid, the aqueous humor, cornea, and iris-ciliary body after instillation or ocular injection into the anterior chamber in rabbits. After instillation of bunazosin in rabbits, intraocular pressure (IOP) was also determined with a pneumatic tonometer. The PK/PD parameters were estimated by fitting the concentration-time profiles and the hypotensive effect-time profiles to the developed PK/PD models using the MULTI (RUNGE) program.

**Results.** On the basis of the concentration-time profiles of bunazosin, a PK model, including seven compartments, was developed for examining the behavior of bunazosin after instillation. Then, two PK/PD models for hypotensive effect of bunazosin were developed using an indirect response (model A) and the relationship between IOP and aqueous humor flow (model B). These models well described the concentration-time profiles and hypotensive effect-time profiles of bunazosin after instillation.

**Conclusions.** This study is the first trial to develop a PK/PD model for an antiglaucoma agent using an indirect response and the relationship between IOP and aqueous humor flow.

**KEY WORDS:** bunazosin; eye; indirect response; intraocular pressure; pharmacokinetic/pharmacodynamic model.

### INTRODUCTION

Upon instillation of an ophthalmic drug, most of the instilled drug is rapidly eliminated from the precorneal area due to drainage through the nasolacrimal duct and dilution by tear turnover (1). These physiological complexities have interfered with the development of pharmacokinetic (PK) models for ophthalmic agents. Logical use of ophthalmic drugs based on a knowledge of pharmacokinetics is important for effective medication. In a previous report, we successfully developed

an ocular PK model that accounted for the corneal diffusion process to predict the behavior of instilled  $\beta$ -blockers (2).

On the other hand, the pharmacodynamics of an instilled drug are more important than the pharmacokinetics in the design of an appropriate dosage form and regimen to achieve the desired therapeutic effect on patients. However, there have been few reports of ocular pharmacokinetic/pharmacodynamic (PK/PD) models for ophthalmic drugs after instillation. Especially, the pharmacokinetics of antiglaucoma agents in ocular tissues and their relationship to the corresponding intraocular pressure (IOP) has never been thoroughly investigated.

In the current study, we newly developed an ocular PK/PD model that enabled us to predict both the drug concentrations in ocular tissues and the IOP after instillation of bunazosin in rabbits. Bunazosin ( $\alpha_1$ -blocker) is commercially available as a new antiglaucoma agent.

### MATERIALS AND METHODS

#### Animals

Male Nippon albino rabbits (2.0–3.0 kg) were individually housed in cages in an air-conditioned room and maintained on a standard laboratory diet (ORC4, Oriental Yeast Co., Ltd., Tokyo, Japan). The rabbits had free access to water and were maintained in a 12-h light-dark cycle (lights on at 7:00 AM, and lights off at 7:00 PM). All experiments in the current study conformed to the "Principles of Laboratory Animal Care" (NIH Publication No. 85-23, revised 1985).

#### Materials

Bunazosin hydrochloride was kindly supplied from Santen Pharmaceutical Co., Ltd. (Osaka, Japan). Prazosin hydrochloride was purchased from Wako Pure Chemical Industries, Ltd. (Osaka, Japan). All other chemicals of reagent grade were obtained from Nacalai Tesque Inc. (Kyoto, Japan). The drug solution was prepared with pH 7.4 phosphate-buffered saline (PBS).

#### Drug Disposition After Instillation

Unanesthetized rabbits were placed in restraint boxes. Twenty-five  $\mu$ l of drug solution (bunazosin: 0.1%) were carefully instilled with a micropipette (Gilson Medical Electronics, Villiers-le-Bel, France) into the middle of the lower conjunctival eye sac. At the appropriate time after instillation, tear fluid (0.5  $\mu$ l) was collected in glass capillary (EM minicaps, Hirschmann Laborgerate, Eberstadt, Germany) from the middle of a lower marginal tear strip and diluted with 50  $\mu$ l of acetonitrile.

Under the same conditions, the rabbits were sacrificed by an overdose of sodium pentobarbital at the appropriate time after drug instillation. After thoroughly rinsing the corneal and conjunctival surfaces with 0.9% NaCl and blotting them dry, the aqueous humor was aspirated from the anterior chamber using a 1.0-ml disposable syringe with a 27-gauge needle. The cornea and iris-ciliary body were dissected with a surgical knife. The samples were submitted to an HPLC assay.

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### Drug Disposition After Ocular Injection

Unanesthetized rabbits were placed in restraint boxes. About 10 min before the administration of the drug, the eyes were anesthetized locally with 0.4% oxybuprocaine hydrochloride. One microliter of drug solution (bunazosin: 0.01%) was injected into the anterior chamber using a microsyringe fitted with a 30-gauge needle. After injection, the needle was carefully removed to prevent leaks of drug solution. The protocol for sampling the aqueous humor and other tissues was as described above. The samples were then submitted for an HPLC assay.

### Drug Determination

The tear fluid samples for bunazosin were centrifuged at 8000g for 10 min and the supernatants (10–30  $\mu$ l) were mixed with water (60  $\mu$ l) after appropriate dilution with acetonitrile. The samples (50  $\mu$ l) were injected into an HPLC system. The drug concentrations were determined with an external standard.

The aqueous humor samples of bunazosin were mixed with acetonitrile (1 ml) containing an internal standard (20 nM prazosin hydrochloride). The cornea and iris-ciliary body were homogenized with acetonitrile (1 ml) containing an internal standard (20 nM prazosin hydrochloride) on ice. These mixtures were centrifuged at 8000g for 5 min, and the supernatants (100  $\mu$ l) were diluted with 200  $\mu$ l of water. The samples (50  $\mu$ l) were then injected into an HPLC system.

The HPLC system (LC-10AD, Shimadzu Co., Ltd., Kyoto, Japan) was used in the reversed-phase mode for the assay. The stationary phase used was a TSKgel ODS-80T<sub>M</sub> packed column (250 mm length  $\times$  4.6 mm i.d., Tosoh Inc., Tokyo, Japan). A mixture of acetonitrile and 66 mM NaH<sub>2</sub>PO<sub>4</sub> (3:7, v/v) was used for the mobile phase with a flow rate of 0.6 ml/min (for tear samples) or 0.8 ml/min (for ocular tissue samples). Retention of the drug was monitored with a spectrofluorometric detector (RF-10A, Shimadzu Co., Ltd.; excitation wavelength 350 nm, emission wavelength 405 nm).

### Measurement of IOP

Unanesthetized rabbits trained enough to be handled were placed in restraint boxes. Twenty-five microliters of drug solution (bunazosin: 0.002%, 0.01%, and 0.1%) were carefully instilled into the rabbits' eyes. The control group received 25  $\mu$ l of PBS. One drop of 0.4% oxybuprocaine hydrochloride was instilled into both eyes prior to IOP measurements. IOP was determined using a pneumatic tonometer (Modeling 30 Classic Pneumatometer, Mentor Co., Ltd., Santa Barbara, CA, USA).

### Data Analysis

The concentration-time profiles for bunazosin in the tear fluid (Fig. 1) were analyzed by a two-compartment model. The concentration ( $C_T$ ) at time  $t$  is expressed as follows:

$$C_T = A \cdot e^{-\alpha t} + B \cdot e^{-\beta t} \quad (1)$$

Hybrid parameters,  $A$ ,  $B$ ,  $\alpha$ , and  $\beta$ , are defined as  $\alpha + \beta = K_{TR1} + K_{R1T} + K_{Te}$ ,  $\alpha \cdot \beta = K_{R1T} \cdot K_{Te}$  and  $(A \cdot \alpha + B \cdot \beta)/(A + B) = K_{R1T}$ . The parameters  $K_{TR1}$  and  $K_{R1T}$  are the

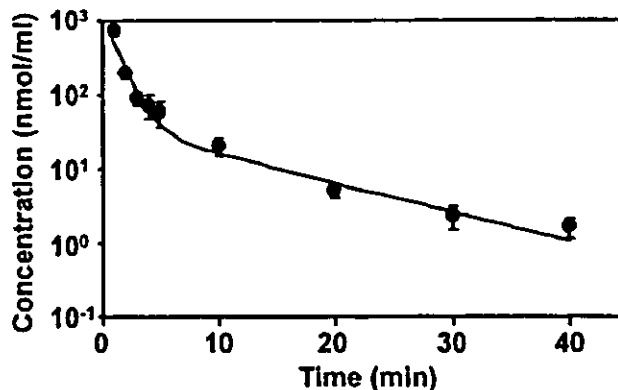


Fig. 1. Concentration of bunazosin in the tear fluid after instillation. (●) Experimental data and (—) fitting line. Each point represents the mean  $\pm$  SE of at least six experiments.

transfer rate constants between the tear fluid and reservoir-1 (Res. 1), and  $K_{Te}$  is the elimination rate constant from the tear fluid. The parameters were estimated from the tear fluid concentrations after instillation of bunazosin using MULTI, a nonlinear least-squares computer program (3).

The ocular behaviors of bunazosin after instillation and injection into the anterior chamber (Figs. 2 and 3) were analyzed by the PK model (Fig. 4) in which the cornea was considered to be divided into two subcompartments, one to represent the epithelium and the other to represent the stroma. Based on this model, the differential equations of drug amounts in the corneal epithelium ( $X_{CE}$ ), the corneal stroma ( $X_{CS}$ ), the aqueous humor ( $X_A$ ), the iris-ciliary body ( $X_I$ ), and Res. 2 ( $X_{R2}$ ) can be expressed as follows:

$$\frac{dX_{CE}}{dt} = K_{TC} \cdot X_T - K_{ES} \cdot X_{CE} \quad (2)$$

$$\frac{dX_{CS}}{dt} = K_{ES} \cdot X_{CE} + K_{AC} \cdot X_A - K_{CA} \cdot X_{CS} \quad (3)$$

$$\frac{dX_A}{dt} = K_{CA} \cdot X_{CS} + K_{R2A} \cdot X_{R2} - (K_{AC} + K_{AR2} + K_{AI} + K_{Ae1} + K_{Ae2}) \cdot X_A \quad (4)$$

$$\frac{dX_I}{dt} = K_{AI} \cdot X_A - K_{Ie} \cdot X_I \quad (5)$$

$$\frac{dX_{R2}}{dt} = K_{AR2} \cdot X_A - K_{R2A} \cdot X_{R2} \quad (6)$$

where  $X_T$  is drug amount in the tear fluid,  $K_{TC}$  is the transfer rate constant from the tear fluid to the corneal epithelium,  $K_{ES}$  is the transfer rate constant from the corneal epithelium to the corneal stroma,  $K_{AI}$  is the transfer rate constant from the aqueous humor to the iris-ciliary body,  $K_{CA}$  and  $K_{AC}$  are the transfer rate constants between the corneal stroma and the aqueous humor,  $K_{AR2}$  and  $K_{R2A}$  are the transfer rate constants between the aqueous humor and Res. 2,  $K_{Ie}$  is the elimination rate constant from the iris-ciliary body, and  $K_{Ae1}$  and  $K_{Ae2}$  are elimination rate constants from the aqueous humor by aqueous humor flow and other routes, respectively. The value of  $K_{Ae1}$  was obtained from the data reported by Sakurai *et al.* (4). The PK parameters were estimated from the drug amount-time profiles in the ocular tissues after instilla-

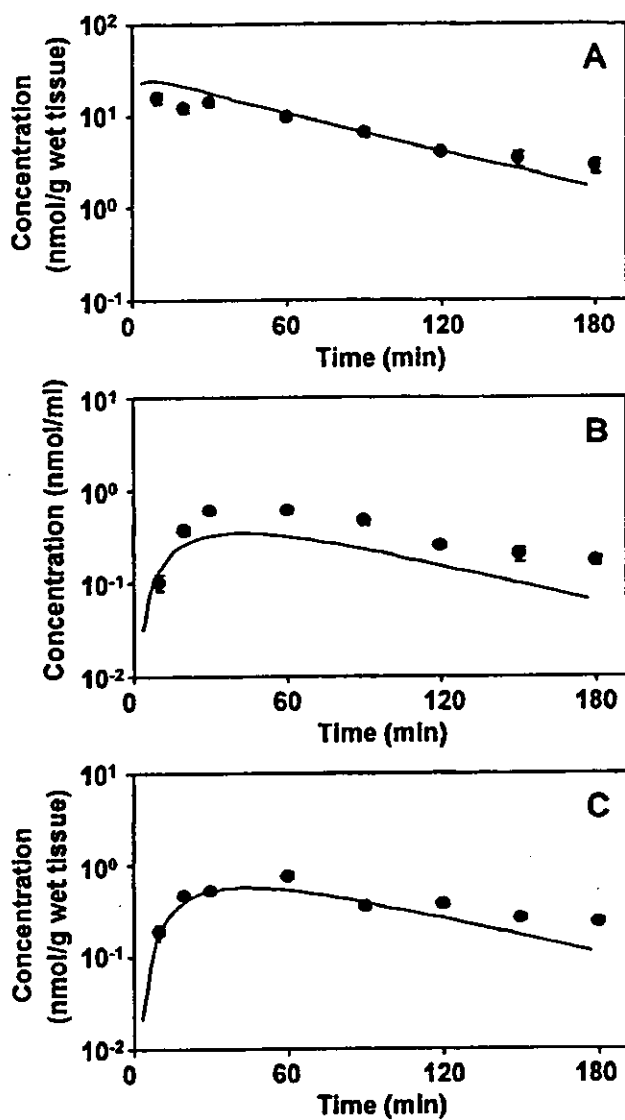


Fig. 2. Concentration of bunazosin in the (A) cornea, (B) aqueous humor, and (C) iris-ciliary body after instillation. (●) Experimental data and (—) fitting line. Each point represents the mean  $\pm$  SE of at least three experiments.

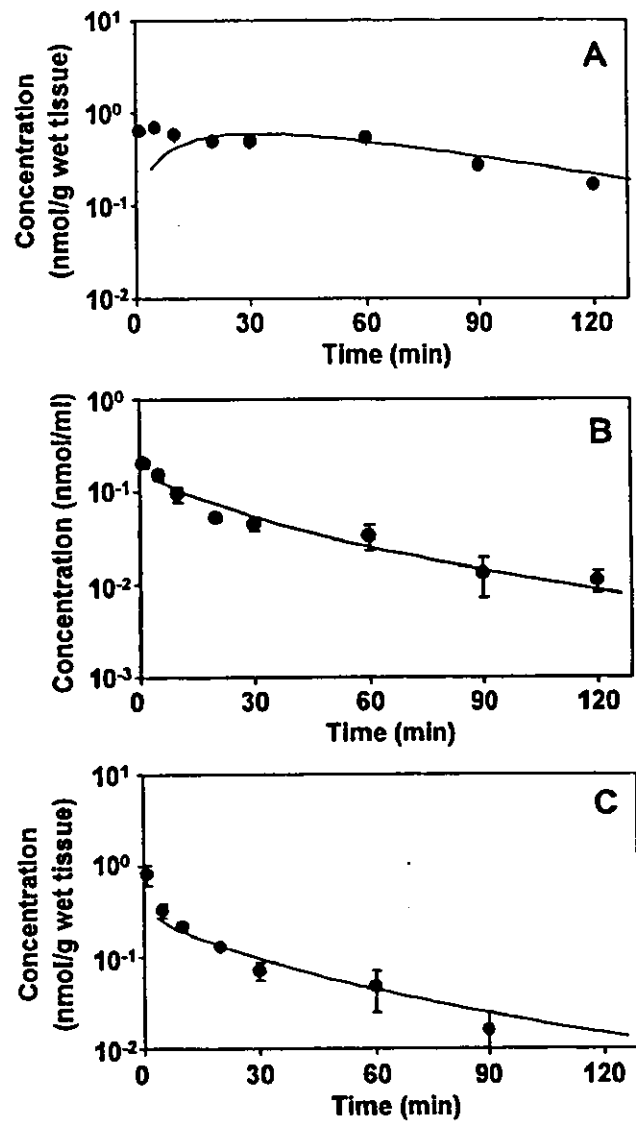


Fig. 3. Concentration of bunazosin in the (A) cornea, (B) aqueous humor, and (C) iris-ciliary body after injection into the anterior chamber. (●) Experimental data and (—) fitting line. Each point represents the mean  $\pm$  SE of at least three experiments.

tion and injection. The estimations were carried out by using MULTI (RUNGE), a nonlinear least-squares computer program based on the Runge-Kutta-Gill method (5).

Based on reports that bunazosin reduces IOP by increasing outflow of aqueous humor (6,7), the hypotensive effect-time profiles after instillation (Fig. 5) were analyzed by the PK/PD model using an indirect response (Fig. 6A) (8). The model represents a general approach where the rate of change in response is controlled by a zero-order process (zero-order rate constant  $K_{in}$ ) for production of the response and a first-order process (first-order rate constant  $K_{out}$ ) for loss of the response. In the indirect response modeling, IOP is produced at zero-order, and IOP loss is indirectly controlled by drug concentrations in the aqueous humor ( $C_A$ ) according to an  $E_{max}$  model, which can be related to receptor theory. The differential equation of IOP can be expressed as follows:

$$\frac{dIOP}{dt} = K_{inA} - K_{outA} \cdot \left( 1 + \frac{E_{maxA} \cdot C_A}{EC_{50A} + C_A} \right) \cdot IOP \quad (7)$$

where  $K_{inA}$  is the zero-order rate constant for IOP production, and  $K_{outA}$  is the first-order rate constant for IOP loss. It is assumed that  $K_{inA}$  and  $K_{outA}$  fully account for production and loss of the IOP.  $E_{maxA}$  is the maximum effect attributed to the drug, and  $EC_{50A}$  is the drug concentration producing 50% of the maximum effect.  $C_A$  is defined as  $C_A = X_A/V_A$ . The aqueous humor volume  $V_A$  is obtained from the data reported by Conrad and Robinson (9). The PD parameters were estimated from the hypotensive effect-time profiles after instillation. The estimations were carried out using MULTI (RUNGE).

On the other hand, the hypotensive effect-time profiles after instillation (Fig. 5) were also analyzed by another PK/PD model considering aqueous humor flow (Fig. 6B). This



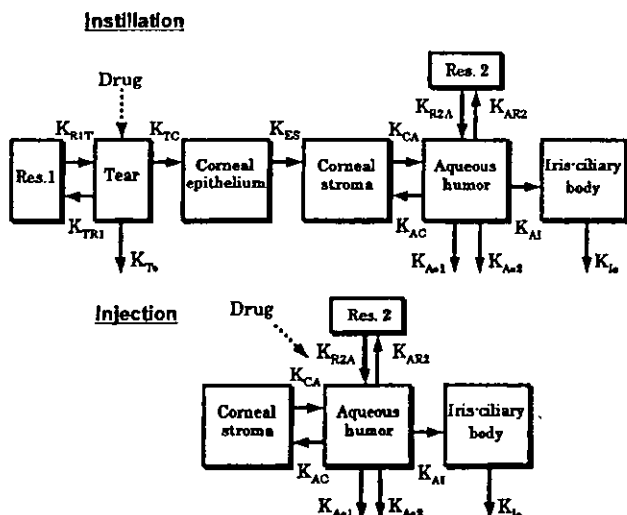


Fig. 4. Pharmacokinetic models for bunazosin in the precorneal area and other tissues after instillation, and after injection into the anterior chamber.

model includes an equilibrium of the aqueous humor flow ( $F_{in}$ ) to the aqueous humor flow ( $F_{us}$ ), with trabecular outflow ( $F_{tra}$ ).  $F_{tra}$  expressed as follows:

$$F_{tra} = C_{of} (IOP - P_v) \quad (8)$$

where  $P_v$  is the episcleral venous pressure and  $C_{of}$  is the outflow facility. Based on these relationships, the IOP was expressed by aqueous humor flow as follows (10):

$$IOP = P_v + \frac{F_{in} - F_{us}}{C_{of}} \quad (9)$$

As bunazosin reduces IOP by increasing uveoscleral outflow (7), the differential equation of IOP after instillation of bunazosin can be expressed as follows:

$$\frac{dIOP}{dt} = - \frac{dF_{us}}{dt} \cdot \frac{1}{C_{of}} \quad (10)$$

In indirect response modeling,  $F_{us}$  is indirectly controlled by drug concentrations in the aqueous humor ( $C_A$ ) according to the  $E_{max}$  model. Bunazosin is considered to be a suppressing biological factor, which reduces  $F_{us}$  via the  $\alpha_1$ -receptor (11). Therefore, the differential equation of  $F_{us}$  can be expressed as follows:

$$\frac{dF_{us}}{dt} = K_{inB} - K_{outB} \cdot \left( 1 - \frac{E_{maxB} \cdot C_A}{EC_{50B} + C_A} \right) \cdot F_{us} \quad (11)$$

where  $K_{inB}$  is the zero-order rate constant for production of  $F_{us}$ , and  $K_{outB}$  is the first-order rate constant for loss of  $F_{us}$ . It is assumed that  $K_{inB}$  and  $K_{outB}$  fully account for production and loss of the  $F_{us}$ .  $E_{maxB}$  is the maximum effect attributed to the drug, and  $EC_{50B}$  is the drug concentration producing 50% of the maximum effect.

When Eq. 9 and Eq. 11 were substituted in Eq. 10, we obtained the equation for the PK/PD model considering the relationship between IOP and aqueous humor flow as follows:

$$\frac{dIOP}{dt} = - \left\{ K_{inB} - K_{outB} \cdot \left( 1 - \frac{E_{maxB} \cdot C_A}{EC_{50B} + C_A} \right) \right.$$

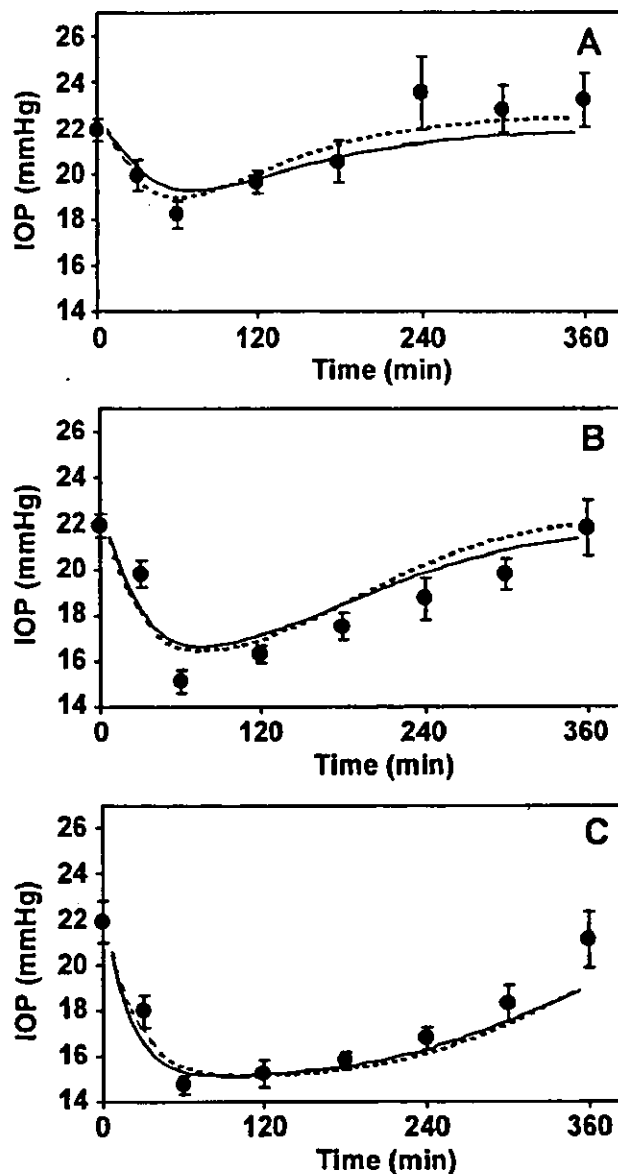


Fig. 5. Intraocular pressures after instillation of bunazosin in rabbits. (A) 0.002%, (B) 0.01%, (C) 0.1%. (●) Experimental data revised by control, (—) fitting line for the indirect response model, (---) fitting line for the aqueous humor flow model. Each point represents the mean  $\pm$  SE of at least 10 experiments.

$$\cdot [F_{in} - C_{of} \cdot (IOP - P_v)] \cdot \frac{1}{C_{of}} \quad (12)$$

Physiological parameters  $P_v$ ,  $F_{in}$ , and  $C_{of}$  were obtained from the data reported by Sakurai *et al.* (4). PD parameters  $K_{inB}$ ,  $K_{outB}$ ,  $E_{maxB}$ , and  $EC_{50B}$  were estimated from the hypotensive effect-time profiles after instillation of bunazosin. The estimations were carried out using MULTI (RUNGE).

RESULTS

The concentration-time profiles of bunazosin in the ocular tissues were determined after its instillation into the rabbits' eyes. Figure 1 shows the concentrations of bunazosin in the tear fluid. The profile showed a biexponential curve. The

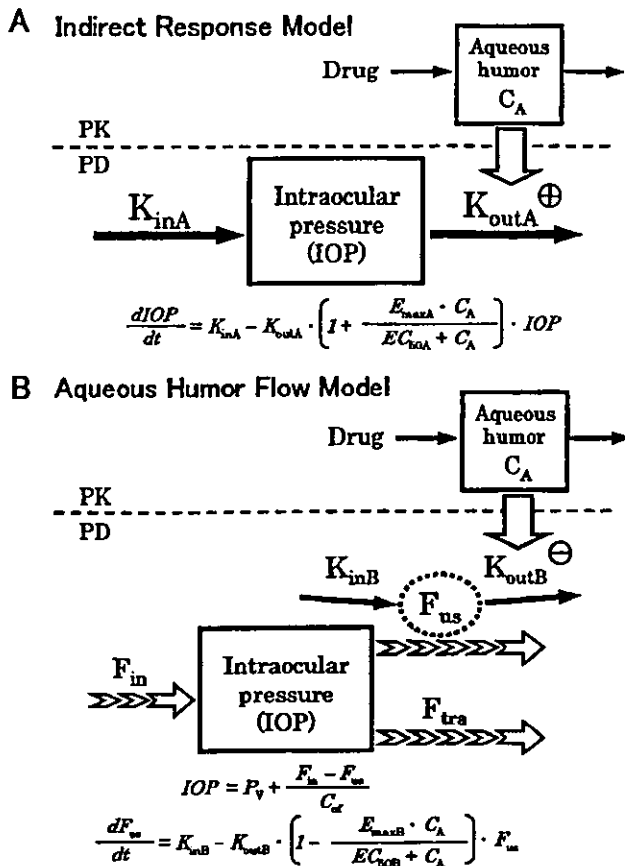


Fig. 6. Pharmacokinetic/pharmacodynamic model for bunazosin: (A) the indirect response model, (B) the aqueous humor flow model.

elimination rate constant and apparent distribution volume were estimated according to a two-compartment model. Figure 2 shows the concentrations of bunazosin in the cornea, aqueous humor, and iris-ciliary body after instillation. The bunazosin concentrations in the cornea were higher than those in the aqueous humor and the iris-ciliary body. The aqueous humor and the iris-ciliary body concentrations of bunazosin reached a maximum at 30 or 60 min after dosing and then gradually disappeared.

The concentration-time profiles of bunazosin in the cornea, aqueous humor, and iris-ciliary body were also determined after injection into the anterior chamber of the rabbits and are shown in Fig. 3. The bunazosin concentrations in the cornea were not much higher than those in the aqueous humor and the iris-ciliary body. The profiles in the aqueous humor and the iris-ciliary body showed a biexponential curve.

In order to develop a PK/PD model for bunazosin, the IOP were determined with a pneumatic tonometer after instillation into the rabbits. Figure 5 shows the hypotensive effect-time profiles of bunazosin after instillation at 0.002%, 0.01%, and 0.1%. The intraocular pressure reductions reached a maximum at 60 min after instillation. Thereafter, the response gradually returned to the baseline with a later return found with the larger dose.

## DISCUSSION

Glaucoma is characterized by high IOP, hardening of the eyeball, and partial or complete loss of vision (1). The IOP is

maintained by a complex and dynamic equilibrium of aqueous humor production and escape. Bunazosin ( $\alpha_1$ -blocker) is commercially available as a new antiglaucoma agent. After topical application, bunazosin has been demonstrated to significantly lower IOP in rabbits (7,11), cats (11), and humans (6). We newly developed a PK/PD model for bunazosin, as there have been no such reports, after its instillation into rabbits.

The physiological complexity of the eye leads to a complicated mathematical model with many compartments and pathways of transportation. Various compartment models have already been reported for antiglaucoma agents such as pilocarpine and timolol (1,12,13). Kato and Iwata (14) successfully described ocular absorption of bunazosin with a simple compartment model after its topical instillation into rabbits. In order to develop an ocular PK model for bunazosin, two experiments using both instillation and injection into the anterior chamber were designed to increase the reliability of the pharmacokinetic parameters.

At first, the concentration-time profiles of bunazosin in the ocular tissues were determined after its instillation (Figs. 1 and 2). Based on the results of the biexponential profiles in tear fluid, a two-compartment model was used to describe the disposition of bunazosin in the precorneal area. Second, the concentration-time profiles of bunazosin in the ocular tissues were determined after its injection into the anterior chamber (Fig. 3). Based on the results of the biexponential profiles in the aqueous humor and the iris-ciliary body, it is assumed that bunazosin distributed to intraocular tissues such as the lens and sclera.

Based on these results, a PK model for bunazosin was established as described in Fig. 4. The PK parameters were estimated according to the model given in Table I. Although the PK model was complicated and had many parameters, the fitting curves using the PK model and the PK parameters almost agreed with the measurements (Figs. 1, 2, and 3).

In order to develop a PK/PD model for bunazosin, IOP was determined with a pneumatic tonometer after its instillation into rabbits (Fig. 5). It was already reported that topical administration of bunazosin significantly reduced IOP in rabbit eyes (7,11). Both single and multiple doses (0.1%  $\times$  2 drops) caused a statistically significant decrease in IOP of at least 29% (7), which is comparable with the current study.

Table I. Ocular Pharmacokinetic Parameters for Bunazosin

Parameter	Value	Standard deviation
$K_{TR1}$ ( $\text{min}^{-1}$ )	0.173	0.054
$K_{R1T}$ ( $\text{min}^{-1}$ )	0.118	0.030
$K_{Tc}$ ( $\text{min}^{-1}$ )	0.661	0.044
$K_{TC}$ ( $\text{min}^{-1}$ )	0.018	0.002
$K_{ES}$ ( $\text{min}^{-1}$ )	0.212	0.146
$K_{CA}$ ( $\text{min}^{-1}$ )	0.025	0.004
$K_{AC}$ ( $\text{min}^{-1}$ )	0.059	0.015
$K_{AT}$ ( $\text{min}^{-1}$ )	0.141	0.017
$K_{AR2}$ ( $\text{min}^{-1}$ )	1.267	0.562
$K_{R2A}$ ( $\text{min}^{-1}$ )	0.359	0.167
$K_{Ac1}$ ( $\text{min}^{-1}$ )	0.010*	
$K_{Ac2}$ ( $\text{min}^{-1}$ )	$2 \times 10^{-6}$	$1 \times 10^{-6}$
$K_{Ic}$ ( $\text{min}^{-1}$ )	0.495	0.060

\* Value reported by Sakurai *et al.* (4)

Cats and humans were also reported to show a similar IOP reduction (6,11).

In the current study, the relationship between aqueous humor concentrations of bunazosin and IOP changes showed anticlockwise hysteresis, indicating a delayed response. The time for maximal response shifted slightly with dose. Temporal dissociation between the time courses of concentration and effect might be caused by an indirect response mechanism resulting in anticlockwise hysteresis for the concentration-effect (IOP change) relationship (15). Therefore, these results suggest that bunazosin showed a hypotensive effect via an indirect response. Actually, in the preliminary analysis, the hypotensive effect-time profiles of bunazosin showed poor fitting to the common PK/PD (direct response) model (data are not shown). Therefore, we tried to develop the new models.

### Indirect Response Model

Many drugs act by inhibiting or stimulating the release of an endogenous substance in nature. These responses via receptors may be considered indirect. Dayneka *et al.* indicated that the actions of adrenergic agonists/antagonists and cholinergic agonists/antagonists could be characterized by the four basic indirect response modelings (8).

The ocular hypotensive mechanism of bunazosin has been investigated in several studies. Bunazosin reportedly is selective for the receptor regulating IOP. In rabbits, it did not affect aqueous flow, but appeared to increase uveoscleral outflow (7). Therefore, in indirect response modeling, the drug concentration in the aqueous humor ( $C_A$ ) was linked to the PD parameter for IOP loss ( $K_{outA}$ ). The PD parameters were estimated according to model A (Fig. 6A) as given in Table II. The fitting curves using the PK/PD model and PD parameters were consistent with the measurements (Fig. 5).

### Aqueous Humor Flow Model

IOP is required for an optically efficient globe and is generated by the secretion of aqueous humor. Ciliary processes as a result of active transport of solutes form the aqueous humor over the double-layered ciliary epithelium and,

secondary to this, diffusion of water. After entering the anterior chamber via the pupil, the aqueous humor is drained by two different pathways at the iridocorneal chamber angle. Some of the aqueous humor enters Schlemm's canal by way of the trabecular meshwork and then passes via the collector channels into the episcleral veins. As there is no epithelial barrier between the anterior chamber and the ciliary muscle, aqueous humor can pass between the muscle bundles into the supraciliary and suprachoroidal spaces, from which it is drained through the sclera. These outflow routes are called the uveoscleral outflow routes (10).

On the other hand, it was reported that IOP is described by an equation including episcleral venous pressure ( $P_v$ ), aqueous humor flow ( $F_{in}$ ), uveoscleral outflow ( $F_{us}$ ), and outflow facility ( $C_{of}$ ) (10). Antiglaucoma agents change the IOP by influencing the inflow and outflow rates of aqueous humor. Zhan *et al.* reported that bunazosin is an effective ocular hypotensive drug capable of lowering IOP by increasing uveoscleral outflow alone (7). Effects of bunazosin on outflow facility in rabbits have not been reported. Therefore, this PK/PD model considering the relationship between IOP and aqueous humor flow was newly developed to assess the hypotensive effect of bunazosin using physiological parameters (model B, Fig. 6B). In this model, bunazosin indirectly induces uveoscleral outflow by suppressing the receptor (11). The PD parameters were estimated according to model B (Fig. 6B) as given in Table II. The fitting curves using this PK/PD model and PD parameters were consistent with the measurements (Fig. 5). The PK/PD model considering the relationship between IOP and aqueous humor flow (model B, Fig. 6B) showed 73% and 0.002 nmol/ml of  $E_{maxB}$  and  $EC_{50B}$ , respectively. Kimoto *et al.* reported that the receptor dissociation constant ( $K_d$ ) of bunazosin was 0.0041 nmol/ml with smooth muscles in the rabbit proximal urethra (16). This  $K_d$  value almost agreed with the present  $EC_{50B}$  value. Using this  $K_d$  for bunazosin on the  $\alpha_1$ -receptor, the receptor occupancy values of bunazosin 1 h after instillation of 0.002%, 0.01%, and 0.1% were 60.9%, 88.6%, and 98.7%, respectively (17). A receptor occupancy model can be introduced from the PK/PD model using  $K_d$  instead of  $EC_{50B}$ . This PK/PD model will be effective in estimating an appropriate regimen for ophthalmic pharmacotherapy and development of ocular drug delivery systems.

The equation for the indirect response model is almost compatible with that of the aqueous humor flow model. Therefore, Akaike's information criteria (AIC) value of the indirect response model (AIC = -5.7) was similar to that of the aqueous humor flow model (AIC = -4.0). Although the indirect response model is a simplified model, the aqueous humor flow model is complicated. The aqueous humor flow model, however, can analyze the physiological behavior of aqueous humor flow after drug administration.

In conclusion, the PK/PD models well described the concentration of bunazosin in the ocular tissues and its hypotensive effect after instillation into rabbits. This study is the first to develop a PK/PD model for an antiglaucoma agent using an indirect response model and the idea of aqueous humor flow.

### ACKNOWLEDGMENTS

The author wishes to thank Dr. Yamaoka for supplying MULTI (RUNGE).

Table II. Ocular Pharmacodynamic Parameters for Bunazosin

Parameter	Value	Standard deviation
PD parameters		
Indirect response model		
$K_{inA}$ (mmHg · min <sup>-1</sup> )	0.845*	
$K_{outA}$ (min <sup>-1</sup> )	0.039	0.037
$E_{maxA}$	0.478	0.177
$EC_{50A}$ (nmol/ml)	0.014	0.014
Aqueous humor flow model		
$K_{inB}$ (μl/min · min <sup>-1</sup> )	0.077	0.068
$K_{outB}$ (min <sup>-1</sup> )	0.155	0.175
$E_{maxB}$	0.727	0.170
$EC_{50B}$ (nmol/ml)	0.002	0.003
Physiological parameters†		
$F_{in}$ (μl/min)	2.80	
$C_{of}$ (μl · min <sup>-1</sup> · mmHg <sup>-1</sup> )	0.170	
$P_v$ (mmHg)	9.00	

\*  $K_{inA}$  was estimated by  $K_{outA}$ .

† Values reported by Sakurai *et al.* (4).

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